Journal of King Saud University - Science 35 (2023) 102904

Contents lists available at ScienceDirect

Journal of King Saud University – Science

journal homepage: www.sciencedirect.com



Original article

Broad-spectrum antibacterial and antibiofilm activity of biogenic silver nanoparticles synthesized from leaf extract of *Phyllanthus niruri*



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ARTICLE INFO

Article history: Received 24 May 2023 Revised 9 September 2023 Accepted 11 September 2023 Available online 15 September 2023

Keywords: Green Synthesis Phyllanthus niruri Silver nanoparticles Antibacterial Drug-resistant bacteria Biofilm inhibition

ABSTRACT

Multidrug resistance (MDR) among pathogenic bacteria is a global health concern as it has rendered the current antibiotic therapy ineffective. A recent and rapidly developing research area in the biomedical sector is the use of biogenic nanoparticles as antibacterial agents. In the current investigation, aqueous leaf extract of Phyllanthus niruri (Pn-AgNPs) were employed for the biogenic synthesis of silver nanoparticles, further characterized by SEM, EDX, FTIR, TEM, UV-Vis and XRD techniques. Electron microscopy images of the green-synthesized Pn-AgNPs showed them as evenly dispersed spherical particles having an average size of about \sim 20 nm. The Pn-AgNPs were further tested for potential antibacterial activity against isolated drug-resistant pathogenic bacteria such as Burkholderia cepacia complex, carbapenemresistant Enterobacterales (CRE), Citrobacter freundii, Enterococcus faecalis, Escherichia coli, Enterobacter clocae, MRSA, Salmonella typhi, Streptococcus spp, and vancomycin-resistant Enterococci (VRE). The Pn-AgNPs significantly inhibited growth all of the studied bacterial isolates, including antibiotic-resistant isolates such as MRSA, VRE, and CRE. Electron micrographs clearly showed the enhanced penetration of AgNPs leading to disruption of cell membrane causing death of the cell. Further, Pn-AgNPs dramatically reduced the ability of P. aeruginosa, Bacillus cereus, E. coli, and S. aureus to form biofilms. Leakage of protein from bacterial cells could be main reason for the effective bactericidal action of synthesized AgNPs. This antibacterial and antibiofilm potential of Phyllanthus niruri biosynthesized AgNPs against clinically significant drug-resistant bacteria is impressive and can be developed as a novel antimicrobial agent to combat the threat of drug-resistant bacterial infections.

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1. Introduction

Emergence and rapid spread of multidrug resistance (MDR) among pathogenic bacteria is a big issue globally as it is a threat to the public health and incurs huge financial losses (Maheshwari et al., 2016). The unabated and uncontrolled use of

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Peer review under responsibility of King Saud University.



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the antibiotics is the one of the major causes for the increasing resistance among pathogenic bacteria (Qais et al., 2019). Moreover, the development of new antibiotics is a slow, time taking process that requires tremendous labour and huge capital investment (Lee et al., 2019). Bacteria such as *Acinetobacter baumanii*, ES β L-producing *E. coli*, M β L-producing *Pseudomonas* sp., vancomycin resistance *Enterococcus* (VRE), Methicillin resistant *S. aureus* (MRSA) are the most common persistent infection causing drug-resistant pathogens (Alekshun and Levy, 2007). Emergence of these MDR bacteria have rendered the current antibiotic therapy ineffective and there is an urgent need to develop or discover antibacterial agents with broad-spectrum activity.

Nanotechnology is a discipline that focuses on the creation and manipulation of nanoparticles that can be used in a variety of contexts, including biomedical applications. The growing field of nanoscale applications has currently transformed the entire world

https://doi.org/10.1016/j.jksus.2023.102904

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(Prasad et al., 2017). The metallic nanomaterials/nanoparticles possess distinct, genuine biochemical and physical characteristics that differ from their counterpart bulk materials. Nanoparticles of silver, gold, zinc, copper etc has been of much interest in the past few years for various biomedical applications (Rafique et al., 2017). Silver nanoparticles (AgNPs) have attracted most attention and are reported to demonstrate antibacterial, antifungal, antiviral, antioxidant and anti-inflammatory properties (Khandel et al., 2018). Traditionally AgNPs of desired shapes and sizes were synthesized using physical and chemical methods to produce nanoparticles but the use of highly toxic chemicals, high pressure and energy along with the formation of toxic end products has limited the use of such protocols and compelled the researchers to produce biologically compatible nanoparticles (Bala and Rani, 2020). Therefore, the stress has been to produce metallic nanoparticles using biological methods that are affordable, efficient, ecofriendly, and beneficial in the global effort to eliminate hazardous by-products (Burdusel et al., 2018).

