



Full Length Article

Antibacterial Efficacy of AgNPs synthesized from *Aloe vera* extract and *Staphylococcus aureus* Culture Supernatant

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ABSTRACT

Biological synthesis of silver nanoparticles (AgNPs) is an emerging method that avoids the need for costly equipment and hazardous chemicals. Therefore, in this study, *Aloe vera* (AV) extract and culture supernatant of *Staphylococcus aureus* (CS-*S. aureus*) were used to reduce and stabilise silver nanoparticles. The NPs were characterized using UV-vis spectrophotometry at 445 nm for AV-AgNP and 444.5 nm for CS-*S. aureus* AgNP, and the presence of silver was confirmed by energy dispersive X-ray (EDX). Furthermore, scanning electron microscopy (SEM) and transmission electron microscopy (TEM) revealed that the AV-CS of *S. aureus*-AgNPs had irregular and spherical shapes, with average sizes of 18.5 nm for AvAgNPs and 7.03 nm for CS-*S. aureus*-AgNPs. Additionally, the minimum inhibitory concentration (MIC) of both AgNPs showed a similar reaction. *K. pneumoniae*, *P. aeruginosa*, and *S. epidermidis* were more sensitive to the AgNPs compared to methicillin-resistant *Staphylococcus aureus* (MRSA). In conclusion, our study used a green method to synthesise silver nanoparticles using AV extract and CS of *S. aureus*. The NPs showed a remarkable antibacterial effect, especially CS-*S. aureus*-AgNP, which could have application as a treatment against both Gram-positive and Gram-negative bacteria.

1. Introduction

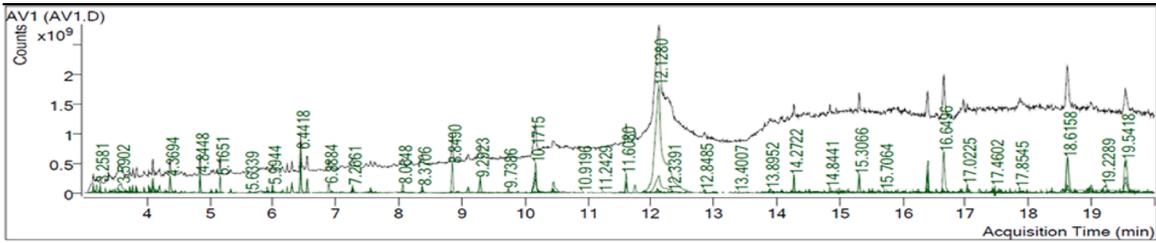
The problem of human pathogens that are resistant to antibiotics is a major concern and a health issue; there is a pressing need for a solution (Anandaradje et al., 2020). The problem of resistance has emerged due to inappropriate use of commercial antimicrobial drugs for treating infectious diseases (Salayová et al., 2021). These resistance mechanisms emerge in bacteria are due to several enzymatic and genetic mutations (Singh et al., 2018; Deljou and Goudarzi, 2016). This situation has forced scientists to look for new antimicrobial therapeutic drugs from different bio-sources, such as medicinal plants and microorganisms, which have been proven to be the best sources (Arbab et al., 2021). Therefore, it is essential to develop novel biological nanomedicine (Radulescu et al., 2023). Nanotechnology is the manufacturing of material at the nanoscale level. It is a rapidly growing platform, as it has applications in science and technology (Singh et al., 2018). Nanoparticles (NPs) are distinguished by their small size (1–100 nm), their unique properties, large surface areas, increased reactivity and ability to enter the human body easily (Huq, 2020). In addition, they have applications in multiple fields such as medicine, deliver of drugs and genes,

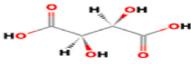
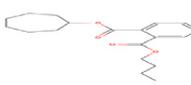
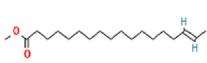
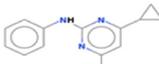
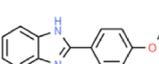
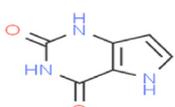
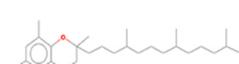
cosmetics, electronics, biosensors and environmental remediation (R. El Shanshoury et al., 2020). Though there are many types of metal nanoparticles, silver nanoparticles (AgNPs) are of particular interest due to their distinct features, such as chemical stability, catalytic activity and optical behaviour (Al Zoubi et al., 2024). They also exhibit broad-spectrum bactericidal and fungicidal activity (Jadoun et al., 2021). Various techniques of synthesising nanoparticles, such as chemical and physical methods, have been documented as being expensive and consisting of hazardous and toxic substances, like stabilisers that are potentially harmful to the environment and biological systems (Radulescu et al., 2023). Hence, an alternative eco-friendly and safe synthesis method was required, which led to the green synthesis of nanoparticles using plants, microorganisms including fungi, yeast, algae and bacteria (Xu et al., 2020). The biological synthesis of AgNPs is considered to be an effective solution, because it is unlikely that pathogens are able to develop resistance against nano-silver without developing a broad range of mutations simultaneously in order to protect themselves (Singh et al., 2018). Using microorganisms to safeguard the environment is an appropriate method. (Saeed et al., 2020). They are also a rich source of bioactive chemicals that are crucial for the

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Table 1
GC-MS analysis of the bioactive constituents of AV extract.



