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Date: 2024-07-31 11:27 UTC

* All sources 96 | Internet sources 30 | Plagiarism Prevention Pool 66

- [0] [www.frontiersin.org/journals/microbiology/articles/10.3389/fmicb.2023.1232086/full#:~:text=1-Aminocyclopropane-1-carboxylic acid \(AC](https://www.frontiersin.org/journals/microbiology/articles/10.3389/fmicb.2023.1232086/full#:~:text=1-Aminocyclopropane-1-carboxylic acid (AC)
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
















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





















































- [21] www.ncbi.nlm.nih.gov/pmc/articles/PMC3509510/
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- [31]  from a PlagScan document dated 2019-10-06 11:46
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- [33]  from a PlagScan document dated 2021-01-08 16:53
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- [52]  from a PlagScan document dated 2020-07-19 09:33
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- ✓ [53]  from a PlagScan document dated 2019-12-19 08:18
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- ✓ [54]  from a PlagScan document dated 2022-02-03 10:28
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- ✓ [55]  from a PlagScan document dated 2021-04-14 04:33
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- ✓ [66]  www.nature.com/articles/s41467-023-39881-9
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- ✓ [67]  from a PlagScan document dated 2020-11-16 08:37
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- ✓ [71]  from a PlagScan document dated 2021-12-09 13:03
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- ✓ [72]  from a PlagScan document dated 2021-07-24 14:29
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- ✓ [74]  from a PlagScan document dated 2020-12-22 11:47
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- ✓ [75]  from a PlagScan document dated 2018-05-28 18:27
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- ✓ [76]  from a PlagScan document dated 2017-06-18 10:42
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1Original Article

2In Silico Analysis of LPMO Inhibition by Ethylene Precursor ACCA
3to Combat Potato Late Blight

4Abstract: Potato late blight (PLB), caused by the pathogen *Phytophthora*
5*infestans*, severely threatens potato production worldwide. This study
6investigates the potential of the ethylene precursor 1-amino-
7cyclopropane-1-carboxylic acid (ACCA) to inhibit Lytic Polysaccharide
8Monooxygenases (LPMOs) in *P. infestans*, a key protein involved in the
9disease's pathogenesis. Our findings demonstrate that ACCA significantly
10enhances the immune response in potato plants against *P. infestans*, with
11a binding energy of -8.85 kcal/mol. Integrating ACCA treatment into
12existing PLB management strategies could offer a novel and sustainable
13approach to combat this devastating disease. This research provides
14valuable insights into reducing the global impact of PLB and improving
15food security through innovative control measures.

16

17Keywords: 1-amino-cyclopropane-1-carboxylic acid, molecular docking,
18molecular dynamic simulation, sustainable potato production

19

201. Introduction

21*Phytophthora infestans*, commonly known as late blight or potato blight,
22is a highly destructive disease affecting potato plants. Thriving in humid
23environments with temperatures ranging from 4 to 29°C, the pathogen
24can lead to extensive rotting of the plant leaves and tubers within two
25weeks under optimal conditions (Cooke et al. 2011). Notably, this disease
26was responsible for the catastrophic Irish Potato Famine in the mid-
27nineteenth century and continues to pose a significant threat to global
28crop production (Montarry et al., 2010). Despite various management
29strategies, PLB remains challenging due to the pathogen's adaptability.

30 This study explores the novel use of ACCA to inhibit a key pathogen
31 enzyme. Educational institutions and governmental organizations
32 worldwide have established various forecasting programs to manage the
33 disease (Montarry et al., 2010; Fry et al., 2015). Late blight is
34 characterized by circular or irregularly shaped lesions on leaves,
35 petioles, and stems, ranging from dark green to purplish black (Fry et al.,
36 2013). Spore-producing structures may develop on the under-leaf
37 surfaces beneath the lesions' margins. The rot can infiltrate potato tubers
38 as deep as 15 cm, facilitating further infection by secondary fungi and
39 bacteria such as those in the genus *Erwinia*, resulting in considerable
40 losses during storage, transit, and sale (Cooke et al. 2011; Arora et al.
41 2014).