The biological synthesis of metal nanoparticles using plant extract has gained lot of focus in the recent years as phytomediated fabrication of NPs is achieved at room temperature, it is cost effective and rapid as compared to the production of NPs using microbial cultures (Bahrulolum et al., 2021). Plant materials help in reduction of silver ion, act as capping and stabilising agents due to the presence of various metabolites as well as complex biomolecules (Urnukhsaikhan et al., 2021).

Phyllanthus niruri Linn (family: *Euphorbiaceae*) is a common herb of India. Locally called "Bhoomi aavla", it is a popular medicinal plant whose leaves have been used historically in Ayurveda as a medicine to treat various gastroenteritis issues. It is also reported for anti-inflammatory, Hypoglycaemic, antiviral, antioxidant, hepatoprotective, and anti-calculus formation properties (Lee et al., 2016). The plant is rich in secondary metabolites such as acidic diterpenes, arabinogalactan, alkaloids, flavonoids, lignans, terpenes, and tannins (Shilpa et al., 2018). Its phytoconstituents have the capacity to reduce and stabilize the silver nanoparticles.

Considering the well documented medicinal value of *Phyllanthus niruri* and limited work on the silver nanoparticle synthesis from this plant, leaves of this plant were used to produce green AgNPs. Further, since finding novel alternatives to currently used antibiotics has become a critical challenge, the synthesized AgNPs were subjected to detailed characterization and assessed for antibacterial potential against clinically important drug resistant strains of both gram-positive and gram-negative bacteria.

2. Materials and methods

2.1. Extract preparation

The *Phyllanthus niruri* leaves were gathered from the Aligarh Muslim University's Botany Department in Aligarh, India. In the shade, plant leaves were dried. Fine powder from the dried plant leaves was prepared, and 10 gm of powder was stirred into 100 mL of filtered sterilized water. Subsequently, aqueous extract was filtered using 0.22 mm filter paper and Whatman No. 1 filter paper (Maidstone, UK) (Millipore). The heavy biomaterials were extracted in an aqueous state that could be stored at 4 °C for subsequent usage after centrifugation at 10000 rpm for 5 min (Bose and Chatterjee, 2015).

2.2. Synthesis of silver nanoparticles (PA-AgNPs)

In a nutshell, 10 mL of aqueous extract was combined with 100 mL of a 1 mM AgNO₃ solution and maintained for 5–6 h at 50 °C until brown precipitate had formed. This coloured precipitate

was centrifuged for 15 min at 3000 rpm; the pellet was put on a glass plate and dried in an oven for 24–48 h at 45° Celsius. Once completely dry, the powder was pulverized using a mortar and pestle (Fig. 1).

2.3. Uv-vis spectral analysis of Ag-NPs

Synthesized *Pn*-AgNPs were tentatively characterized by recording the absorbance of the UV–Vis spectrum over the 200–800 nm scan range. A UV–Vis dual beam spectrophotometer (Lasany LI295) were observed to record nanoparticle changes in surface plasmon resonance (SPR). To verify that AgNO₃ - ions have been reduced, all spectra were obtained in 1 cm optical path quartz cell at room temperature. The scan range of the sample UV–Visible spectrophotometer was 200–800 nm, with a scanning speed of 480 nm per minute. The spectrophotometer was equipped with "UVProv" software for recording and evaluating data. A blank reference was used to perform baseline correction for the spectrophotometer (Anandalakshmi et al., 2016).

2.4. FTIR analysis

In order to locate functional groups, the material was examined using Fourier transform infrared spectroscopy (FTIR) (Zafar et al., 2020). It is based on the absorption precisely corresponding to the bonds within the molecule. The solid powder sample was used for FTIR analysis using a FTIR spectrophotometer (PerkinElmer Spectrum). The petri dish was put in a forced-air oven, and the samples were dried at 60° C for 96 hrs. One milligram of powder sample was mixed with 100 mg KBr powder (FTIR grade) and finely grinded in a pestle and mortar. The pellets were then prepared using a hydraulic press and tested in permeation mode compared to plain KBr pellets (used as blanks) in 400–4000 cm⁻¹.