RT (min)	Compound Name	Formula	Area	Structure
3.5515	(R,R)-Tartaric acid	C ₄ H ₆ O ₆	117,674,363	
5.9207	Benzoic acid, 3,5-dihydroxy	C ₇ H ₆ O ₄	33,331,645	
8.5507	Phthalic acid, butyl cyclobutyl ester	C ₁₉ H ₂₆ O ₄	8,455,539	
10.4565	Decanedioic acid	C ₁₀ H ₁₈ O ₄	926,532,760	
11.6080	16-Octadecenoic acid, methyl ester	C ₁₉ H ₃₆ O ₂	507,171,941	
12.1280	Oleic Acid	C ₁₈ H ₃₄ O ₂	17,259,234,829	
15.7064	Cyprodinil	C ₁₄ H ₁₅ N ₃	32,768,685	
16.0453	2-(4-Methoxy-phenyl)-1H-benzimidazole	C ₁₄ H ₁₂ N ₂ O	32,194,148	
17.0225	Pyrolo[3,2-d]pyrimidin-2,4(1H,3H)-dione	C ₆ H ₅ N ₃ O ₂	236,483,815	
17.8593	δ-Tocopherol	C ₂₇ H ₄₆ O ₂	85,691,761	

manufacture of nanomedicines (Singh et al., 2020). Plant-mediated synthesis of AgNPs has several advantages since it is quick, repeatable and ecologically sound. It is also economical and easy to apply on an industrial scale (Rónavári et al., 2021). Moreover, plants have intricate structures that can be employed to stabilise and produce nanoparticles. Plant parts, such as leaves, fruits, stems, seeds and roots, can all be used to make metal nanoparticles (Vanlalveni et al., 2021). Extracts of *Lavandula angustifolia* and *Origanum vulgare* have been used to synthesise AgNPs (Mustapha et al., 2022). Some plant's phyto-constituents, specifically polyphenols, proteins and organic acids, may serve as reducing and capping agents for the surface of nanoparticles (Arshad et al., 2022). Fig S1.

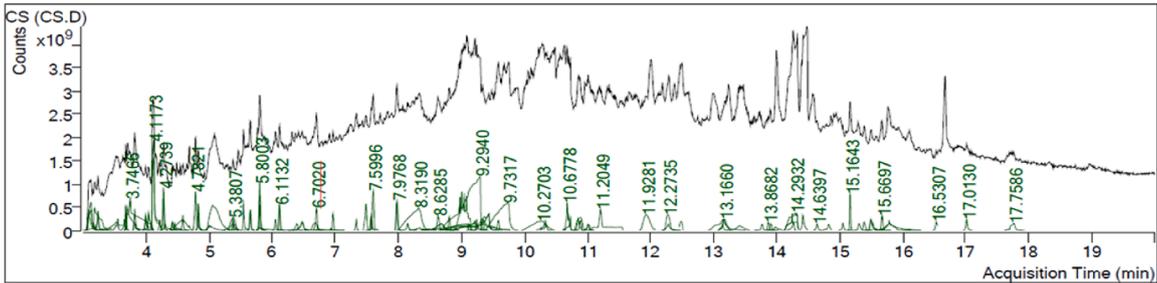
In this study, two distinct biological methods, which are *Staphylococcus aureus* cell-free culture supernatant and AV plant extract, were used to synthesise silver nanoparticles. Then the antibacterial efficacy against pathogenic bacteria of the nanoparticles generated by these approaches was investigated.

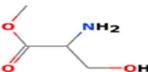
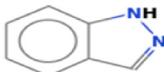
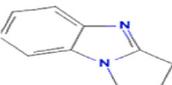
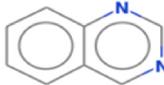
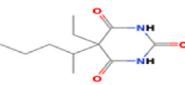
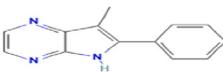
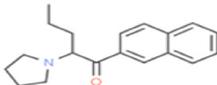
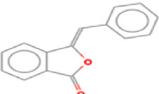
2. Materials and methods

2.1. Collection of experimental materials

AR-grade silver nitrate, AgNO₃ (Sigma-Aldrich Chemicals, USA) and *Aloe vera* plant extract were used as the starting materials. Fresh *Aloe vera* (AV) plant was purchased from a local market in Riyadh. Mueller-Hinton agar (MHA) and Mueller-Hinton broth (MHB), sterilised distilled water and chloramphenicol (50 µg) were used. Meanwhile, bacterial strains including Gram-positive *Staphylococcus aureus* (*S. aureus*) [ATCC 29213], MRSA [ATCC 43300], *Staphylococcus epidermidis* (*S. epidermidis*) [ATCC 12228], and Gram-negative *Klebsiella pneumoniae* (*K. pneumoniae*) [ATCC 700603], *Pseudomonas aeruginosa* (*P. aeruginosa*) [ATCC 27853] were obtained from the King Khalid University Hospital in Riyadh, Saudi Arabia.

Table 2
GC-MS analysis of bioactive constituents of CS- *S. aureus*.



RT (min)	Compound Name	Formula	Area	Structure
3.2368	Piperidine	C ₅ H ₁₁ N	733,754,457	
3.7466	Serine, methyl ester	C ₄ H ₉ NO ₃	6,954,140,744	
4.8249	1H-Indazole	C ₇ H ₆ N ₂	493,740,511	
5.8003	2,3-dihydro-1H-pyrrolo[1,2-a]benzimidazole	C ₁₀ H ₁₀ N ₂	1,311,393,575	
5.8009	Quinazoline	C ₈ H ₆ N ₂	700,040,100	
7.5675	Pentobarbital	C ₁₁ H ₁₈ N ₂ O ₃	555,198,321	
8.3190	4-Dimethylamino-2(5H)-furanone	C ₆ H ₉ NO ₂	6,646,490,736	
14.2652	2-Phenyl-3-methyl-pyrrolo(2,3-b) pyrazine	C ₁₃ H ₁₁ N ₃	428,223,218	
14.2932	Naphyrone	C ₁₉ H ₂₃ NO	2,741,401,772	
16.5307	Benzalphthalide	C ₁₅ H ₁₀ O ₂	367,395,119	

2.2. Preparing the plant extract

The chopped AV leaves were boiled in 100 ml of sterilised distilled water for 15 min and allowed to cool overnight. The AV extract was filtered then stored at 4°C.

2.3. Preparing the cell-free culture supernatant (CFCS)

S. aureus was cultivated on MHA plates and incubated for 24 h at 37°C then subcultured in MHB medium under the same previous incubation conditions. The culture was then centrifuged for 10 to 15 min at 8000 rpm, after which the supernatant was harvested and the pellet

discarded. CS-*S. aureus* was preserved at 4°C.

