



ORIGINAL ARTICLE

N-Acetylcysteine reduces the neurotoxic effects of propionic acid in rat pups



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Received 3 May 2013; accepted 22 August 2013

Available online 11 September 2013

KEYWORDS

Propionic acid;
N-Acetylcysteine;
Lactate dehydrogenase;
Lipid peroxides;
Glutathione

Abstract Propionic acid, a metabolite produced by intestinal bacteria, has been implicated in autism. N-Acetylcysteine is a well-known antioxidant. The present study investigated the protective effects of N-Acetylcysteine on propionic acid-induced neurotoxicity in rat pups. Male Western albino rats were divided into four groups. The first group served as the normal control group, the second was treated with propionic acid (PA group), the third group received propionic acid followed by N-Acetylcysteine (PANA group), and the fourth group received N-Acetylcysteine followed by propionic acid (NAPA group). In the PA group, there was a significant increase in lactate dehydrogenase activity, decrease in glutathione levels, and decrease in sodium levels ($P < 0.05$, all comparisons). Calcium and potassium levels did not significantly change. Additionally, urea and lipid peroxide levels were significantly elevated in propionic acid-intoxicated rats and significantly rescued in the N-Acetylcysteine-treated groups ($P < 0.05$, all comparisons). These results indicate that N-Acetylcysteine can both protect against and treat propionic acid intoxication.

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1. Introduction

Propionic acid is a natural body metabolite that is utilised by most organs and tissues. It is produced as a result of fermentation of undigested carbohydrates by intestinal bacteria (Jan et al., 2002; Al-Lahham et al., 2010; Nyhan et al., 1999; Thompson et al., 1990; Zarate et al., 2004). Propionates also

exist naturally in various foods, especially dairy products. In addition, studies have reported that several bacteria, such as clostridia, produce propionic acid during their natural metabolism (Finegold et al., 2010). Consumption of excessive propionate in children results in irritability, inattention, and restlessness associated with sleep disorders (Dengate and Ruben, 2002). Propionic acid crosses the blood–brain barrier and can cause mild, reversible intracellular acidification, which can affect the release of neurotransmitters including glutamate, dopamine, and serotonin (Gupta and Deshpande, 2008; Song et al., 2004). Propionic acid can increase N-methyl-D-aspartate (NMDA) receptor sensitivity, promote intracellular calcium release, elevate nitric oxide levels, and inhibit Na^+/K^+ ATPase, all of which can influence synaptic transmission and/or neuronal activity (Bronstein et al., 1993; Wajner et al., 2004;

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DeMattos-Dutra et al., 2000). Mitochondrial fatty acid metabolism may be affected by propionic acid because of its interaction with propionyl coenzyme A and sequestration of carnitine (Wajner et al., 2004; Brass et al., 1986). This impaired fatty acid metabolism may be related to autism, which may be a mitochondrial disorder.

N-Acetylcysteine, a derivative of the amino acid L-Cysteine, is currently used as an antioxidant, nutritional supplement, and pharmaceutical drug. It is an excellent source of sulphhydryl groups and can be converted into metabolites that can elevate glutathione synthesis in the body. N-Acetylcysteine is currently used as a mucolytic agent for respiratory diseases. It is also used to treat paracetamol overdoses (Tsai et al., 2005), in which case it acts by promoting detoxification by increasing free radical scavenger activity, thus maintaining normal glutathione levels in hepatocytes. N-Acetylcysteine is a precursor for glutathione, which acts as a free radical scavenger in neurons (Cooper and Kristal, 1997) and improves neuronal survival (Ratan et al., 1994). These functions are performed by preserving mitochondrial function and maintaining the mitochondrial oxidative metabolism by protecting the cytochrome oxidase complex I from NO-mediated damage (Moncada, 2000). N-Acetylcysteine can also act as a heavy metal chelating agent to clear the body of some toxic metals. Additionally, N-Acetylcysteine raises glutathione levels when taken as a supplement by immunocompromised patients.