2.5. Scanning electron microscopy (SEM) and energy dispersive x-ray (EDX)

SEM equipped with EDX was used to measure the dispersion, chemical make-up and structure of the produced AgNPs. By splattering the glass coverslips, a thin coating of nanoparticles was produced. Using gold coating sputtering, samples were gold-coated. After that, the films were examined at 20 kV voltages.

2.6. Transmission electron microscopy (TEM)

TEM (Jeol) 2100 was used to determine the size and shape of the nanoparticles. The sample (nanoparticle particles) was immobilized on a copper grid and irradiated in vacuum. An electron beam was used to analyse the material.

2.7. Bacterial isolates

The clinical isolates were obtained from patient's samples sent to the Microbiology Department of JN Medical College & Hospital, Aligarh, for routine diagnosis. Samples such as Pus, urine, fluids, and blood, etc were cultured on conventional media as per the laboratory policy and protocols. The growth of the culture was routinely biochemically analysed and identified by the Vitek-2 automated system (Biomerieux, France) using the VITEK 2 ID card. In addition, Mueller Hinton agar was used for antibiotic susceptibility using the Kirby-Bauer disc diffusion method and the Vitek-2 automated system. Clinical bacterial isolates; *Enterococcus faecalis, Streptococcus spp*, methicillin-resistant *Staphylococcus aureus* (MRSA), vancomycin-resistant *Enterococci* (VRE), *Escherichia coli, Enterobacter cloacae*, *Citrobacter freundii, Burkholderia cepacia* complex, *Salmonella typhi*, and carbapenem-resistant *Enterobacterales*



Fig. 1. Diagrammatic representation of the overall process of formation of AgNPs using Phyllanthus niruri herb.

were used in the present investigation. In addition to the above bacteria, sensitive strains such as *E. coli* ATCC 25922 and *Pseudomonas aeruginosa* ATCC 27853 were also used for the antibacterial screening of *Pn*-AgNPs (Kumar et al., 2020).

2.8. Antibacterial activity of Pn-AgNPs

2.8.1. Well diffusion assay

Agar well diffusion method was used to assess the antimicrobial action using respective bacterial isolates that were incubated overnight at 37 °C (Shahid et al., 2007). Bacterial culture (about1–2 × 10^{8} CFU/mL) was lawn-cultured on to Muller-Hinton agar petri dish making use of a sterile cotton swab. Using a sterile borer, 6 mm wells were created on an agar plate and 100 µL of biosynthesized *Pn*-AgNPs were added to each well (Mathew et al., 2018). The controls used were AgNO₃, distilled water, and native plant extract (without synthesized AgNPs). A criterion for determining the effectiveness of the NPs was the inhibition zone around the well (in mm). Tests were performed in duplicate and average results were used.

2.8.2. Minimum inhibitory concentration (MIC)

MIC is the lowest dose value/the least concentration of nanoparticles needed to kill the bacterial growth. The microtiter broth dilution technique was used to calculate the MIC, as reported previously (Alyousef et al., 2021). 96-well ELISA plates were used to perform the experiments, and Triphenlytetrazolium chloride (TTC-3 mg/mL) was used as a cell viability indicator.

2.8.3. Interaction of Pn-AgNPs with bacterial cells examined HR-TEM and SEM

The interaction of Pn-AgNPs with *E. coli* and MRSA was observed using 20 kV high resolution transmission electron microscopy (Jeol 2100)). MRSA and *E. coli* cells were grown at 37 °C in nutrient broth (NB) for 24 h. Cells were centrifuged after incubation, pellet from the culture was dispersed in 1X PBS (phosphate buffer saline). After being dissolved in PBS, the pellet was tested with respective MICx2 of Pn-AgNPs and incubated for 24 h at 37 °C. Subsequently, the treated pellets were centrifuged, the supernatant was removed, pellet was rinsed thrice and fixed in 2.5% glutaraldehyde at 4 °C for 4 h followed by three washes with scaled series of alcohols (30, 50, 70, and 90%). Further, the cells were immersed in white resin overnight for bonding (Ali et al., 2018). Following this, drops of Pn-AgNPs-treated bacterial cells and a control were placed separately on the coverslip for the final examination under TEM and SEM.