2.4. Chemical composition assessment using gas chromatography-mass spectrometry (GC/MS)

2.4.1. GC-MS analysis of AV extract

To analyse the AV extract, the leaf broth was air dried then the residue was dissolved in ethanol and then subjected to GC-MS analysis (Agilent Technologies 7890B Gas Detector, Santa Clara, CA, USA). These settings were used for a full scan of the sample (Research Center, Princess Noura University, Riyadh, Saudi Arabia).

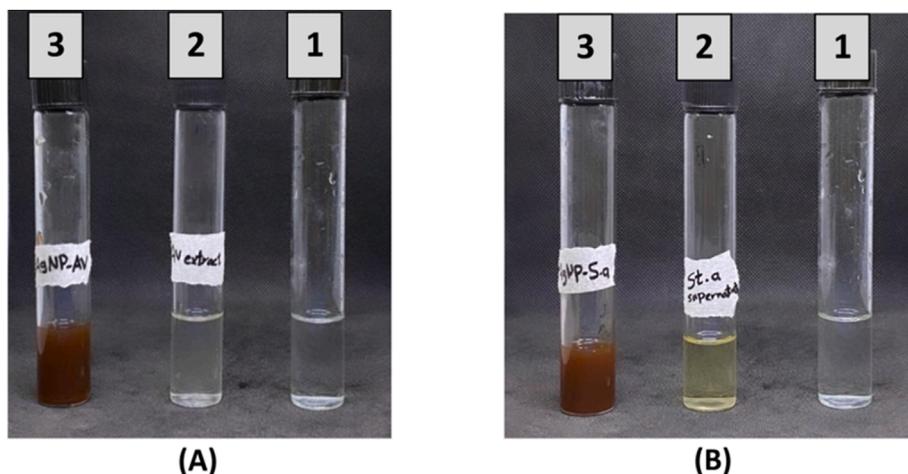


Fig. 1. Production of AgNPs-AV (A3) and AgNPs-CS of *S. aureus* (B3). A1 and A2 were AgNO₃ and Av extract, while B1 and B2 were AgNO₃ and CS of *S. aureus*.

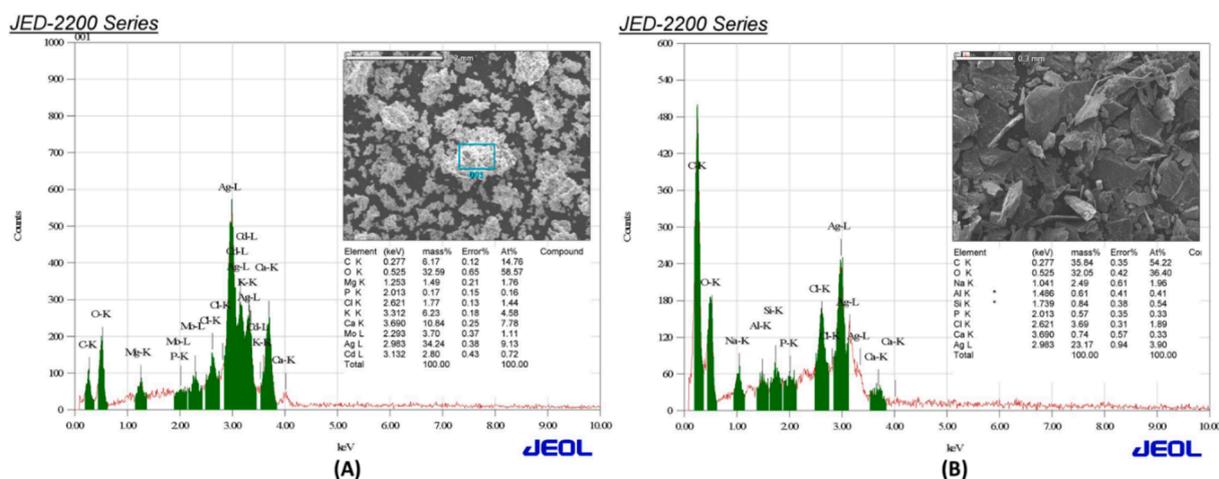


Fig. 2. EDX images of synthesised AvAgNPs (A) and CS of *S. aureus*-AgNPs (B).

2.4.2. GC –MS analysis of CS-*S. Aureus*

To analyse the CS-*S. aureus* by GC–MS, the recovered supernatant was evaporated then the dried residue was dissolved in methanol, before being subjected to GC–MS. The GC–MS analysis conditions were the same as those used for the Av extract.

2.5. Biosynthesis of silver nanoparticles

2.5.1. Using AV extract

Mixture of AV extract and AgNO₃ solutions were placed in the sunlight and mixed continuously until a colour change was observed; it was then covered to prevent any oxidation.

2.5.2. Using CFCS of *S. Aureus*

20 ml of AgNO₃ (4 mM of AgNO₃ in 50 ml of sterilised distilled water) was mixed with 10 ml of CFCS of *S.aureus* (pH:6.7) and left in sunlight and mixed continuously until a colour change observed

2.6. Characterisation of silver nanoparticles

The AgNPs biosynthesised with Av extract and CS- *S. aureus* were characterised.

2.6.1. Uv–visible spectrophotometer

The UV–visible spectrophotometer (UV-1800, Shimadzu) was used

to measure the absorption peaks of the AV extract and CS-*S. aureus* containing reduced silver ions. The spectra were recorded at (200–900 nm).

2.6.2. Fourier-transform infrared spectroscopy (FT-IR)

The functional groups in charge of stabilising and capping AgNPs are shown by FT-IR spectrum analyses. With a resolution of 4 cm⁻¹, the spectra were captured using FT-IR (IR Prestige-21 – Shimadzu) within 400–4000 cm⁻¹ range.

2.6.3. Energy dispersive X-ray (EDX) analysis

The elemental investigation of the sample was carried out using EDX. It was recorded using a scanning electron microscope coupled with EDX (JSM-6380 LA).

