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Protocatechuic acid mitigates CuO nanoparticles-induced toxicity by strengthening the antioxidant defense system and suppressing apoptosis in liver cells

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ABSTRACT

Due to exceptional properties, copper oxide nanoparticles (CuO NPs) are applied in diverse fields, including agriculture, cosmetics, food industry, and medicine. There is enough scientific evidence that CuO NPs are toxic to living organisms. However, no effective method is available to mitigate the health hazard of the CuO NPs. This study aims to explore the possible cytoprotective mechanism of protocatechuic acid (PCA) from CuO NPs induced toxicity in human liver cells (HepG2) and primary rat hepatocytes. CuO NPs are prepared by a simple hydrothermal method. X-ray diffraction and electron microscopy characterization data confirmed the formation of spherical and crystalline CuO NPs with smooth surface morphology. On the basis of preliminary results, cells are divided into four groups; (i) control group, (ii) PCA group (10 μ g/mL, a non-cytotoxic dose), (iii) CuO NPs group (IC₅₀), and (iv) co-exposure group (PCA+CuO NPs). Results show that pre-treatment with PCA significantly abrogates the CuO NPs-induced cytotoxicity and lactate dehydrogenase leakage in both types of cells. CuO NPs-induced reactive oxygen species, hydrogen peroxide, and malondialdehyde levels are remarkably mitigated by the PCA pre-treatment. The depletion in glutathione and several antioxidant enzymes (e.g., glutathione peroxidase, superoxide dismutase, and catalase) because of the CuO NPs exposure is effectively restored by PCA. Caspase-3 gene up-regulation and mitochondrial membrane potential depletion due to CuO NPs exposure are also alleviated by the pre-treatment using PCA.^[1] Therefore, this first study reports the mitigating potential of PCA from CuO NPs toxicity by restoring the antioxidant defense system and suppressing apoptosis in liver cells.

Keywords: Protocatechuic acid; CuO NPs; Liver cells; Antioxidants; Apoptosis; Cytoprotective mechanism

1. Introduction

Copper oxide nanoparticles (CuO NPs) have received great attention for various applications because of their unique thermal, electrical and optical properties (Islam et al., 2020). Due to the unique physicochemical characteristics, CuO NPs are used in diverse fields, including gas sensor, solar cell, catalysis, agriculture, cosmetics, paints, textiles, environmental remediation, food industry, and biomedical purposes (Khalaj etal., 2018). It is estimated that from 2020-2025 the world will consume approximately 200-830 tons of CuO NPs each year (Ahamed et al., 2022).^[27] The huge demand and diverse application of CuO NPs will significantly enhance the chance of their release into the environment and cause potential hazards to human health and the eco-system (Bialy et al., 2020).

The toxicological potential of CuO NPs has been comprehensively studied in a series of cultured cells, and in vivo models. CuO NPs can be toxic to various mammalian cells including lung, skin, breast, brain, kidney, neuronal, intestine, and liver cells, using oxidative stress pathways (Abudayyak et a., 2020; Fahmy et al., 2020). A recent in vivo study demonstrated that an oral dose of CuO NPs induces toxicity in the hepatic, renal, and splenic tissue of mice (Bialy et al., 2020). Another study observed that the oral exposure of rats to CuO NPs induces liver toxicity by free radical generation (Anreddy, 2018).^[23] The toxicity of CuO NPs was also reported in non-mammalian in vivo systems.^[1] For example, oral exposure to CuO NPs disrupts the gut homeostasis of Drosophila melanogaster, and oxidative stress was one of the primary causes of toxicity (Baeg et al., 2018).