The purpose of this study was to examine the toxic effect of propionic acid on rat pups and evaluate the potential neuroprotective effects of N-Acetylcysteine on propionic acid-induced neurotoxicity.

2. Materials and methods

2.1. Animals

Twenty Male Western albino rats (approximately 4 weeks old, weighing 45–60 g) were used in this study. They were separately housed at a controlled temperature of 21 ± 1 °C with *ad libitum* access to food and water.

2.2. Experimental design

The experimental design and treatment schedule was as follows:

Group 1: Control rats that did not receive any treatment (Normal healthy control; control group).

Group 2: Rats treated with a neurotoxic dose of propionic acid (purchased from Sigma) (250 mg/kg body weight over three days) to induce autism (PA group).

Table 1 The one-way ANOVA test between the control, PPA and OMEGA groups in LD enzyme (U/L), Sodium (mmol/l), Glutathione (ug/ml), Calcium (mmol/L), Potassium (mmol/l), Urea (mmol/L) and Lipid peroxide (nmole/ml) groups and Dunnett test as multiple comparisons.

Parameters	Groups	Min.	Max.	Percent change	Mean \pm S.D.	P value
LD enzyme (U/L)	Control	102.18	253.79	100.00	160.89 \pm 50.76	0.001
	PPA	112.06	372.45	160.962	258.96 \pm 85.94 ^a	
	N-acetyl protective	102.18	288.41	108.920	175.24 \pm 84.71	
	N-acetyl therapeutic	29.66	125.25	40.563	65.26 \pm 36.70 ^a	
Sodium (mmol/l)	Control	48.99	95.00	100.00	66.94 \pm 17.37	0.001
	PPA	22.00	62.99	61.01	40.84 \pm 13.80 ^a	
	N-acetyl protective	18.20	58.80	54.70	36.62 \pm 15.00 ^a	
	N-acetyl therapeutic	9.70	26.60	24.90	16.67 \pm 7.75 ^a	
Glutathione (ug/ml)	Control	34.48	37.06	100.00	35.57 \pm 0.96	0.001
	PPA	23.27	31.03	78.25	27.83 \pm 3.20 ^a	
	N-acetyl protective	32.32	36.20	96.10	34.18 \pm 1.52	
	N-acetyl therapeutic	20.68	28.01	68.46	24.35 \pm 2.54 ^a	
Calcium (mmol/L)	Control	1.75	3.45	100.00	2.42 \pm 0.54	0.169
	PPA	1.84	2.38	84.37	2.04 \pm 0.17	
	N-acetyl protective	2.15	2.62	96.87	2.34 \pm 0.19	
	N-acetyl therapeutic	2.07	2.39	91.82	2.22 \pm 0.13	
Potassium (mmol/l)	Control	4.70	8.32	100.00	6.40 \pm 1.16	0.166
	PPA	4.46	6.87	88.89	5.69 \pm 0.97	
	N-acetyl protective	6.09	7.49	104.12	6.67 \pm 0.59	
	N-acetyl therapeutic	5.18	7.92	107.48	6.88 \pm 1.02	
Urea (mmol/L)	Control	4.97	8.59	100.00	6.49 \pm 1.23	0.012
	PPA	7.08	10.06	128.54	8.34 \pm 1.20 ^a	
	N-acetyl protective	6.73	8.24	113.66	7.38 \pm 0.58	
	N-acetyl therapeutic	5.96	7.72	107.49	6.98 \pm 0.64	
Lipid peroxides (nmole/ml)	Control	1.02	1.70	100.00	1.40 \pm 0.24	0.001
	PPA	2.33	6.54	295.30	4.13 \pm 1.37 ^a	
	N-acetyl protective	1.28	2.09	115.86	1.62 \pm 0.37	
	N-acetyl therapeutic	1.50	1.81	118.22	1.65 \pm 0.10	

Significant levels between the three groups are illustrated as superscript letters when $P < 0.05$.

Group 3: Rats treated with N-Acetylcysteine (purchased from Sigma) (50 mg/kg body weight over a period of one week), followed by propionic acid intoxication as described above (NAPA group).