2.6.4. Scanning electron microscope (SEM)

The shape and size of the AgNPs were observed using SEM. SEM was also used to examine the damage to the bacterial cell caused by the synthesised AgNPs. The images were captured using SEM (JSM-6380 LA) and field emission scanning electron microscope (FESEM-JSM 7610F-Japan) at different magnifications for detailed analysis. (Electron Microscope Unit, Central laboratory, Science College, King Saud University – Central Research Laboratory, Female Campus, King Saud University).

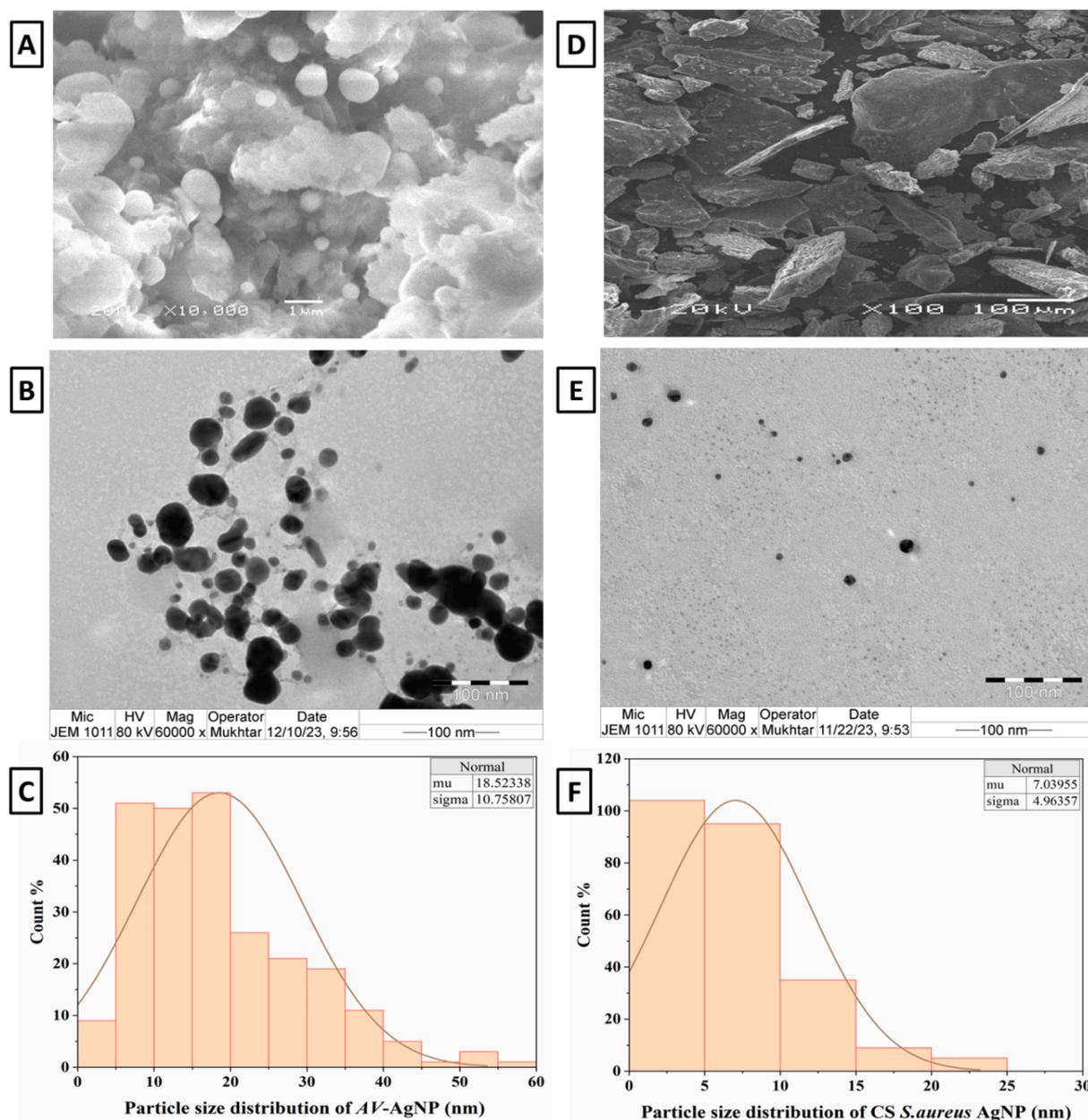


Fig. 3. SEM and TEM images of synthesised AV-AgNPs (A and B) and CS of *S. aureus*-AgNPs (D and E), histograms of the particle size distribution (C and F).

2.6.5. Transmission electron microscope (TEM)

TEM (JEOL-JEM-1011) was used to characterise the internal morphology and crystallographic data. (Electron Microscope Unit, Central laboratory, Science College, King Saud University).

2.7. Minimum inhibitory and bactericidal concentration (MIC – MBC) assessment

The Clinical and Laboratory Standards Institute's standard broth dilution method (CLSI M07-A11) used to assess the antibacterial potential of the silver nanoparticles (Weinstein et al., 2018). Using an adjusted bacterial concentration (0.5 McFarland's standard) with AgNPs serial two-fold dilution at concentrations ranging from 500 $\mu\text{g}/\text{ml}$ to 3.906 $\mu\text{g}/\text{ml}$, the MIC was performed in MHB.

2.8. Determining antibacterial efficacy of nanoparticles using well-diffusion assay

The antibacterial activity testing was carried out by using an agar well-diffusion assay according to CLSI M02-A12 (Patel et al., 2015).

2.9. Statistical analysis

The data presented in the figures and tables in this study are values from triplicate experiments (\pm SD). To identify significant differences, One-Way-Analysis of variance (ANOVA) ($p \leq 0.05$) was conducted using Origin 2024.