2.8.4. Protein leakage assay

Assay to determine the Protein leakage was done employing method described previously (Dharmaraj et al., 2021). Briefly, treated and untreated MRSA and *E. coli* cells were grown in NB at 37 °C. The supernatant was taken at time 0 and after 3 h after the cells were centrifuged at 6200 g for 10 min. Protein concentration was

estimated using Bradford method. Briefly, 800 μ L of Bradford reagent were added to 200 μ L of the supernatant from each sample solution and incubated for 10 min. At 595 nm after incubation, optical density was measured. As a positive control, bovine serum albumin (BSA) was utilized.

2.9. Effect on biofilm formation

2.9.1. Detection of biofilm by tube method

The *Pn*-AgNP's antibiofilm activity was measured qualitatively using the tube method (Kalishwaralal et al., 2010). The turbidity of a 50 μ L overnight grown bacterial culture of the selected bacteria was corrected to 0.5 McFarland standards in Luria Bertani broth. AgNPs (MIC/2) were mixed in separate tubes; then the mixture was placed into tubes with 2 mL sterile LB broth and incubated for 24 h at 37 °C. The experiment also comprised of an untreated control. The insides of the tubes were stained for half an hour using crystal violet (CV) dye (0.1%), surplus dye was eliminated, gently cleaned with distilled water, and then the inhibition of biofilms was assessed.

2.9.2. Quantitative evaluation of inhibition of biofilm formation

The quantitative efficacy of *Pn*-AgNPs in biofilm prevention was determined using a microtiter plate assay (Qais et al., 2020). Overnight grown bacteria were seeded in 96-well microtiter plate wells amended with or without Pn-AgNPs. Post 24 h incubation, cells were removed, well were washed three times with phosphate buffer solution and CV (0.1%) was used to stain the wells. A microtiter plate reader was used to quantify the optical density of dyed biofilms at 620 nm after excess stain was discarded and dye that stained the biofilm was dissolved in 90% ethanol.

2.9.3. Antibiofilm activity of the Pn-AgNPs by SEM

Biofilms were developed on glass coverslips with or without the addition of nanoparticles. Coverslips were given a 2.5% glutaraldehyde treatment after being rinsed with sterile phosphate buffer. The adhering biofilm cells were dehydrated for 10 min in a gradient of ethanol (20, 40, 60, 80, and 100%) solutions. Dried coverslips were coated with gold particles before SEM examination (JEOLJSM 6510 LV) at 20 kV.

3. Results and discussion

Synthesis of nanoparticles from plants is a quick, economical, and environmentally benign process. Characterization of biogenic Pn-AgNPs were done using various spectroscopic and microscopic techniques and assessed for their antibacterial potential against drug resistant pathogenic bacteria. The reaction mixture's noticeable color change is a sign that AgNPs have been synthesized. Addition of *Phyllanthus niruri* extract to AgNO₃ solution tends to start the nanoparticle synthesis within few minutes which is denoted by change of colour in the reaction mixture (Fig. 1). The reduction reaction from Ag + to AgO is shown by turning a colorless solution of silver nitrate into a dark brown color. This rapid change in colour of the solution indicates speedy AgNP synthesis (Fig. 1).

3.1. Uv-vis analysis

Pn-AgNPs' UV–Vis spectra were measured within 200 and 800 nm. (Aadil et al., 2019). The Peak region of *Pn*-AgNPs was observed between 350 nm and 480 nm. The absorbance peak corresponding to *Pn*-AgNPs was ascribed at 425 nm (Fig. 2A). The absorption maxima for AgNPs was observed in a broad region of 425 nm which suggested that the synthesized particles are visible light active. Previous reports have suggested that the spherical

form of AgNPs generally exhibit absorption maxima in the range of 400–440 which is also in support of our study. Further, the appearance of the broad region from 425 to 520 nm suggests that the AgNPs have been successfully capped by the plant extract (Hasan et al., 2020).