Group 4: Rats treated with a toxic dose of propionic acid followed by N-Acetylcysteine (PANA group).

2.3. Ethics approval and consent

The Ethics committee of the College of Science Research Centre of King Saud University, Riyadh, Saudi Arabia approved this study. The approval number is 8/25/220358.

2.4. Blood collection

At the end of the experimental period, the rats were sacrificed using ether anaesthesia. Blood samples were collected from the inferior vena cava of each rat. The serum was separated from the blood cells and stored at -80 °C until analysis.

2.5. Biochemical analysis

2.5.1. Measurement of lactate dehydrogenase (LD)

The activity of LD₅ was measured with the pyruvate to lactate kinetic method and the two-point colorimetric method for lactate dehydrogenase (LD). An increase in the absorbance at 340 nm indicated NADH formation, which was directly proportional to serum LD-L activity. The procedure was based on the modifications of the Amador and Wacker method (Amador et al., 1963; Snodgrass et al., 1959).

2.5.2. Determination of sodium levels

Sodium levels were assayed by enzymatic determination of sodium, i.e., measurement of sodium-dependent galactosidase activity using ONPG as a substrate (Tiez, 1986).

2.5.3. Determination of glutathione levels

Glutathione levels were measured by the development of a relatively stable yellow colour (read at 412 nm) upon addition of 5, 5-dithiobis-2-nitrobenzoic acid (Beutler et al., 1963).

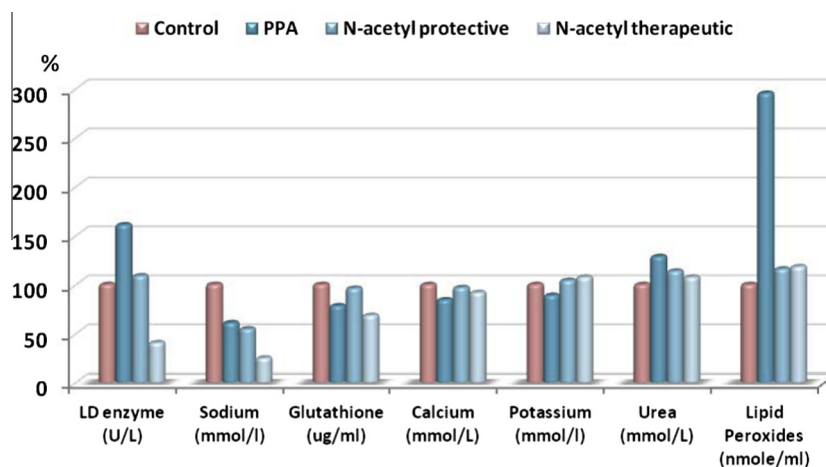


Figure 1 Percentage change of control, PPA, N-acetyl protective and N-acetyl therapeutic groups in LD enzyme (U/L), Sodium (mmol/l), Glutathione (ug/ml), Calcium (mmol/L), Potassium (mmol/l), Urea (mmol/L) and Lipid peroxide (nmole/ml) groups compared to control.