3. Results and Discussion

This study compared the antibacterial effectiveness of AgNPs manufactured using AV extract and CS-*S. aureus* against drug-resistant bacteria. The initial substance and the resultant mixtures were identified

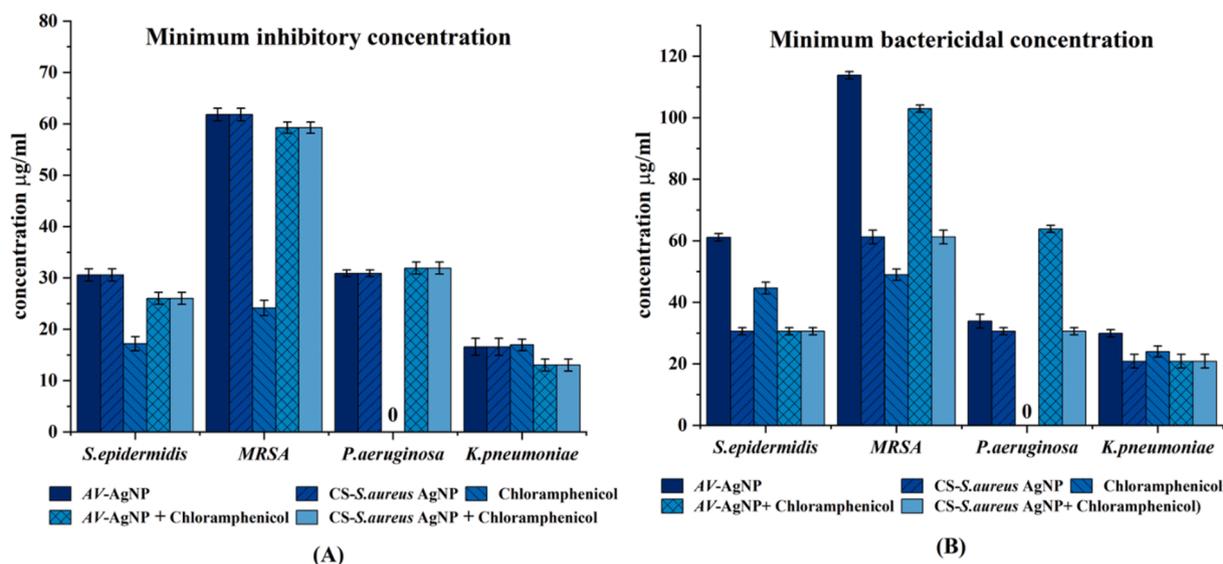


Fig. 4. (A) MIC and (B) MBC values of synthesised AV-AgNPs, CS of *S. aureus*-AgNPs and chloramphenicol (in µg/ml). The values expressed are the mean of triplicate experiments (\pm SD) and carried out using one-way ANOVA ($p \leq 0.05$).

through the use of GC–MS, UV–vis spectrophotometry, FT-IR, EDX, SEM and TEM. Also, FE-SEM was used to visualise bacterial cell damage. To establish antibacterial susceptibility, the MIC, MBC and well agar diffusion techniques were employed.

3.1. Chemical composition assessment using GC–MS

3.1.1. GC –MS analysis of AV extract

GC–MS analysis identified twenty- three active phytochemical compounds in the AV extract Table 1 lists several of the active compounds discovered. GC–MS analysis showed a variety of organic acids. Some of these organic acids such as oleic acid, 16-octadecenoic acid, phthalic acid and benzoic acid, which are known to possess antioxidant, anti-inflammatory and antimicrobial capabilities (Ngurah et al., 2020; Zahara et al., 2022).

3.1.2. CS-S. Aureus GC –MS

Few studies investigated the components of CS-*S. aureus* and none of them match our results presented in Table 2. Twenty- five chemical compounds of CS-*S. aureus* were determined by GC–MS analysis. Table 2 displays a selection of the chemical compounds that were detected. Most of them, including 1H-indazole, 2,3-dihydro-1H-pyrrolo[1,2-a]benzimidazole, quinazoline and 4-dimethylamino-2(5H)-furanone, possess a significant antimicrobial, antitumor and anti-inflammatory activity (Alagarsamy et al., 2018; Siwach and Verma, 2021; Sulaiman et al., 2023).

3.2. Biosynthesis of silver nanoparticles

The AV extract and CS-*S. aureus* were used for AgNP synthesis since they possess a broad range of metabolites that may help in the reduction of AgNPs. The colour change of reaction mixtures from colourless or pale yellow to dark brown (Fig. 1) was indicated by visual observation. Fig. 2..

3.3. Silver nanoparticles characterisation

3.3.1. Uv–vis spectrophotometer examination

Figure S1: UV–vis spectrophotometry results of the synthesised AV-AgNPs at 445 nm (A) and CS- *S. aureus*-AgNPs at 444.5 nm (B) at different time intervals (0 h, 24 h, 48 h and 72 h) at room temperature.

3.3.2. FT-IR results

The study used FT-IR to identify functional groups involved in the bio-reduction of silver nanoparticles (AgNPs) by AV extract and their roles in stabilization. The spectrum of AV extract revealed peaks indicating the presence of alcohols, alkanes, amines, and phenols, confirming its capability as a reducing agent for AgNPs synthesis. After AgNPs formation, shifts in the peaks suggested the formation of a capping layer to prevent agglomeration. Additionally, FT-IR analysis of CS-*S. aureus* showed the presence of various biomolecules, which are essential for stabilizing AgNPs and reducing metal salts (Figure S2).

3.3.3. EDX findings

The EDX spectrum of AvAgNPs and CS-*S. aureus*-AgNPs shows a strong peak at 3 keV, indicating the presence of silver. AvAgNPs contain magnesium, phosphorus, chlorine, potassium, calcium, molybdenum, and cadmium, while CS-*S. aureus*-AgNPs include aluminum, sodium, chlorine, silicon, calcium, and phosphorus. These elements are believed to originate from the AV extract and CS-*S. aureus* used in the synthesis. Previous studies by Ansar et al. (2020) and Desai et al. (2020) also noted that these non-silver elements likely served as capping agents in biogenic synthesis.

3.3.4. SEM and TEM results

SEM and TEM were performed to examine the size and the nano-structure of the AgNPs. Images of AvAgNPs appear as spherical in shape and ranging in size from 3.1 to 59.2 nm (Fig. 3A and B). Meanwhile, CS of *S. aureus*-AgNPs was in the range of 1.2–23.4 nm and was spherical to irregular (Fig. 3D and E). Therefore, CS of *S. aureus*-AgNPs has a special size range compared with AvAgNPs. The histograms (Fig. 3C and F) show the average of 200 particles calculated using (Image J software incorporated with Origin software). It presents the average size of AV-AgNPs to be 18.5 nm, while the average size of CS of *S. aureus*-AgNPs was 7.03 nm.