There is no effective method for reducing the CuO NPs exposure; hence, humans can be regularly exposed to CuO NPs from the workplace, environment, and contaminated food and water. It is a relevant concern of researchers to develop new methods to minimize the hazardous effects of CuO NPs in humans.^[34]CuO NP toxicity is closely linked with the generation of free radicals and oxidative stress mechanism, therefore, the supplementation of antioxidants, as a therapeutic strategy, could be a possible way to attenuate CuO NPs-induced toxicity in humans. Protocatechuic acid (PCA, 3,4-dihydroxybenzoic acid) can prevent the toxic effects of CuO NPs. PCA is a kind of phenolic compound widely present in several edible nuts, vegetables, and fruits, and it is easily absorbed by animals and humans (Antony and Wasewar, 2020; Owumi et al., 2019). PCA has shown several biological activities, including antibacterial, antidiabetic, antioxidant, and anti-inflammatory properties (Antony and Wasewar, 2020; Semaming et al., 2015). Moreover, PCA has a structural resemblance with caffeic acid, gallic acid, and vanillic acid, which are known for antioxidant activities (Kakkar and Bais, 2014).^[29] There is growing evidence showing that the pharmacological activities of PCA are closely linked with the regulation of oxidative stress, inflammation, and apoptosis (Semaming et al., 2015). A recent finding indicated that PCA mitigates Cd-induced neurotoxicity in rats using its antioxidant activity (AlOlayan et al., 2020).

PCA is investigated for possible therapeutic actions against several human ailments; however, data are scarce on its protective mechanisms against metal oxide NPs-induced toxicity. The liver is one of the most targeted organs of metal oxide NPs after they enter the body through any possible route (Lee et al., 2016). Hence, this work was planned to study the cytoprotective effects of PCA from CuO NPs-induced toxicity in human liver cells (HepG2) and primary rat hepatocytes. The possible mitigating mechanism of PCA from CuO NPs induced toxicity was also delineated using oxidative stress and apoptosis pathways.

2. Materials and methods

2.1.^[0] Synthesis and characterization of CuO NPs

CuO NPs was prepared by a simple hydrothermal method using copper nitrate (CuNO₃.2H₂O) (Sigma-Aldrich, St. Louis, MO, USA) and sodium hydroxide (NaOH) (Ahamed et al., 2022).^[0] The phase purity and crystalline nature of CuO NPs were assessed by X-ray diffraction (XRD) (PanAnalytic X`Pert Pro) (Malvern Instruments, WR14 1XZ, UK) using Cu-K α radiation (λ = 0.154 nm at 45kV and 40 mA).^[7] Structural characterization was further done by field emission scanning electron microscopy (FESEM, JSM-7600F, JEOL, Inc., Tokyo, Japan) and field emission transmission electron microscopy (FETEM, JEM-2100F, JEOL).

¹²The hydrodynamic size and zeta potential of CuO NPs and CuO NPs + PCA suspensions in distilled water and complete culture medium (DMEM+10%FBS) were examined by dynamic light scattering (DLS) (ZetaSizer Nano-HT, Malvern Instruments). ^[1]CuO NPs (IC₅₀) and CuO NPs (IC₅₀) + PCA (10 µg/mL) were suspended in distilled water and culture media. Subsequently, the suspensions were sonicated for 15 min at 40 W using a water bath sonicator and DLS experiment was performed.

2.2^[89] Cell culture

HepG2 cell line was purchased from American Type Culture Collection (ATCC) (Manassas, VA, USA). Primary rat hepatocytes were isolated by a collagenase perfusion procedure as reported by Moldeus and co-workers (Moldeus et al., 1978). Cells were grown in Dulbecco's modified eagle's medium (DMEM) (Invitrogen, Carlsbad, CA, USA) or

Roswell Park Memorial Institute (RPMI) 1640 medium (Sigma-Aldrich) supplemented with streptomycin (100 µg/mL, Invitrogen) and penicillin (100 U/mL, Invitrogen), and 10% fetal bovine serum (FBS, Invitrogen). The cells were maintained with 5% CO₂ supply at 37 °C in a humidified incubator.