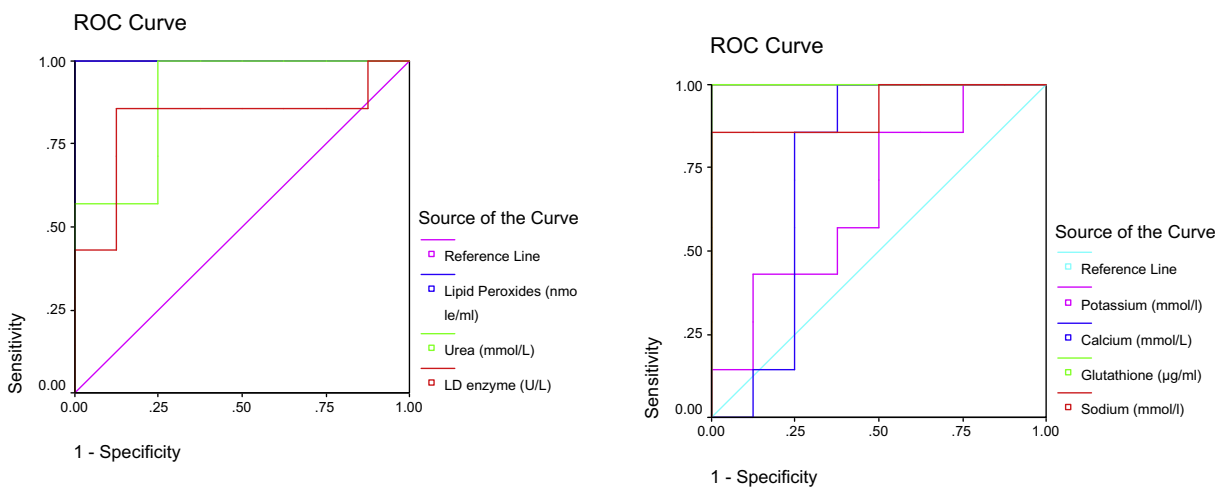


Figure 2 ROC Curve of LD enzyme (U/L), Sodium (mmol/l), Glutathione (ug/ml), Calcium (mmol/L), Potassium (mmol/l), Urea (mmol/L) and Lipid peroxides (nmole/ml) in the PPA group.

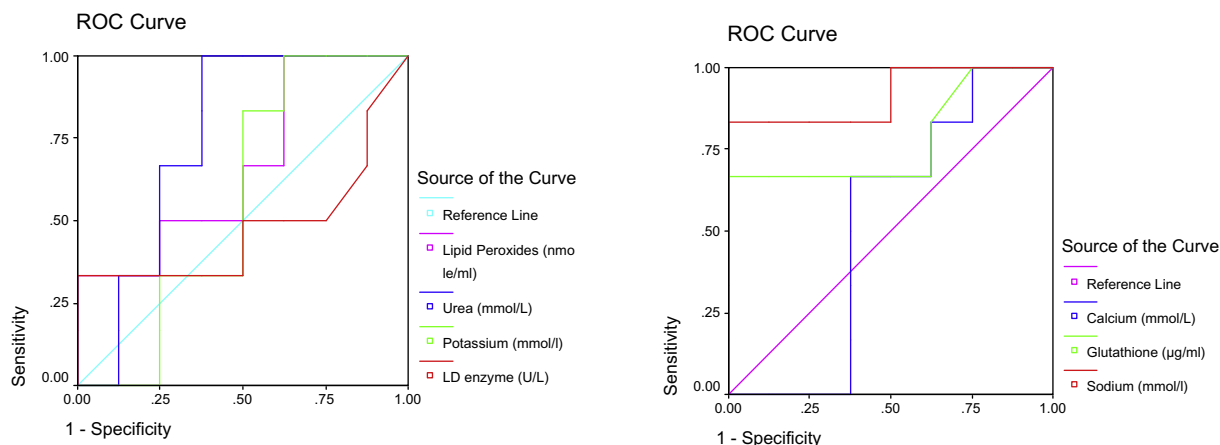


Figure 3 ROC Curve of LD enzyme (U/L), Sodium (mmol/l), Glutathione (ug/ml), Calcium (mmol/L), Potassium (mmol/l), Urea (mmol/L) and Lipid peroxides (nmole/ml) in the N-acetyl protective group.

2.5.4. Determination of calcium levels

Calcium levels were measured using a procedure involving a chromogenic complex that forms upon reaction of o-cresolphthalein complexone (o-CPC) with calcium. The chromogenic complex absorbs light and was therefore measured photometrically between 570 nm and 580 nm (Faulker and Meites, 1982).

2.5.5. Determination of potassium levels

Potassium levels were measured in a protein-free alkaline medium by reaction with sodium tetraphenyl boron, which produced a colloidal suspension. The turbidity of such a

suspension is proportional to the potassium concentrations in the range of 2–7 mmol/L (Terri and Sesin, 1958).