3.4. MIC and MBC assessment

Since the MIC is the lowest amount of antibacterial substance that inhibits microbial growth, the tube's visual turbidity used as a marker for the MIC. It was observed that both types of AgNPs exhibited the same MIC activity. The highest sensitivity was for *K. pneumoniae* (15.625 µg/ml) followed by *P. aeruginosa* and *S. epidermidis*. MRSA showed the lowest sensitivity at 62.5 µg/ml (Fig. 4A). The MIC test was also performed

Klebsiella pneumoniae

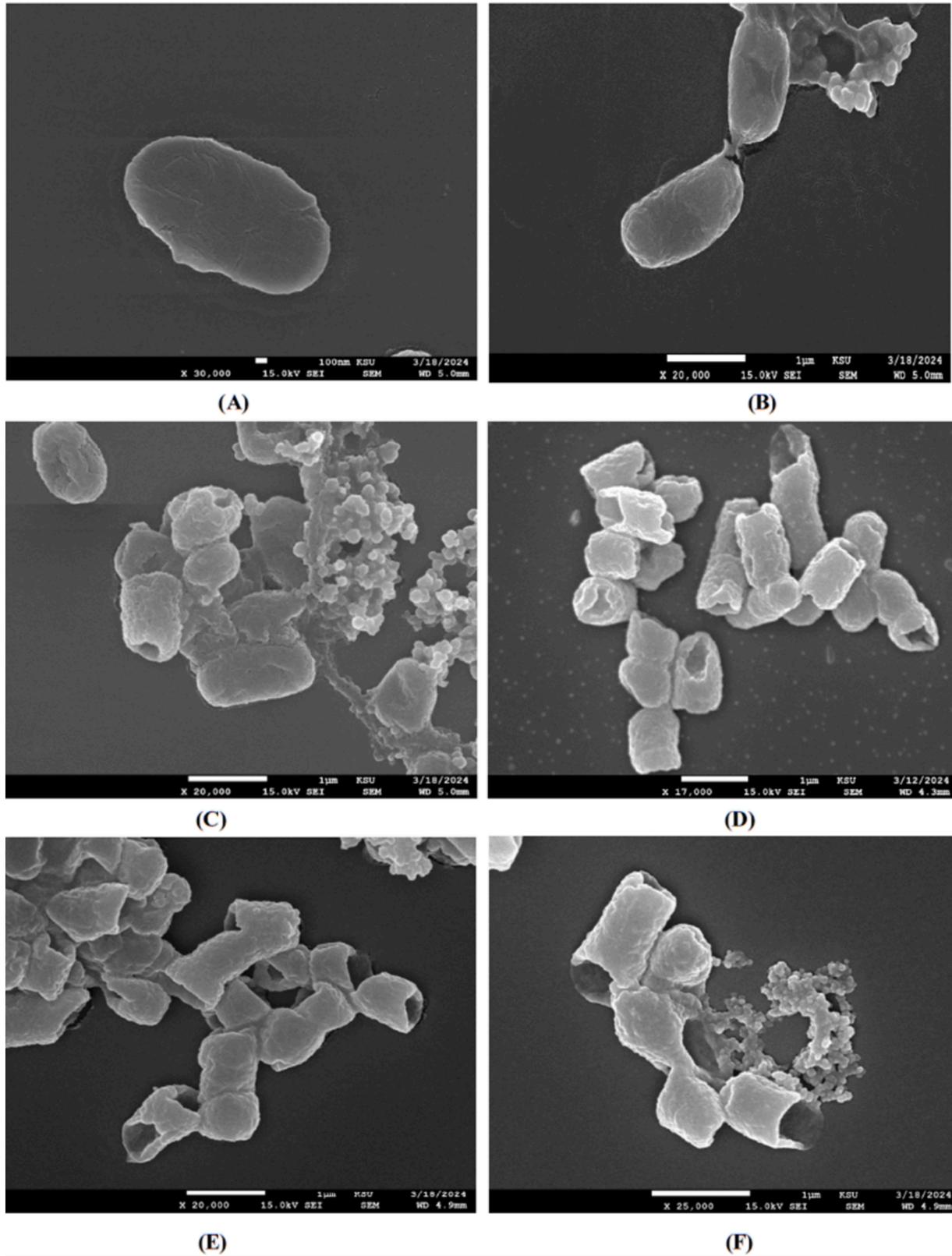


Fig. 5. FE-SEM images of the resulting effect upon *K. pneumoniae* of untreated (A and B), synthesised AvAgNPs (C), CS of *S. aureus*-AgNPs (D), chloramphenicol (E), chloramphenicol with CS of *S. aureus*-AgNPs (F).