2.3.^{[19]▶} Exposure protocol and selection of the appropriate concentrations of CuO NPs and PCA

Sock solution (1 mg/mL) of synthesized CuO NPs and purchased PCA (Sigma-Aldrich) was prepared in culture medium. First, we performed a screening 3-(4,5dimethylthiazol-2-yl)-2,5-diphenyltetrazoliumRbromide (MTT) assay to choose the appropriate dosages of CuO NPs and PCA. HepG2 cells and primary rat hepatocytes were individually treated to various concentrations of CuO NPs (0, 1, 2^[29], 5, 10, 25, 50, and 100 µg/mL) and PCA (0, 1, 2^[29], 5, 10, 25, 50, and 100 µg/mL) for 24 h. Results showed that PCA did not induce cytotoxicity in the selected concentrations in both types of cells (Fig. S1, Supplementary material). Conversely, CuO NPs induced dose-dependent cytotoxicity in the concentration range of 5-100 µg/mL in both types of cells (Fig. S3, Supplementary material). The IC₅₀ values of CuO NPs were 26.56 µg/mL and 29.66 µg/mL for HepG2 cells and primary rat hepatocytes, respectively (Fig. S3, Supplementary material). Therefore, in coexposure experiments we have the selected IC₅₀ value of CuO NPs. To further choose the appropriate concentration of PCA, we performed experiments where different concentrations of PCA (1-100 µg/mL) are co-exposed with the IC₅₀ value of CuO NPs (Fig. S4, Supplementary material).^[1]Results showed that cell viability declined to approximately 50% following the exposure to CuO NPs (IC₅₀), and treatment with PCA increased the cell viability in a dose-dependent manner in the concentration range of 1-10 µg/mL.^[0] Hence, our data suggested that treatment with PCA at the concentration of 10 µg/mL achieved maximum mitigating effect against CuO NPs (IC₅₀) induced cytotoxicity in both cells. This indicated that 10 µg/mL PCA was enough to mitigate the IC₅₀ toxicity of CuO NPs (Fig. S4, Supplementary material). Hence, for this study, we utilized 10 µg/mL of PCA to explore its mitigating potential against CuO NPs (IC₅₀) induced toxicity. The cells were divided into four groups; control group, PCA group (10 µg/mL), CuO NPs group (IC₅₀), and co-exposure group (PCA+CuO NPs). In the co-exposure group, 10 µg/mL of PCA was exposed to cells 30 min before the treatment of CuO NPs (IC₅₀). Exposure time in all experiments was 24 h.

2.4. Biochemical parameters

MTT assay (Mosmann, 1983) with few modifications (Ahamed et al., 2011) was applied to measure cell viability. Lactate dehydrogenase (LDH) enzyme leakage from

cytoplasm to culture media after membrane disintegration was assayed as described earlier (Welder, 1992). The cell morphology of the control and treated cells was captured using a phase-contrast microscope (Leica Microsystems, GmbH, Wetzlar, Germany). Images were grabbed at 20× magnification. Red-orange cationic fluorescent dye tetramethylrhodamine, metyl ester, perchlorate (TMRM) (Thermo Fisher Scientific, Waltham, MA, USA) is rapidly taken up by mitochondria in a potential dependent manner. Hence, TMRM probe was applied to assess the mitochondrial membrane potential (MMP) level in treated and control cells. MMP level was determined by two different methods; quantification by a microplate reader (excitation/emission wavelength: 548/574 nm) (Synergy-HT, BioTek, Winooski, VT, USA) and the intracellular brightness of TMRM was captured using a DMi8 fluorescent microscope using green excitation filter (detecting red-yellow TMRM emission) (Leica Microsystems, **GmbH**, Wetzlar, Germany).^[31] Probe 2,7-dichlorofluorescein diacetate (DCFH-DA, Sigma-Aldrich) was utilized to measure the intracellular level of reactive oxygen species (ROS) as reported earlier (Siddiqui et al., 2013). The quantitative analysis of ROS was performed using a microplate reader (excitation/emission wavelength: 485/520 nm) (Synergy-HT, BioTek). The fluorescent images of intracellular ROS were also capture using a DMi8 fluorescent microscope using blue excitation filter (detecting green DCF emission) (Leica Microsystems). The fluorometric assay of intracellular hydrogen peroxide (H_2O_2) was performed using a commercial kit (MAK164 green fluorescence, Sigma-Aldrich).^[0] The fluorometric assay of the caspase-3 enzyme was carried out using the 7-amido-4trifluoromethylcoumarin (AFC) standard (Akhtar et al., 2020).^[7] The expression level of the mRNA of the caspase-3 gene (CASP-3) was determined using a real-time polymerase chain reaction (PCR, ABI PRISM, 7900HT Sequence Detection System) (Applied Biosystems, Foster City, CA, USA) according to the procedures described earlier (Ahamed et al., 2011). ^[38]▶ Malondialdehyde (MDA), an end product of lipid peroxidation was determined using methods in a previous study (Ohkawa et al., 1979). Glutathione (GSH) level was quantified using Ellman's method (Ellman, 1959). The activity of the glutathione peroxidase (GPx) enzyme was measured using the protocol of Rotruck and co-workers (Rotruck et al., 1973). The enzymatic activity of catalase was assayed according toRthe procedure of Sinha et al. (Sinha, 1972).^[2] The colorimetric assay of superoxide dismutase (SOD) enzyme was carried out using a kit from Cayman Chemical Company (Ann Arbor, MI, USA). Protein content was quantified using Bradford's protocol (Bradford, 1976).