3.2. FTIR analysis of nanoparticles

Fig. 2B depicts the FTIR spectra of the biosynthesized Pn-AgNPs to identify the presence of functionality of *P. niruri* leaf extract, that cause the bio-reduction of silver ions (Ag +) and increase the stability of AgNPs. The FTIR spectra exhibited several absorption bands at 687 cm⁻¹ associated with Ag coordinated with –OH and –C = O groups, 1630 cm⁻¹ (-C = O stretching vibrations), 2066 cm⁻¹ due to C = N stretching vibrations and 3437 cm⁻¹ (– OH stretching vibrations) present in the extract of *P. niruri*. Findings of the obtained spectra suggested that the AgNPs were capped through the key functional groups of plant extract which also act as stabilizing agent. These results are supported by the findings reported earlier with porous Ag nanoparticles (Venkatesan et al., 2017).

3.3. XRD

The XRD spectra of biosynthesized Silver NPs, showing distinctive peaks at 29.31, 32.29, 38.08, 44.13, 64.48 and 77.36 corresponding to Miller indices values (100), (110), (111), (200), (220) and (311) simulated with JCPDS No. 01–087-0719 are shown in Fig. 2C. The spectra suggested the NPs with some amorphous nature which evaluated due to stabilization and functionalization through the polyphenolic groups of plant extract. Using the Debye-Scherer formula which is D = $0.9\lambda/\beta$ Cos θ ; where D is crystalline size in nm, λ is wavelength, β is the intensity of half width of maximum intensity line, 18 nm was the observed average crystalline size which was further validated by TEM analysis.

3.4. SEM and EDX analysis

The distribution of nanoparticles was demonstrated using scanning electron microscopic (SEM) images and further their elemental makeup was determined by EDX as depicted in Fig. 3. Micrograph showed the particles to be highly agglomerated and this could be accredited to the centrifugation and drying process applied for the preparation of sample of SEM visualization (Fig. 3A) (Moodley et al., 2018). Due to their surface plasmon resonance and tiny peaks that belong to carbon and oxygen that likely represent the capping structures (surface biomolecules) present in the plant extract, Pn-AgNPs exhibit a specific signal for silver at 3 keV in their EDX examination. (Fig. 3B and C) (Moodley et al., 2018). In the produced nanomaterial, Ag's weight percentage and atomic percentage were, respectively, 74.20% and 28.20%, whereas O's weight percentage and atomic percentage were, respectively, 19.18% and 49.18%. The weight and atomic percentages of C were 6.62% and 22.60%, respectively.

3.5. TEM analysis

The transmission electron microscope (TEM) was also used to characterize biogenic Pn-AgNPs, which revealed spherical-shaped NPs with ~ 20 nm being the mean particle size (Fig. 4). Most of nanoparticles were evenly dispersed but few were observed to form small clumps plausibly due to improper capping or agglomeration. TEM micrograph shows some particles of different shape and size and these variations in nanoparticles produced using plant extracts have been reported previously [2, 39].



Fig. 2. (A). UV-Vis spectrum analysis of silver NPs; (B). FTIR analysis; (C). XRD pattern of Pn-AgNPs.



Fig. 3. SEM and EDAX analysis of AgNPs biosynthesized from *P. niruri*. (A). SEM images of Pn-AgNPs; (B). Weight (%) of atoms present in Pn-AgNPs; (C). EDAX spectrum of Pn-AgNPs.

3.6. Antibacterial activity

Fig. 5 shows that Pn-AgNPs demonstrate bactericidal potential against both standard (ATCC) strains as well as drug

resistant human bacterial isolates like MRSA, VRE, and CRE. Zone of inhibition ranged from 10 to 30 mm against all the test bacteria. Comparable inhibition zones were reported with AgNPs produced from *M. koengii* leaf extract against drug-



Fig. 4. TEM image of Pn-AgNPs.



Fig. 5. Antibacterial activity (zone of inhibition) of biosynthesized Pn-AgNPs against drug-resistant isolates using well diffusion method.

resistant strains (MRSA and *E. coli* producing ESBL) (Qais et al., 2019).