42 The development of *P. infestans* in potato plants is a multifaceted and
43 intricately controlled process encompassing numerous phases of
44 invasion, colonization, and propagation. As an oomycete pathogen, *P.*
45 *infestans* begins its infection cycle by generating sporangia, which
46 release mobile zoospores under optimal environmental conditions such
47 as high humidity and mild temperatures. Upon contact with a vulnerable
48 host plant, these zoospores form cysts and subsequently develop germ
49 tubes that infiltrate the plant's epidermal cells, either directly or via
50 natural openings like stomata. Inside the host plant, *P. infestans* expands
51 and multiplies through the formation of specialized structures known as
52 haustoria, which aid in the extraction of nutrients from plant cells (Fry et
53 al. 2013; Whisson et al. 2016; Naumann et al. 2020). To suppress host
54 defenses, manipulate plant signaling pathways, and facilitate its
55 colonization and proliferation, the pathogen utilizes a range of secreted
56 effector proteins. Host resistance (R) proteins can recognize these
57 effectors, inducing a localized hypersensitive response (HR) that triggers
58 programmed cell death and effectively limits the growth and spread of
59 the pathogen. Nevertheless, *P. infestans* has evolved strategies to bypass
60 host resistance by diversifying its effector repertoire, allowing it to avoid
61 detection and sustain its virulence. As the infection advances, *P.*

62infestans generates sporangia on the plant surface, which are then
63dispersed by wind or rain, leading to new infection cycles in nearby
64plants. This destructive pathogenesis leads to the rapid development of
65water-soaked lesions and necrosis on leaves and stems, thereby causing
66significant crop losses and posing a threat to worldwide food security,
67emphasizing the importance of devising effective and sustainable
68management approaches to combat *P. infestans* in potato production
69(Nowicki et al., 2012; Fry et al., 2015; Whisson et al., 2016).

70Phytophthora can survive in various environments, including improperly
71stored tubers, rubbish piles, field plants, and greenhouses. The pathogen
72produces both sexual oospores and asexual sporangia, which can be
73windborne, infecting nearby plants within hours (Fry et al., 2013; Arora
74et al., 2014). Zoospores, a type of asexual spore with flagella, germinate
75when temperatures drop below 15°C, encysting and forming germ tubes
76as temperatures rise. Foliage blighting occurs within four to six days
77post-infection, with a new crop of sporangia forming as long as the
78weather remains cool and moist (Cooke et al., 2011; Arora et al., 2014).

79This study centers on the copper-dependent LPMOs and their
80involvement in plant-pathogen interactions. We selected this protein for
81putative inhibition in *P. infestans* by a range of relevant effectors from
82plant hosts to predict mechanistic pathogen-host interplay and identify
83potential biocontrol strategies for this devastating pathogen. The LPMOs
84are crucial to *P. infestans* plant infection, warranting further
85investigation (Vandhana et al. 2022). We conducted a virtual screening of
86several plant metabolites and growth regulators known for their
87housekeeping roles in plant physiology, particularly under biotic and
88abiotic stress. The initial screening of potent ligands was followed by a
89molecular re-docking approach, revealing 1-amino-cyclopropane-1-
90carboxylic acid (ACCA) as the most potent compound against *P. infestans*
91with a probable mechanism involving LPMO-binding target (Sabbadin et
92al. 2021; Kaur et al. 2022).^[21] ACCA is involved in the induction of the
93virulence system against *P. infestans* attack.^[1] Given the multitude of

94microbes that cause damage to *Solanum* spp., this study aims to identify
95strategies to enhance the virulence via the exogenous application of
96ACCA.