2.5.6. Determination of urea levels

Urea levels were measured using the enzymatic, colorimetric, endpoint-Berthelot method (Kits from Crescent Diagnostics Co., Saudi Arabia), in which the conversion of urea to ammonia was catalysed by urease.

2.5.7. Determination of lipid peroxide levels

Endogenous lipid peroxidation was evaluated by measuring the malondialdehyde concentration at 532 nm and calculating

Table 2 ROC analysis parameters showing area under the curve, cutoff values, sensitivity and specificity of the measured parameters in the four studied groups.

Group parameter	Area under the curve	The curve	Cutoff Value	Sensitivity (%)	Specificity (%)
PPA	LD enzyme (U/L)	0.821	201.880	85.7	87.5
	Sodium (mmol/l)	0.929	47.595	85.7	100.0
	Glutathione (ug/ml)	1.000	32.755	100.0	100.0
	Calcium (mmol/L)	0.750	2.394	100.0	62.5
	Potassium (mmol/l)	0.661	6.510	85.7	50.0
	Urea (mmol/L)	0.893	7.035	100.0	75.0
	Lipid peroxides (nmole/ml)	1.000	2.016	100.0	100.0
N-acetyl protective	LD enzyme (U/L)	0.479	262.031	33.3	100.0
	Sodium (mmol/l)	0.917	47.595	83.3	100.0
	Glutathione (ug/ml)	0.781	34.265	66.7	100.0
	Calcium (mmol/L)	0.521	2.346	66.7	62.5
	Potassium (mmol/l)	0.563	5.866	100.0	37.5
	Urea (mmol/L)	0.750	6.602	100.0	62.5
	Lipid peroxides (nmole/ml)	0.667	1.276	100.0	37.5
N-acetyl therapeutic	LD enzyme (U/L)	0.958	96.408	83.3	100.0
	Sodium (mmol/l)	1.000	37.795	100.0	100.0
	Glutathione (ug/ml)	1.000	31.245	100.0	100.0
	Calcium (mmol/L)	0.667	2.396	100.0	62.5
	Potassium (mmol/l)	0.625	6.846	66.7	75.0
	Urea (mmol/L)	0.677	6.506	83.3	62.5
	Lipid peroxides (nmole/ml)	0.802	1.478	100.0	62.5

the concentration using an extinction coefficient (ϵ) of $156 * 10^5 \text{ M}^{-1} \text{ cm}^{-1}$ (Buege and Aust, 1978).

2.6. Statistical analysis

The one-way analysis of variance (ANOVA) was used for statistical analysis. The values were expressed as means \pm standard deviation. A P value of <0.05 was considered statistically significant. The Receiver Operating Characteristic (ROC) curve is a fundamental tool for evaluating biomarkers. On the ROC curves, the true positive rate was plotted against the false positive rate (100-Specificity) for various cut-off values.

3. Results

Table 1 and Fig. 1 present the neurotoxic effects of propionic acid along with the protective and therapeutic effects of *N*-Acetylcysteine. Propionic acid induced significant increases in serum LD activity and oxidative stress through the induction of glutathione depletion and a significant increase in lipid peroxidation ($P < 0.05$). Propionic acid did not affect calcium or potassium levels but did induce sodium depletion. Urea levels were dramatically elevated in the serum of propionic acid-treated rats. Table 1 shows the protective effect of *N*-Acetylcyste-

ine on propionic acid neurotoxicity. *N*-Acetylcysteine-pre-treated rats had much lower LD activity and lipid peroxidation and higher levels of glutathione. *N*-Acetylcysteine treatment did not affect sodium levels.

Table 2 and Figs. 2–4 display ROC curves for the parameters measured, including the area under the curve, cut-off, and specificity and sensitivity values.