2.5^[2] Statistical analysis

One-way analysis of variance (ANOVA) followed by Dennett's multiple comparison tests was applied for the analysis of results. The p 0.05 was assigned as statistically significant.^[7] All the quantitative data presented in this study were mean \pm SD of five independent experiments (n = 5).

3. Results

3.1^[1]▶ Characterization of CuO NPs

The XRD pattern of prepared CuO NPs is shown in Fig. 1A^[78] If the diffraction peaks noticed at 2 were matching to the planes (-110), (002), (111), (-202), (020), (202), (-113), (-311), (320), (311), and (22-) that were matched with Joint Committee on Powder Diffraction Standards (JCPDS) card no. 45-0937, indicating the monoclinic structure of CuO NPs (Arun et al., 2020).^[0] The particle size of the CuO NPs was estimated from the strongest peak (111), applying Scherrer's formula.^[0] The average particle size of the CuO NPs was approximately 33 nm. Impurity peaks were not detected, which further confirmed the purity of the synthesized CuO NPs.^[64] The high intensity of the peaks indicates the high crystallinity of the prepared CuO NPs. FESEM (Fig. 1B) and FETEM (Fig. 1C) micrographs indicated that the prepared CuO NPs had spherical morphology with smooth surfaces.^[4] The average particle size calculated from TEM (100 NPs) was approximately 31 nm, which corresponded with the XRD results. In the high-resolution TEM micrograph (Fig. 1D), the calculated inter-planner distance was 0^[0] 54 nm, corresponds to the (002) plane and is consistent with the monoclinic structure of the CuO phase.

^[51] The DLS study showed that the hydrodynamic size of the CuO NPs was several times larger than that determined by XRD and TEM (Table 1). The relatively large size of the NPs in the aqueous medium was due to the agglomeration of the NPs. Zeta potential values (26-30 eV) showed that the CuO NPs suspensions were fairly stable in distilled water and culture media. Zeta potential data further indicated that the CuO NPs suspended in water displayed a positive surface charge, while that in complete culture media (DMEM+10% FBS) exhibited a negative surface charge. The negative surface charge of the CuO NPs in the culture medium might be due to the adsorption of negatively charged proteins on the surface of CuO NPs. ^[27] The presence of PCA in the CuO NPs suspension in distilled water and culture medium did not affect the aqueous behavior of CuO NPs.

3.2^[0] PCA mitigates CuO NPs-induced cytotoxicity in HepG2 cells

HepG2 cells were exposed for 24 h to CuO NPs (IC₅₀) and/or PCA (10 μ g/mL, a noncytotoxic concentration) and cytotoxicity was assessed. Fig. 2A shows that the CuO NPsinduced cell viability reduction was significantly increased by the PCA pre-treatment (p 0.05). Fig. 2B shows that the CuO NPs exposure significantly increased LDH leakage, but these changes were effectively reverted by the PCA pre-treatment. Fig. 3C depicts the morphology of HepG2 cells in control and experimental groups. CuO NPs treated group showed unhealthy cells and relatively low cell density. However, PCA pre-treatment resulted in remarkable improvement in the morphology and density of the CuO NPs treated cells.