MRSA

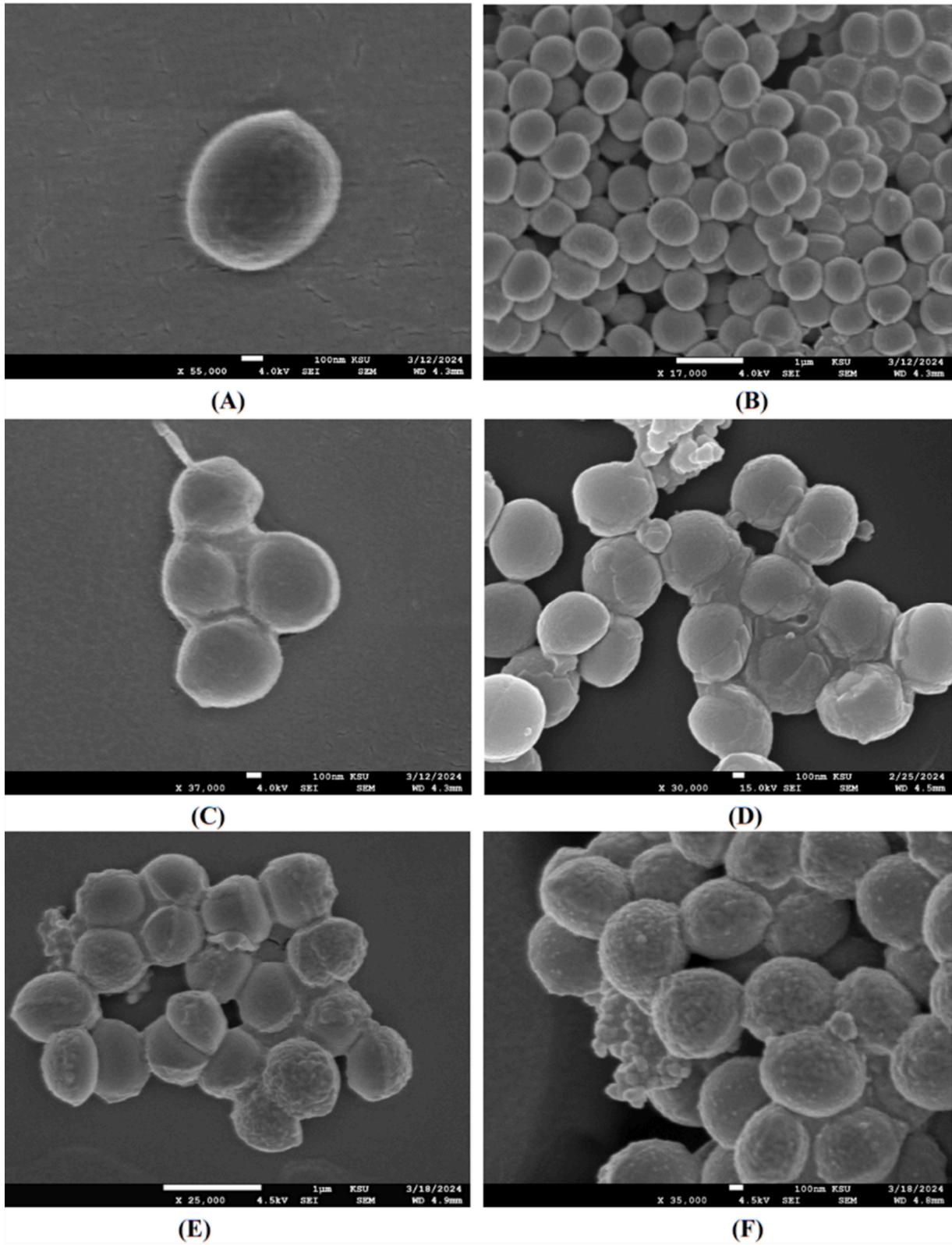


Fig. 6. FE-SEM images of the resulting effect upon MRSA of untreated (A and B), synthesised AvAgNPs (C), CS of *S. aureus*-AgNPs (D), chloramphenicol (E), chloramphenicol with CS of *S. aureus*-AgNPs (F).

Table 3

The DIZs of AgNPs against tested bacteria using the well-diffusion method. The values present in the table are the mean of triplicates (\pm SD) and run through one-way ANOVA to determine ($p \leq 0.05$).

Tested bacteria	Zone of inhibition (mm)				
	AV-AgNP (well 1)	CS- <i>S.aureus</i> -AgNP (well 2)	Chloramphenicol (50 μ g) (well 3)	AV-AgNP + Chloramphenicol (well 4)	CS- <i>S.aureus</i> -AgNP + chloramphenicol (well 5)
Methicillin-resistant <i>Staphylococcus aureus</i>	14.3 \pm 0.6	18.6 \pm 0.6	19.5 \pm 0.5	15.1 \pm 0.2	18.2 \pm 0.5
<i>Staphylococcus epidermidis</i>	18.3 \pm 0.6	22.3 \pm 0.6	19.8 \pm 0.3	18.1 \pm 0.4	21.8 \pm 0.3
<i>Klebsiella pneumoniae</i>	13.2 \pm 0.3	15 \pm 0.2	22.1 \pm 0.3	20.3 \pm 0.3	21.2 \pm 0.3
<i>Pseudomonas aeruginosa</i>	13.1 \pm 0.4	15.1 \pm 0.2	0	13.1 \pm 0.1	15 \pm 0.5

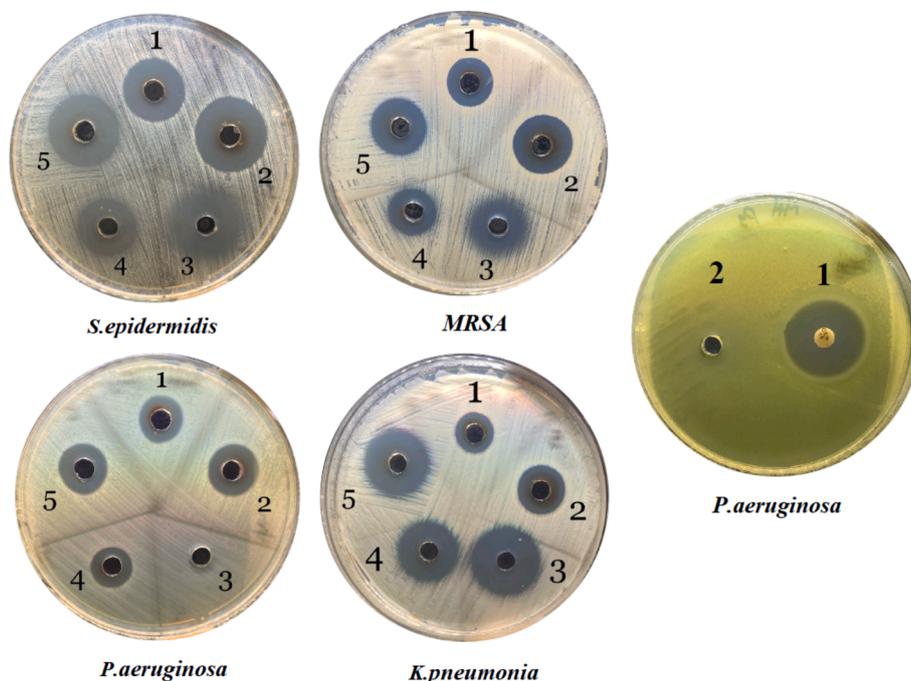


Fig. 7. The antibacterial effect upon different bacterial strains treated with AV-AgNP (1), CS of *S. aureus*-AgNP (2), chloramphenicol (3), AV-AgNP + chloramphenicol (4), CS of *S. aureus*-AgNP + chloramphenicol (5).