4. Discussion

Propionic acid toxicity is an organic acidemia (Xu et al., 2012) that is often present during the neonatal period along with lethargy, poor feeding, and vomiting. Progression may lead to coma if not recognised and appropriately treated. Table 1 and Fig. 1 show a dramatic increase in LD activity in the plasma of propionic acid-treated rats compared to the control group. The increase in LD in the serum could be due to the leakage of enzyme from the brain into the plasma, which may indicate cellular and membrane damage. El-Ansary et al.'s (2011) study supports this observation. They found a remarkable decrease in LD activity in the brain homogenates of propionic acid-treated rats compared to untreated controls (El-Ansary et al., 2011). The outcomes of our study indicate that *N*-Acetylcysteine is effective against propionic acid-induced neurotoxicity (Table 1). *N*-Acetylcysteine has both pro-

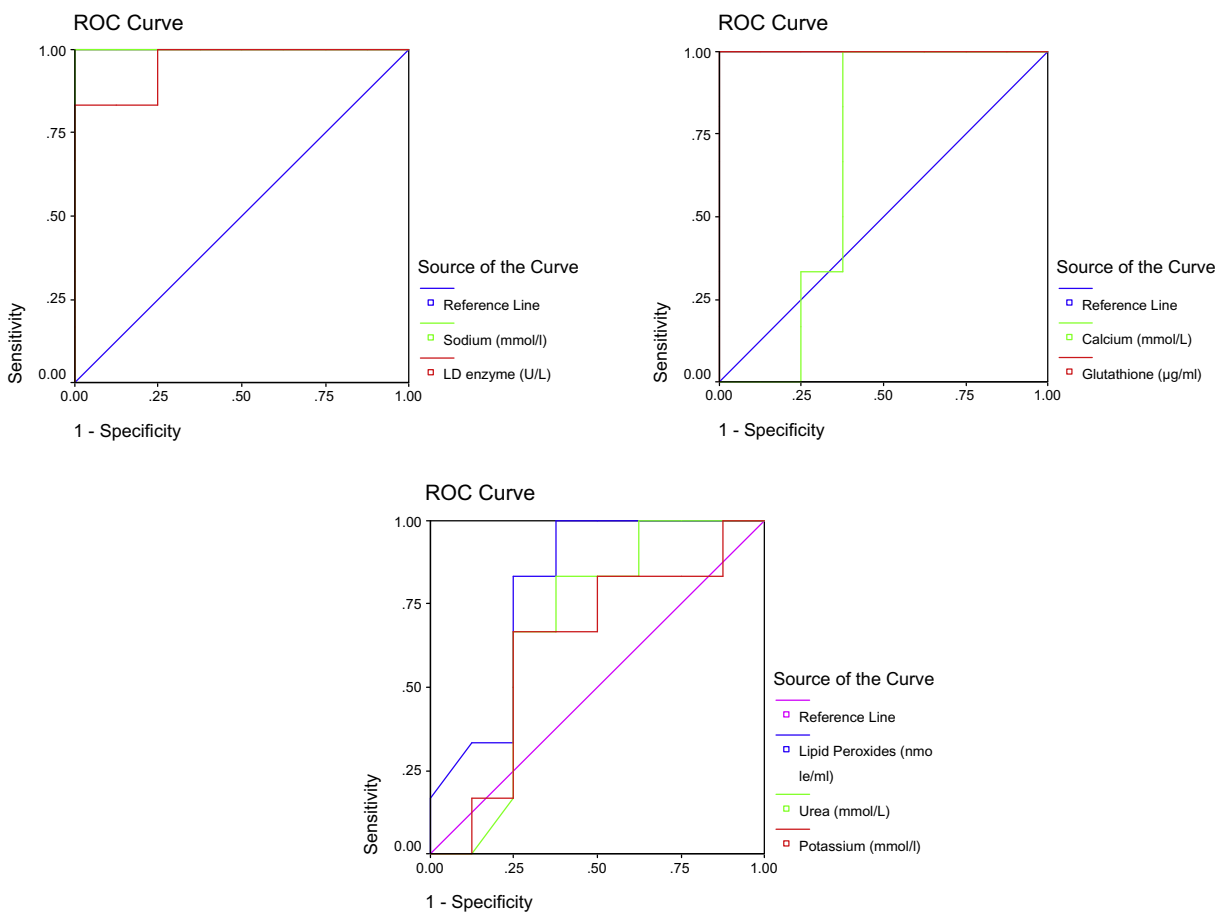


Figure 4 ROC Curve of LD enzyme (U/L), Sodium (mmol/l), Glutathione (ug/ml), Calcium (mmol/l), Potassium (mmol/l), Urea (mmol/L) and Lipid peroxides (nmole/ml) in the *N*-acetyl therapeutic group.