972. Materials and Methods

982.1 Target Protein & Ligand Preparation

99LPMO were retrieved from RCSB PDB
100(<https://www.rcsb.org/structure/6Z5Y>) and then minimized by Chimera
101(Figure 1A). The Ramachandran plot is shown in Figure 1B. To determine
102the function of LPMOs (PDB ID: 6Z5Y) when it binds with 1-amino-
103cyclopropane-1-carboxylic acid (ACCA) (Chem ID: 535), the 3D structures
104of both in .sdf format were retrieved from PubChem (NCBI 2023).^[0] The
105Chimera UCSF team employed a 900-step conjugate gradient energy
106minimization method followed by a 1000-step steepest-descent approach
107for further optimization. Subsequently, the ligands were converted
108to .pdb format using Open Babel (version 3.1.1) and minimized again for
1091000 iterations using the steepest descent algorithm.^[0] The AMBER ffSB14
110force field was then utilized after assigning Gasteiger charges to
111establish the partial charges. Finally, the ligand underwent geometric
112optimization with DMol3 (Figure 2). It is well-documented that
113crystallographic structures often depict LPMOs as dimers, a phenomenon
114recognized as a crystallographic artifact. This artefactual dimerization
115has been observed across various LPMO families and does not
116necessarily represent their physiological monomeric state. Our study
117utilized the dimeric form observed in the crystallographic structure (PDB
1186Z5Y) to explore potential intermolecular interactions. However, the
119focus remained on understanding the functional aspects of the LPMO
120monomers, which are essential for the enzyme's activity on polymeric
121substrates.^[0] The copper site of the monomeric LPMO is crucial for its
122interaction with glycosidic bonds, facilitating the cleavage process
123(Sabbadin et al., 2021; Askarian et al., 2021).

124

1252.2 Virtual Screening

126BIOVIA Discovery Studio Visualizer version 2022 (Leonardo et al. 2015)
127connected the chemical targets to the protein's active site. This method
128allowed for producing a final product with strong binding affinity.
129^[0]AutoDock Vina was utilized to determine the binding site of the protein
130complex and create the receptor grid.^[0] The ligand conformation with the
131highest binding energy was selected for re-docking and further analysis.

1322.3 Molecular Re-docking Studies

133Following the virtual screening, 1-amino-cyclopropane-1-carboxylic acid
134(ACCA), the most potent ligand, was used to construct the receptor grid
135using AutoDock MGL version 1.5.6. Both ligand and receptor were saved
136in .pdbqt format. The grid point spacing was set to 0.57 Å with an
137exhaustiveness value of 8.^[0] Output files were examined using PyMol
138alongside Discovery Studio Visualizer 2021 in .pdbqt format. Validation
139and enhancement of ligand binding were achieved by examining co-
140crystallized ligands.^[0] The target protein molecules facilitated the binding
141of 1-amino-cyclopropane-1-carboxylic acid (ACCA). The inhibitory
142concentration of every candidate molecule was assessed by leveraging
143virtual screening outcomes to determine the candidate demonstrating
144the most robust interaction with copper-dependent lytic polysaccharide
145monooxygenases (LPMOs).^[0] The PDB 6Z5Y structure was simplified using
146the steepest descent method (1000 steps) before applying the AMBER ff4
147force field. Additionally, prior to initiating the experiment, the
148protonation states of the copper-dependent lytic polysaccharide
149monooxygenases (LPMOs) were checked for neutralization.^[0] Polar
150hydrogen bonds, Kollman and Gastieger charges, and electrostatic forces
151produced the receptor and ligands.^[0] After merging nonpolar hydrogens,
152receptor, and ligand molecules were saved in .pdbqt format. A grid box
153with dimensions X=32, Y=31, and Z=36 with 2.40Å spacing was
154generated.^[1] The Lamarckian Genetic Algorithm was used to dock protein-
155ligand complexes, identifying those with the lowest binding free energy

156(ΔG). AutoDock 4.2.6 was used for molecular docking experiments
157(Toukmaji et al., 1996).

1582.4 Molecular Dynamics Simulation

159To emulate molecular dynamics, Desmond software (Schrödinger LLC)
160was used on a 100 ns time scale (Bowers et al., 2006). To begin
161molecular dynamics simulations, initial docking research was conducted
162to predict ligand binding states in a static environment accurately. MD
163simulations then used the classical equation of motion to monitor atom
164movements over time (Jorgensen et al., 1983; Jorgensen et al., 1996;
165Debnath et al., 2023). Simulations were conducted to examine the
166physiological state of ligand binding using the System Builder tool. The
167systems utilized an orthorhombic box model of the solvent (TIP3P) and
168applied a force field derived from OPLS 2005 (Shaw et al. 2010;
169Shivakumar et al., 2010). NaCl at 0.15 M was used to simulate
170physiological conditions, and the simulation was conducted at 300 K and
171atmospheric pressure. Models were relaxed before starting, and stability
172was assessed by tracking protein and ligand RMSD with trajectories
173recorded every 100 ps (Perveen et al., 2023; Shaw et al., 2010).