Additionally, the minimum inhibitory concentration (MIC) of Pn-AgNPs was evaluated in terms of its antibacterial activity. The MIC values of green synthesized *Pn*-AgNPs ranged from 10 to

40 µg/mL. These nanoparticles demonstrated MIC values of 40 µg/mL against *Pseudomonas aeruginosa ATCC, Staphylococcus aureus, Streptococcus spp, Salmonella typhi,* MRSA-1, CRE, and MRSA-2; 20 µg/mL against *Escherichia coli, Pseudomonas aeruginosa, Citrobacter freundii*; and 10 µg/mL against *Enterococcus*

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faecalis and Enterobacter clocae, as shown in Table 1. Our research findings agree with those reported with green synthesized AgNPs using *Carum copticum* extract wherein, MIC values of 10, 20, and 30 μg/mL was observed against *C. violaceum*, *S. marcescens* and *P.*

Table 1

Minimum Inhibitory Concentration (MIC) values of *Pn*-AgNPs against tested bacterial isolates.

Bacterial Isolates	MIC value (µg/mL)
Escherichia coli ATCC 25922	20
Pseudomonas aeruginosa ATCC	40
Escherichia coli	20
Pseudomonas aeruginosa	20
Staphylococcus aureus	40
Streptococcus spp	40
Enterococcus faecalis	10
Enterobacter cloacae	10
Citrobacter freundii	20
Salmonella typhi	40
MRSA-1	40
VRE	40
CRE	40
MRSA-2	40

aeruginosa, respectively (Qais et al., 2020). The intrinsic tolerance levels of the strains utilized, shape and size of the nanoparticle, and the MIC measurement method may be responsible for these variations in the measured MICs. It is hypothesized that Pn-AgNPs' antibacterial activity results from Ag's strong adhesion to and penetration of the bacterial cell wall, which releases intracellular components and ultimately causes cell death.

3.7. SEM examination to assess effect of Pn-AgNPs on cell morphology

Scanning electron microscopy (SEM) imaging was performed to decipher as to how Pn-AgNPs interacted with the MRSA and *E. coli* cells. The cell walls of the bacteria treated with Pn-AgNPs showed increasing signs of damage during the experiment. Most cells had rumples and had distorted morphology (Fig. 6). The fact that untreated cells did not exhibit any symptoms of damage provides compelling evidence that Pn-AgNPs are effective at killing bacteria. The key factor contributing to the morphological defects is the rupturing of the cell membrane brought on by treatment with AgNPs (Fig. 6). Pn-AgNPs can cause damage in a number of ways, such as by destroying parts of bacterial cells, tearing the cell membrane, stopping the mechanism that controls microbial respiration, and getting into the cytoplasm [41, 42].



Fig. 6. Scanning electron micrograph of MRSA and *E. coli* grown in the presence and absence of Pn-AgNPs. (Control) Untreated, with intact cell wall, no change in morphology; (Treated) treatment with Pn-AgNPs at 35 µg/ml, cell wall damage, cell distortion and clumping.

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3.8. TEM examination to assess effect of Pn-AgNPs on bacterial cells

The internal structures of untreated (control) MRSA and *E. coli* cells appeared normal, with a cytoplasmic and outer membrane (Fig. 7). After treatment with Pn-AgNPs at respective MICx2, the bacterial cells seem severely injured with the leaking of the cell's interior components is visible in Fig. 7. AgNPs cause cell wall breakage, as well as these nanoparticles enter the cells from various locations, severely damaging the cell and ultimately killing the cells. Our results of the microscopic examination corroborate well with observations made with the in vitro antibacterial activity and MIC determination.

3.9. Protein leakage assay

Effect of Pn-AgNPs on protein leakage in *E. coli* and MRSA cells was determined using Bradford method. Experimental results showed that, compared to the corresponding untreated controls, protein leakage in Pn-AgNPs-treated *E. coli* and MRSA was 4.7 and 5.1 times higher as depicted in Fig. 8. Thus, it is envisaged that the observed antibacterial action of Pn-AgNPs is caused by bacterial cell membrane rupture, which causes Ag to infiltrate the cell wall more frequently, eventually triggering the release of intracel-

Control

lular contents and cell death. (Angamuthu et al., 2023; Dharmaraj et al., 2021).