protective and therapeutic effects, as indicated by the rescued levels of LD activity in the serum. Hsu et al. (2006) showed that pre- versus post-treatment with N-Acetylcysteine decreases plasma LD as a marker of organ injury induced during sepsis. Additionally, the protective effects shown in our study agree with Abd El-Fattah and El-Sheikh (2012), who reported that N-Acetylcysteine partially corrects propionic acid-induced changes in enzyme activities and protects the kidney from tubular damage.

Increased oxidative damage due to neuroinflammation, mitochondrial dysfunction, and impaired GSH metabolism may be related to the development of autism (James et al., 2004; Al-Gadani et al., 2009). Oxidative stress has been implicated in synaptic plasticity. MacFabe et al. (2007) and El-Ansary et al. (2012) hypothesised that similar processes could arise in animals given an oral or intraventricular neurotoxic dose of propionic acid. We have found that propionic acid treatment induces a substantial increase in lipid oxidation, indicating increased oxidative stress. We found that N-Acetylcysteine exerted a protective effect on GSH depletion induced by propionic acid and a smaller effect when given after propionic. Increased lipid peroxide levels serve as a marker for the neurotoxic effects of propionic acid. The antioxidant effect of N-Acetylcysteine was confirmed by its ability to decrease the concentration of lipid peroxides. Hence, N-Acetylcysteine exerts both protective and therapeutic effects. The effectiveness of N-Acetylcysteine is due to increases in cysteine levels that increase the available pool of glutathione, compared to other antioxidants examined in autism. Previous studies support this suggested antioxidant effect. For example, Gibson et al. (2009) concluded that N-Acetylcysteine is a weak antioxidant that requires prior conversion into GSH to impart antioxidant and antithrombotic benefits at therapeutic levels. Additionally, N-Acetylcysteine acted as an antiplatelet agent only when the intra-platelet activity that converts it into GSH is functional (Gibson et al., 2011). The failure of N-Acetylcysteine to exert a therapeutic antioxidant effect in the present study may suggest that propionic acid irreversibly impaired the glutathione synthesis pathway.

The effectiveness of the low dose of N-Acetylcysteine (50 mg/kg/day) used in the present study concurs with a previous animal study by Sprong et al. (1998). In their study, high-dose N-Acetylcysteine treatment (550 and 950 mg/kg in 48 h) decreased lung glutathione (75 mg/kg/day) and resulted in a significantly lower survival rate compared to low-dose N-Acetylcysteine (Sprong et al., 1998).

Our results indicate that hyponatraemia could be one of the neurotoxic effects of propionic acid (Table 1). Although plasma calcium and potassium levels were not significantly altered by propionic acid, plasma sodium levels were dramatically decreased. We found that N-Acetylcysteine was not effective at ameliorating propionic acid-induced hyponatraemia (Table 1). This observation indicates that N-Acetylcysteine is likely to work through the proposed anti-oxidative mechanism. Additionally, the weak therapeutic effect of N-Acetylcysteine in the present study concurs with a study by El-Ansary et al. (2012) on the persistent neurotoxic effects of propionic acid and the induction of biochemical autistic features in rats.

Intraperitoneal administration was selected to test the role of the gut-brain axis in the neurotoxicity of propionic acid and the possibility of treating autistic patients through oral administration of N-Acetylcysteine.

5. Conclusion

The outcomes of this study indicate that N-Acetylcysteine could be used to both protect against and treat propionic acid intoxication.

Acknowledgement

This research project was supported by a grant from the research centre of the Centre for Female Scientific and Medical Colleges at King Saud University.

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