3.3.^[43]▶ PCA mitigates CuO NPs-induced oxidative stress in HepG2 cells

HepG2 cells were exposed to CuO NPs and/or PCA for 24 h, and oxidative stress response was evaluated by determining the various parameters of the pro-oxidants and antioxidants.^[0] Fluorescent microscopy images suggested that the brightness of the dichlorofluorescein (DCF) probe increased (an indicator of ROS generation) in the CuO NPs treated cells, and the pre-treatment of the PCA remarkably reduced the ROS generation as the brightness of the DCF probe decreased in the PCA + CuO NPs group (Fig. 3A). Similarly, quantitative data indicated that the increment in the intracellular ROS level because of the CuO NPs exposure was significantly attenuated by the PCA pre-treatment (p 0.05) (Fig. 3B). ^[0] Moreover, the CuO NPs-induced H₂O₂ (pro-oxidant) and MDA (an end product of lipid peroxidation) levels were effectively abrogated by PCA (p 0.05) (Figs. 3C and D). The mitigating potential of PCA against CuO NPs-induced antioxidants depletion was further examined in HepG2 cells.^[0] Results demonstrated that the GSH level and activity of antioxidant enzymes (GPx, SOD, and CAT) were lower in the CuO NPs treated group than in the control group (p 0.05). Additionally, the CuO NPs-induced antioxidant depletion was restored by the PCA pre-treatment (p 0.05) (Figs. 4A-D).

3.4.^[0] PCA mitigates CuO NPs induced apoptosis in HepG2 cells

HepG2 cells were treated to CuO NPs and/or PCA for 24 h, and the apoptotic response was assessed by measuring the MMP level and regulation of the caspase-3 gene. ^[0] Fluorescent microscopy data showed that the brightness of the TMRM probe decreased (an indicator of MMP reduction) in the CuO NPs treated cells, and the pre-treatment of PCA significantly increased the brightness of the TMRM probe (an indicator of MMP gain) (Fig. 5A). Likewise, the quantitative analysis of MMP showed that the CuO NPs-induced MMP loss was significantly mitigated by PCA (p 0.05) (Fig. 5B). Real-time PCR data demonstrated that the down-regulation of the caspase-3 gene (mRNA level) due to the CuO NPs exposure was significantly reverted by the PCA pretreatment (Fig. 5C). To support the mRNA data, the activity of the caspase-3 enzyme was further examined in HepG2 cells

following the exposure to CuO NPs and/or PCA for 24 h.^[4] agreement with the mRNA data, the CuO NPs induced caspase-3 enzyme activity was effectively ameliorated by PCA (Fig. 5D).

3.5^[44] PCA mitigates CuO NPs-induced cytotoxicity, oxidative stress, and apoptosis in primary rat hepatocytes

The protective effects of PCA against CuO NPs-induced toxicity were further explored in primary rat hepatocytes. The cells were treated for 24 h to CuO NPs (IC₅₀ = 29.66 µg/mL) and/or PCA (10 µg/mL). Results showed that the CuO NPs induced cell viability reduction and LDH leakage were significantly alleviated by PCA pre-treatment (p 0.05 for each) (Figs. 6A and B). The CuO NPs induced ROS generation and GSH depletion were also remarkably attenuated by PCA (p 0.05 for each) (Figs. 6C and D). The alleviating potential of PCA from CuO NPs induced apoptosis was further explored in primary rat hepatocytes by measuring the MMP level and caspase-3 enzyme activity. We observed that the CuO NPs induced MMP loss and caspase-3 enzyme activation was significantly suppressed by pre-treatment of PCA (p 0.05 for each) (Fig. 7A and B). In agreement with the HepG2 cells data, these results also indicated the cytoprotective effects of PCA against CuO NPs-induced toxicity in primary rat hepatocytes.