3.10. Inhibition of biofilm formation

Biofilms have therapeutic importance because they play a role in the bacterial pathogenesis and make the pathogen almost 1000 times more resistant to antibiotics (Hall and Mah, 2017). The persistence of the biofilm is linked to the matrix, which contains the bacteria living in biofilm mode and is mostly made up of EPS, proteins, lipids, and eDNA. [38, 39]. Considering the importance of biofilm in persistent drug-resistant infections, Pn-AgNPs were evaluated for their biofilm inhibitory potential. Biofilm development in the understudy bacteria cultured with and without Pn-AgNPs was evaluated in tubes. Fig. 9A clearly demonstrates the reduced biofilm formation in tubes amended with Pn-AgNPs as compared to the untreated control. Further, the extent of biofilm inhibition was determined using crystal violet assay. Treatment with 10 µg/mL Pn-AgNPs reduced E. coli biofilm formation by 64.70% and P. aeruginosa biofilms by 64.39%. However, a reduction of biofilm by 67.45% and 67.08% was noticed in S. aureus and B. subtilis, respectively, when treated with 20 µg/mL of Pn-AgNPs (Fig. 9B). Findings of CV assay were further confirmed by the

Treated



Fig. 7. Transmission electron micrograph of MRSA and *E. coli* grown in the presence and absence of Pn-AgNPs. (Control) Untreated, with intact cell wall and organelles; (Treated) treatment with Pn-AgNPs at 35 μg/ml, cell wall damage, loosening of cell membrane; lysis and shrinkage and penetration of AgNPs.



Fig. 8. Effect of Pn-AgNPs on protein leakage in test bacterial pathogens.



Fig. 9. Effect of Pn-AgNPs on biofilm formation (A). qualitative assessment using tube method (B). bar graph showing percent inhibition of biofilm. Scanning electron micrograph (SEM) of *P. aeruginosa* biofilm (C). Untreated control; (D). 0.5xMIC of Pn-AgNPs.

SEM visualization of *Pn*-AgNPs treated and untreated cells of *P. aeruginosa* (Fig. 9C-D). As evident from the micrograph, untreated cells were observed to form dense, compact, intact biofilm structure embedded in a matrix of EPS (Fig. 9C). On the contrary, reduced colonization, loosely scattered cells and a disturbed bio-

film architecture with less EPS was observed in *P. aeruginosa* cells treated with 10 μ g/mL concentration of Pn-AgNPs (Fig. 9D).

It is speculated that AgNPs may diffuse and generate antibiofilm activity via water channels in the matrix of biofilms, which are used to transfer nutrients. Nanoparticles have also been shown to bind to and penetrate bacterial membranes, aggregate within bacteria, and kill bacteria (Ansari et al., 2014). AgNPs work by making bacterial membranes more permeable, damaging cells through reactive oxygen species (Al-Shabib et al., 2020), and interacting with the phosphate residues in DNA (Seo et al., 2021), as well as membrane and intracellular proteins (Kim et al., 2007; Morones et al., 2005).

4. Conclusion

These results demonstrate that plant-mediated biosynthesis of AgNPs from the well-known medicinal plant Phyllanthus niruri is a reliable, ecologically friendly, and a safe technique. Biosynthesized AgNPs have especially outstanding optical, morphological, and colloidal characteristics. The green-synthesized Pn-AgNPs demonstrated antibacterial properties against drug-resistant clinical bacterial strains such as MRSA, VRE, and CRE. Moreover, the biogenic AgNPs were also effective in inhibiting the biofilm formed by pathogenic bacteria. It is reasonable to predict that protein leakage from the ruptured cell membranes of the bacterial cells is what is responsible for the antibacterial and antibiofilm activities of the Pn-AgNPs. Due to its broad-spectrum antibiotic potential against clinically significant drug-resistant bacterial strains, the current work provides a novel perspective on the development of biologically produced silver nanoparticles from Phyllanthus niruri as a therapeutic option. Since, Pn-AgNPs also exhibited anti-biofilm properties, it could further be exploited in treatment of biofilm based persistent infections. Further studies are required to determine and unearth the entire molecular pathways that make the synthesized NPs work against bacteria and biofilm.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Acknowledgement

The authors would like to thank the Researchers Supporting Project Number (RSPD2023R729), King Saud University, Riyadh, Saudi Arabia.

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