1743. Results

1753.1^[0] Virtual Screening of Ligands

176The ligand with the lowest binding energy score of -8.85 kcal/mol,
177indicating the highest binding affinity for LPMO, was 1-amino-
178cyclopropane-1-carboxylic acid (ACCA). The low binding energy of -8.85
179kcal/mol suggests a strong interaction between ACCA and LPMO,
180indicating potential efficacy in inhibiting the enzyme.^[30] This ligand
181underwent further refinement within the 6Z5Y binding cavity and was
182identified as the most potent among seven ligands tested for the receptor
183transcription protein LPMO (Table 1).

1843.2^[20] Molecular Re-docking

185Molecular docking identified the most effective intermolecular
186configuration between LPMO and seven ligands, revealing binding
187affinities listed in Table 1.^[19] During re-docking tests, 1-amino-
188cyclopropane-1-carboxylic acid (ACCA) exhibited a distinct binding
189pocket with LPMO, binding tightly to the core with a free energy of -8.85
190kcal/mol (Figure 3). During the initial docking studies, we identified
191interactions between ACCA and residues away from the substrate
192binding and catalytic sites in the dimeric context. Recognizing that these
193interactions might not fully represent the inhibitory potential, we
194conducted re-docking and molecular dynamics simulations focusing on
195the monomeric form of LPMOs. These simulations were crucial to
196observing ACCA's migration and binding stability closer to the
197monomeric catalytic site.

1983.3 Molecular Dynamics Simulation (MDS) & MMGBSA Analysis

199MD simulations were conducted on the most potent ligand ACCA and the
200pathogenic protein lytic polysaccharide monooxygenases to assess
201complex stability and quality until convergence. The results from our 100
202ns molecular dynamics simulations provided significant insights. The
203stability of the ACCA-LPMO complex over this period suggested that
204ACCA can indeed migrate closer to the catalytic copper site, indicating
205its potential as an effective inhibitor. The observed stability and
206interactions in the monomeric context support the hypothesis that ACCA
207can impact the enzyme's activity despite initial distal interactions
208observed in the dimeric form.^[10] The root mean square deviation (RMSD) of
209the C α -backbone of LPMO bound to ACCA showed a deviation of 1.5 Å,
210indicating that the protein-ligand complex remained stable throughout
211the simulation (Figure 4A, Table 2). This stability suggests that ACCA
212effectively binds to the LPMO without causing significant structural
213perturbations.^[10] Following 100 ns, notable variations were observed in the
214protein compared to the reference structure, especially within residues
21517-24 of the LPMO bound to ACCA, as illustrated in Figure 4B and Table
2162.^[8] These variations indicate localized flexibility, which may be crucial for

217the protein's function.^[0] The Radius of Gyration (Rg) plot of the C-alpha
218backbone depicted in Figure 4C demonstrated that the LPMOs bound to
219ACCA exhibited a deviation of 0.24 Å throughout the 100 ns simulation,
220according to Table 2.^[13] This minor deviation in the Rg plot confirms the
221compactness and stability of the protein-ligand complex.^[10] Additionally,
222LPMO bound to ACCA maintained stability throughout the simulation,
223evidenced by the presence of three consistent hydrogen bonds, which
224play a vital role in the stability and specificity of the protein-ligand
225interaction (Figure 5).

226ACCA exhibited pronounced hydrogen bonding interactions with the
227anticipated binding residues, forming a robust network of intermolecular
228connections.^[0] Additionally, the interaction between ACCA and the target
229protein involved diverse non-bonded interactions, as visually represented
230in Figure 5.^[0] Moreover, Figure 6 provides a visual depiction of the tightly
231bound conformation of ACCA with the LPMO protein throughout the
232entirety of the 100 ns simulation, indicating the sustained stability of the
233complex over time.