using chloramphenicol (50 μ g/ml). The findings indicated that *P. aeruginosa* was resistant, while MRSA had MIC of 25 μ g/ml. *K. pneumoniae* and *S. epidermidis* were highly sensitive, each with a MIC of 12.5 μ g/ml. In addition, chloramphenicol was tested synergistically with both types of AgNPs; the results are shown in Fig. 4A.

Fig. 4B shows the MBC values for the tested bacteria; however, the interesting point is that the MBC values for AgNPs vary. The MBC for *K. pneumoniae* was 15.625 g/ml for CS-*S. aureus*-AgNPs, while for AV-AgNPs, it was 31.25 g/ml. This was followed by *P. aeruginosa*, *S. epidermidis* and lastly MRSA = 62.5 g/ml, 125 g/ml. Farouk et al. (2020) has reported that *S. aureus* was the most resistant with low MIC, MBC to AgNPs than Gram-negative *S. typhi*.

3.5. Cell damage screening using FE-SEM

We used FE-SEM to assess the surface morphological changes caused by the AgNPs and chloramphenicol on *K. pneumoniae*; and the changes were compared to untreated bacteria. The tested samples were taken at their MIC concentrations and *K. pneumoniae* considered as the most sensitive due to MIC results. Fig. 5C and D show the damage to the cell after exposure to AV-AgNPs and CS-*S. aureus*-AgNPs. Treatment led to cellular changes, with clear signs of deformation, shrinkage and fracture. In addition, the walls of *K. pneumoniae* ruptured, causing intracellular components to leak out, which could be observed surrounding

the cell, leading to cell death, especially when treated with CS-*S. aureus*-AgNPs (Fig. 5D). These NPs have a greater bactericidal effect than AV-AgNPs, in which some intact cells were observed (Fig. 5C). Further investigation was conducted to determine the MIC of chloramphenicol + CS-*S. aureus*-AgNPs; a synergic effect was observed. Fig. 5F reveals the *K. pneumoniae* cells were severely damaged. Whereas Fig. 5A and B presents the untreated cells as having intact cell wall and being within a normal size range.

According to the MIC results, MRSA was the most resistant bacteria. Fig. 6A and B show the untreated cells having a smooth surface and intact cell wall. In contrast, Fig. 6C and D show a misshapen cell wall after being treated with AV-AgNPs and CS-*S. aureus*-AgNPs. A greater effect was observed in with CS-*S. aureus*-AgNPs, where there was clear surface damage and minor internal leakage. Additionally, chloramphenicol was scanned both separately and with CS-*S. aureus*-AgNPs (Fig. 6E and F).

3.6. Assessment of the antibacterial activity of the synthesised silver nanoparticles using a well-diffusion assay

Our results showed that AV-AgNPs and CS-*S. aureus*-AgNPs have different antibacterial activity against all the tested bacteria (Fig. 7). The presence of an inhibition zone allowed for the quantitative evaluation of antibacterial activity. MRSA showed a large diameter of inhibition zone (DIZ) (18.6 \pm 0.6 mm) for CS-*S. aureus*-AgNP and (14.3 \pm 0.6 mm) for

AV-AgNP. However, *S. epidermidis* had the largest DIZ (22.3 ± 0.6 mm) and (18.3 ± 0.6 mm) (Table 3). According to Al Mutairi et al. (2022), who manufactured AgNPs and evaluated them against some drug-resistant bacteria found that it had an impressive antibacterial capacity specifically against *S. aureus*.

Moreover, AgNPs were evaluated in combination with chloramphenicol, it emerged that the combination was not potent against all bacteria, only *K. pneumoniae*. However, when chloramphenicol was tested individually, it exhibited moderate antibacterial activity against all bacteria, except *P. aeruginosa*, in which it was found to be resistant. To confirm its resistance, a gentamicin disc was tested only on *P. aeruginosa* (Fig. 7), which demonstrated that it was susceptible to it. Other investigations reported by Lorusso et al. (2022) have demonstrated that *P. aeruginosa* has a remarkably low susceptibility to antibiotics.

Some factors affecting the antibacterial activity of AgNPs such as size, shape, surface modification and the techniques applied for synthesis (Nie et al., 2023).

Yousaf et al. (2020) revealed that the size of NPs has a significant impact upon their antibacterial efficacy. Smaller size particles have shown stronger antibacterial activity, reflecting that they have greater capability to enter into bacteria. The smaller size NPs have a larger surface area than bigger NPs, resulting in higher antibacterial activity. That study agrees with our results, which found that the nanoparticle size generated by CS – *S. aureus* was smaller than the size produced by AV and the former exhibited greater antibacterial activity against bacteria in our tests.

4. Conclusion

Our study successfully produced silver nanoparticle using AV extract and CS of *S. aureus*, which are known as a green, eco-friendly and biological substances. The synthesised AgNPs showed a noticeable effect as an antibacterial agent against both Gram-positive and Gram-negative bacteria. Further investigations will assure that AgNPs of AV extracts and CS of *S. aureus* are applicable as treatments for bacterial infections.

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CRedit authorship contribution statement

Asma Ahmed Al-mehdhar: . Khaloud Mohammed Alarjani: Writing – review & editing, Writing – original draft, Supervision, Methodology, Formal analysis, Data curation, Conceptualization. Noura Salem Aldosari: . Mai Ahmed Alghamdi: Supervision, Writing – review & editing, Writing – original draft.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.jksus.2024.103464>.

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