4. Discussion

Due to their tiny size and surface characteristics, metal oxide NPs may cross cell/tissue barriers and accumulate in different vital organs. CuO NPs are one of the most biologically reactive metal oxide NPs (Ahamed et al., 2015; Moschini et al., 2013). Several in vitro and in vivo studies reported the toxicity of CuO NPs in the liver (Abdelazeim et al., 2020; Siddiqui et al., 2013). However, no effective method is known to reduce the CuO NPs toxicity once they enter the human body. This novel study demonstrated the alleviating potential of PCA against the CuO NPs-induced toxicity in liver cells. Results showed that the CuO NPs-induced cytotoxicity in human liver (HepG2) cells and primary rat hepatocytes was remarkably mitigated by PCA pre-treatment. LDH enzyme is located in the cytoplasm of the cell, which is close to the cell membrane. Hence, slight damages in the cell membrane will result in LDH enzyme leakage into the extracellular fluid e.g., the culture media.⁴⁴⁹ The relatively high LDH activity following the CuO NPs exposure in the HepG2 cells and primary rat hepatocytes in the present study indicated the altered cell membrane permeability with the consequences of its leakage from the cytoplasm to the culture media. However, LDH

activity was significantly decreased by the PCA pre-treatment suggesting the cytoprotective role of PCA against the CuO NPs-induced toxicity in liver cells. Earlier studies demonstrated the protective effects of PCA against heavy metals (e.g., Cd) and antitumor drugs (e.g., cisplatin) induced toxicity in rats (Owumi et al., 2019).

^[76] Uxidative stress due to the excessive production of ROS has been involved in the pathogenesis of liver diseases (Ahamed et al., 2015). Although ROS is normally generated by the metabolism of normal cells and plays a crucial role in cell signaling, in liver diseases, an excessive generation of free radicals occurs, which is not scavenged by the antioxidant defense mechanism, causing hepatic tissue damage (Chen et al., 2020). Oxidative stress also plays an important role in the CuO NPs-induced liver toxicity (Siddiqui et al., 2013). It was reported that PCA utilizes its antioxidant activity and scavenges the excessive ROS (Kakkar and Bais, 2014). This study also demonstrated that the CuO NPs-induced pro-oxidant generation (ROS and H₂O₂) in liver cells were effectively abrogated by the PCA pretreatment, which can be associated with the antioxidant activity of PCA. MDA is a major product of membrane lipid peroxidation, and the increased generation of intracellular free radicals can result in overproduction of MDA (Chung, Rekha, Venkidasamy, & Thiruvengadam, 2019). This study showed that the CuO NPs triggered the MDA level in HepG2 and primary rat hepatocytes. Additionally, the pre-treatment of PCA significantly decreased the CuO NPs-induced MDA level, suggesting the anti-lipid peroxidative potential of PCA.

¹²The protective mechanism of PCA against the effects of CuO NPs on the antioxidant defense ability of liver cells was further investigated. GSH is a well-known antioxidant molecule that protects the by the ROS scavenging activity, reduction of peroxides, and restoration of protein thiols in a reduced state (Eteshola et al., 2020). GSH depletion is also known for its association with apoptosis (De Nicola and Ghibelli, 2014). GPx enzyme reduces the lipid peroxides and hydrogen peroxides using GSH as a substrate, producing GSSG.^[7]SOD enzyme acts as the first line of defense by converting superactive superoxide anion (O₂⁻) into H₂O₂. CAT enzyme further catalyzes toxic H₂O₂ into water and molecular oxygen (Gaucher et al., 2018).^[0]Our data demonstrated that the CuO NPs exposure significantly decreased the GSH level and activity of GPx, SOD, and CAT enzymes in the liver cells. Interestingly, the depletion of antioxidant levels in liver cells following the CuO NPs exposure was significantly restored by the PCA pre-treatment.

⁽⁴²⁾ROS also plays an important role in the signaling pathway of apoptosis by inducing the activation of caspases (Yang et al., 2020). Among all the caspases, caspase-3 is the main

executioner caspase that results in the cytoskeleton breakdown and nuclear changes involved in apoptosis (Ahamed et al., 2022). We observed that the up-regulation of the caspase-3 gene (mRNA) following CuO NPs exposure in liver cells was remarkably alleviated by PCA pretreatment.^[4] Similarly, CuO NPs induced caspase-3 enzyme activity was also mitigated by PCA in HepG2 cells and primary rat hepatocytes. MMP is another reliable marker of stress and apoptosis. MMP level is depleted during the apoptosis process (Seo et al., 2020). This study showed that CuO NPs induced MMP depletion in HepG2 cells and primary rat hepatocytes was reverted by PCA, suggesting the anti-apoptotic effects of PCA against the CuO NPsinduced apoptosis.