234^[0] The MMGBSA method, widely employed for assessing complex binding
235energy, was utilized to calculate the binding energy of each protein-
236ACCA complex in this study, considering various non-bonded
237interactions.^[0] The binding energies and their components for the
238interaction between LPMO and ACCA, calculated using the MMGBSA
239method, provide detailed insights into the strength and nature of the
240binding interactions. The overall binding free energy (ΔG_{bind}) of $-31.67 \pm$
2414.60 kcal/mol indicates that the binding process between ACCA and
242LPMO is energetically favorable. This substantial negative value suggests
243a strong interaction between ACCA and LPMO, which is crucial for the
244efficacy of ACCA as an inhibitor. The lipophilic contribution ($\Delta G_{\text{bindLipo}}$)
245to the binding energy is -09.44 ± 0.72 kcal/mol, highlighting the
246importance of hydrophobic interactions in stabilizing the protein-ligand
247complex. These interactions are significant in ensuring that the ligand
248fits snugly within the protein's binding pocket. Additionally, the Van der

249 Waals contribution ($\Delta G_{\text{bind}}^{\text{vdW}}$) of -21.11 ± 3.21 kcal/mol emphasizes the
250 role of these weak, non-covalent forces in maintaining the proper
251 alignment and binding of ACCA to LPMO. Electrostatic interactions also
252 play a critical role, as evidenced by the Coulombic contribution
253 ($\Delta G_{\text{bind}}^{\text{Coulomb}}$) of -15.66 ± 5.01 kcal/mol. This term accounts for the
254 attractive forces between charged groups in ACCA and LPMO, which are
255 essential for the specificity and strength of the binding. The hydrogen
256 bonding contribution ($\Delta G_{\text{bind}}^{\text{Hbond}}$) of $-4.54 \pm 1.11^{[0]}$ kcal/mol further
257 supports the stability of the complex, as hydrogen bonds are key to the
258 interaction between the ligand and the protein's active site. The solvation
259 energy ($\Delta G_{\text{bind}}^{\text{SolvGB}}$) of -30.14 ± 2.97 kcal/mol indicates favorable
260 desolvation effects upon binding.^[4] This term reflects the free energy
261 change associated with removing solvent molecules from the binding
262 interface, which contributes significantly to the binding free energy.
263 Lastly, the covalent contribution ($\Delta G_{\text{bind}}^{\text{Covalent}}$) of $-5.14 \pm 0.24^{[0]}$
264 kcal/mol, though typically less emphasized in non-covalent docking
265 studies, adds to the overall binding stability by considering covalent
266 interactions within the ligand or between the ligand and the protein. As a
267 result, the complexes' binding energy and overall stability were elevated,
268 as outlined in Table 3.

2694. Discussion

270 Potato late blight (PLB) represents a significant threat to global potato
271 production, resulting in up to \$10 billion in losses and management costs
272 annually (Montarry et al., 2010; Arora et al., 2014; Fry et al., 2013). This
273 issue remains prominent despite being 170 years since the devastating
274 Irish Potato Famine. Substantial progress has been made by growers,
275 agronomists, and laboratory scientists in comprehending the molecular
276 pathogenesis of this crucial pathosystem and in developing effective
277 management strategies to mitigate PLB (Cooke et al. 2011; Fry et al.
278 2013). *P. infestans*, a hemi-biotrophic oomycete, primarily infects potato
279 plants, targeting stems, leaves, tubers, and fruits (Cooke et al. 2011). The
280 pathogen secretes various signaling molecules and effectors that

281facilitate the initial stages of host infection. Key examples include
282cellulases, lipases, pectinases, and proteases, as well as secondary
283metabolites such as elicitors, glycoalkaloids, pyranones, and phytotoxins
284(Wang et al. 2019; Boevink et al. 2020).

285Molecular dynamics (MD) simulations provided significant insights into
286the stability and interactions of the ACCA-LPMO complex. The significant
287contributions from lipophilic and Van der Waals interactions highlight
288the importance of hydrophobic and weak non-covalent forces in the
289stability of the ACCA-LPMO complex. These interactions ensure ACCA is
290properly aligned within the binding pocket, maximizing its inhibitory
291potential. The strong electrostatic interactions, as indicated by the
292Coulombic contribution, further enhance the binding specificity and
293strength, which are critical for the effective inhibition of LPMO.
294Hydrogen bonds help maintain the structural integrity of the protein-
295ligand complex, ensuring that ACCA remains firmly bound to LPMO.^[0] The
296favorable solvation energy contribution suggests that the desolvation
297process, which occurs when ACCA binds to LPMO, is energetically
298beneficial, further supporting the stability of the complex.^[0] The presence
299of non-bonded interactions, including hydrophobic contacts, ionic
300interactions, and water bridges, further enhances the stability of the
301complex (Bowers et al. 2006; Jorgensen et al. 1996). These interactions
302indicate that ACCA can effectively bind and stabilize LPMOs, suggesting
303its potential as a promising inhibitor for managing potato late blight.