5. Conclusion

In summary, the cytoprotective effect of PCA from CuO NPs toxicity in liver cells was examined. In HepG2 cells and primary rat hepatocytes in vitro models, CuO NPsinduced cell viability reduction, LDH leakage, free oxygen radical generation, and antioxidants depletion was remarkably mitigated by PCA. MMP depletion and caspase-3 gene activation by CuO NPs exposure were also successfully reverted by PCA. Conclusively, the present work provides evidence that PCA protects liver cells from CuO NPs toxicity by restoring antioxidants and reducing apoptotic response. This study warrants future research on the mitigating potential of PCA against CuO NPs-induced hepatotoxicity in animal models.

Conflict of interest

The authors declare no conflict of interest.

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Table 1. Dynamic light scattering characterization of CuO NPs. Data are presented as mean±SD of three independent experiment (n=3)

	Hydrodynamic	Zeta potential
	size (nm)	(mV)
Parameters		

			Distilled water	Culture medium	Distilled water	Culture medium
CuO NPs			97.46±4.56	113.53±4.83	27.45±1.24	-26.36±1.31
CuO	NPs	+	93.57±3.87	105.26±4.37	29.23±1.17	-30.43±1.08
PCA						

Figures



Figure 1. Characterization of CuO NPs. (A) XRD spectra, (B), SEM micrograph, (C) Low resolution TEM micrograph, and (D) High resolution TEM micrograph.



Figure 2. Cytotoxicity of HepG2 cells following exposure to the CuO NPs (IC₅₀) and/or PCA (10 μg/mL) for 24 h. (A) MTT cell viability. (B) LDH enzyme leakage. (C) Cellular morphology (scale bar represents 50 μm). *statistically significant difference from the control group (p 0.05). *statistically significant difference from the CuO NPs group (p 0.05).



Figure 3. Pro-oxidants generation in HepG2 cells following exposure to the CuO NPs (IC₅₀) and/or PCA (10 µg/mL) for 24 h. (A) Fluorescent microscope images of intracellular ROS generation (scale bar represents 50 µm). (B) Quantitative analysis of intracellular ROS level. (C) Intracellular H₂O₂ level. (D) MDA level. *statistically significant difference from the control group (p 0.05). *statistically significant difference from the CuO NPs group (p 0.05).



Figure 4. Antioxidants depletion in HepG2 cells following exposure to the CuO NPs (IC₅₀) and/or PCA (10 µg/mL) for 24 h. (A) GSH level. (B) GPx enzyme activity. (C) SOD enzyme activity. (D) CAT enzyme activity. *statistically significant difference from the control group (p 0.05). *statistically significant difference from the CuO NPs group (p 0.05).



Figure 5.^[2] Apoptotic response of HepG2 cells following exposure to CuO NPs (IC₅₀) and/or PCA (10 µg/mL) for 24 h. (A) Fluorescent microscope images of TMRM (indicator of MMP) (scale bar represents 50 µm). (B) Quantitative analysis of MMP. (C) The mRNA level of caspase-3 gene gene. (D) Caspase-3 enzyme activity. *statistically significant difference from the control group (p 0.05). #statistically significant difference from the CuO NPs group (p 0.05).



Figure 6.^[4] Figure 6.^[4] Cytotoxicity and oxidative stress response of primary rat hepatocytes following exposure to the CuO NPs (IC₅₀) and/or PCA (10 µg/mL for 24 h. (A) MTT cell viability. (B) LDH enzyme leakage. (C) ROS level. (D) GSH level. *statistically significant difference from the control group (p 0.05). *statistically significant difference from the CuO NPs group (p 0.05).



Figure 7.^[2] Apoptotic response of primary rat hepatocytes following exposure to the CuO NPs (IC₅₀) and/or PCA (10 µg/mL) for 24 h. (A) MMP level. (B) Caspase-3 enzyme activity. *statistically significant difference from the control group (p 0.05). *statistically significant difference from the CuO NPs group (p 0.05).

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