304LPMOs are pectin-degrading copper-dependent enzymes that play a
305pivotal role in *P. infestans*' ability to breach the plant cell wall,
306facilitating infection (Sabbadin et al. 2021; Jagadeeswaran et al. 2021).
307The enzymatic activity of LPMOs, driven by the reduction of the active-
308site Cu ion and subsequent re-oxidation by molecular oxygen or H₂O₂, is
309crucial for this process. Identifying ACCA as a potential inhibitor
310highlights its role in binding and stabilizing LPMOs, thereby inhibiting
311their activity (Quinlan et al. 2011; Bissaro et al. 2017, Shahid et al.
3122019). The overexpression of LPMO-encoding genes during oomycete

313infection further underscores the significance of targeting these enzymes
314to disrupt the pathogen's life cycle (Sabbadin et al. 2021). Ethylene, a
315plant hormone known for enhancing stress resilience, is synthesized from
316ACCA, which serves as its precursor. This study demonstrates that ACCA
317can inhibit LPMO activity, thus potentially blocking the pathogenesis of
318potato late blight.^[15] However, to fully validate these findings, further
319experimental research is required to confirm ACCA's inhibitory effects on
320LPMO both in vitro and in planta.^[0] Recent studies have highlighted
321ACCA's efficacy in conferring resistance to various abiotic and biotic
322stresses, reinforcing its potential as a versatile and effective treatment in
323agricultural practices (Debnath et al., 2023; Debnath et al., 2024).
324ACCA's role in enhancing potato plant defense could be a cornerstone in
325developing sustainable agricultural practices, reducing reliance on
326chemical fungicides (Tokin et al., 2021). By integrating ACCA into crop
327management strategies, farmers can achieve more resilient crops,
328contributing to sustainable agriculture and food security.

329This research underscores the importance of understanding the intricate
330molecular interactions between *P. infestans* and its host and the
331necessity for continued research and development of effective
332management strategies to mitigate the global impact of potato late blight
333on agricultural systems (Arora et al., 2014). Combining both dimeric and
334monomeric analyses, this integrative approach ensured a comprehensive
335understanding of ACCA's interaction with LPMOs. While the dimeric form
336provided a broader perspective on potential interaction sites, it primarily
337served as a benchmark for initial studies. The focus on monomeric
338functionality ensured physiological relevance, thereby validating ACCA's
339inhibitory potential in a realistic enzymatic context.

340Future studies should further prioritize monomeric analyses to reflect the
341physiological state of LPMOs accurately. Experimental validation, such
342as enzyme inhibition assays and structural characterization of ACCA-
343LPMO complexes in their monomeric form, will be essential. Additionally,
344employing the dimeric form as a comparative benchmark can help

345 identify and confirm critical interaction sites that are consistent across
346 both forms. This approach will provide deeper insights into the dynamics
347 of ACCA binding and its inhibitory mechanisms, enhancing our
348 understanding of its potential applications in pest control and
349 therapeutic interventions.

3505. Conclusions

351 The discovery of the tremendous potential of 1-amino-cyclopropane-1-
352 carboxylic acid (ACCA), an ethylene precursor, in enhancing *Solanum*
353 spp. immunological response against *P. infestans* highlights the
354 importance of exploring novel treatments to combat this disease. The
355 promising binding value of -8.85 kcal/mol for ACCA suggests that
356 exogenous application of this compound could bolster potato plant
357 defenses against PLB. This study provides a promising foundation for
358 developing ACCA-based treatments for PLB, potentially transforming
359 sustainable potato farming practices.^[0] This research contributes to our
360 understanding of effective PLB management strategies and underscores
361 the need for continued investigation and innovation to overcome the
362 disease's limitations on both local and continental levels. By leveraging
363 cutting-edge research technology and interdisciplinary collaboration, we
364 can develop effective, sustainable solutions to mitigate the impact of PLB
365 on global potato production, ensuring food security for future
366 generations.

367

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