



Molecular docking and molecular dynamics simulation of anticancer active ligand '3,5,7,3',5'-pentahydroxy-flavanonol-3-O- α -L-rhamnopyranoside' from *Bauhinia strychnifolia* Craib to the cyclin-dependent protein kinase

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ARTICLE INFO

Article history:

Received 29 January 2019

Revised 28 April 2019

Accepted 15 May 2019

Available online 16 May 2019

Keywords:

Molecular docking

Molecular docking simulation

3,5,7,3',5'-pentahydroxy-flavanonol-3-O- α -L-rhamnopyranoside

Bauhinia strychnifolia Craib

Fabaceae

Fabaceae

Cyclin-dependent protein kinase 2

ABSTRACT

The compound '3,5,7,3',5'-pentahydroxy-flavanonol-3-O- α -L-rhamnopyranoside' reported from *Bauhinia strychnifolia* Craib (family Fabaceae) possess ten times more cytotoxicity against certain cancer cell line than the anti-cancer drugs, but nontoxic to normal cells. Its stability with the protein 'cyclin-dependent protein kinase 2/CDK-2' (-which plays vital role in apoptosis, regulation of the cell cycle, transcription and in the neuronal functions) were performed via molecular docking and molecular dynamics simulation simulations. The molecular dynamics simulation suggests that the protein-ligand complex is stable at least in the time period of 40 ns. This study also shows that after 20 ns of simulation, there is a shift in the ligand position, and it might be due to changes in the binding type. According to the data obtained via MM-PBSA calculations, this shift leads to even stronger binding of ligand. Such binding flexibility can be explained by highly hydrophilic nature of both ligand and binding site.

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1. Introduction

Despite remarkable decline in death rates owing to cancer during last two to three decades, it still remains world's major health issue, and the leading causes of death worldwide too (Siegel et al., 2016). Apart from the strategies such as targeted cancer therapy (Padma, 2015), immunotherapy (Upadhyaya, 2006), hyperthermia (Wust et al., 2002), resistance modulation (Fracasso et al., 2000), antidotes/toxicity modifiers (Hasinoff et al., 1999) and alternative formulations (Krishna and Mayer, 1997), the surgery, chemotherapy and the radiotherapy are still the main tones in the treatment of cancer, but it brings severe side effects too (DeSantis et al., 2014). Recently Several novel anticancer candidates such as natural product (Zengin et al., 2018; Trampetti et al., 2019; Uysal et al., 2019), organic synthetic compound (Mollica et al., 2011, 2012;

Schreiber, 2011; Nicolaou, 2014; Picot et al., 2017) and nanoparticles (Brigger et al., 2002) have been identified recently; however, still the natural products used as an effective drug candidate for inhibiting cancer (Thenmozhi et al., 2018), and are continuously being explored through plant bioprospecting because of its safer nature compared to the synthetic drug molecules (Ali et al., 2014; Newman and Cragg, 2016).

Bauhinia strychnifolia Craib (family Fabaceae) traditionally used for the treatment of the cancer (Wutthithammavet, 1997; Yuenyongsawad et al., 2013). The compound '3,5,7,3',5'-pentahydroxy-flavanonol-3-O- α -L-rhamnopyranoside' isolated from the ethanolic extract of root and stem of *B. strychnifolia* have been reported ten times more cytotoxic (IC₅₀: 0.00054 μ g/mL) against KB cell line than camptothecin (-the anti-cancer drug, IC₅₀: 0.0057 μ g/mL), while nontoxic to normal cells (Yuenyongsawad et al., 2013). The recent advances in molecular docking and molecular dynamics simulation have provided insights of the compounds including mechanism of action, drug-like properties, selectivity, improved potency and side effects (Kitchen et al., 2004). The protein CDK-2 exerts catalytic functions when bound with the cyclins, and plays vital role in apoptosis, regulation of the cell cycle, transcription and in the neuronal functions (Dai and Grant, 2003; Huwe et al., 2003). Hence, in the present molecular docking and

Peer review under responsibility of King Saud University.



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<https://doi.org/10.1016/j.jksus.2019.05.004>

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molecular dynamics simulation simulations aimed to evaluate the binding of the ligand with the core-cell cycle machinery protein.

2. Materials and methods

The 2D structure of ligand '3,5,7,3',5'-pentahydroxy-flavanonol-3-O- α -L-rhamnopyranoside' reported to have anticancer activity (Yuenyongsawad et al., 2013) was modeled, converted to 3D structure (using ChemSketch v. 12.01), optimized using MMFF force field (parameters: 500 steps of steepest descent algorithm, convergence criterion of $10e^{-7}$) (Halgren, 1996), and used for docking simulation (Kumari et al., 2014) with receptor CDK-2 (PDB ID: 1D18) retrieved from Protein Data Bank. During the preparation of receptors, all heteroatoms (i.e. water, ions, etc.) were removed, polar hydrogens were added, and Kollman charges were assigned (Case et al., 2016; da Silva and Vranken, 2012). The ligand was docked using AutoDock4.2 (Trott and Olson, 2010), the trajectories of docked complexes were analyzed through 20 ns of MD simulations using GROMACS 4.6.5 (Hess et al., 2008) following the previously mentioned procedure (Phosrithong and Ungwitayatorn, 2010; Sudhamalla et al., 2010; Gurung et al., 2016; Das et al., 2018a,b).

3. Results and discussion

The molecular docking of the known ligand (DTQ – 4-[3-hydroxyanilino]-6,7-dimethoxyquinazoline) was carried out into CDK-2 complex structure. The RMSD for the position of the atoms was selected as a quantitative measure of quality. In this case, it was about 1.5 Å. The procedure was performed using several molecular docking software approaches for CDK-2. The program resulted in lowest RMSD in comparison to X-ray structure was selected (Table 1). The Fig. 1A shows superposition of docked ligand with ligand in X-ray structure.

Table 1

Software selection table. For each biotarget, the docking study using different docking programs were performed. In each case of the selected software demonstrated the best fit (lowest RMSD) between docked ligand and the same ligand in X-ray structure.

S. No.	Docking software	CDK-2 (PDB code 1D18)
1.	Autodock Vina	1.33
2.	AutoDockFR	3.49
3.	PharmDock	3.75

3.1. Binding site

As the typical protein kinase CDK-2 formed by two flexible fragments: N and C-lobes, between them the most studied site-ATP-competitive binding site is placed. This site has a high degree of conservation among the cyclin-dependent protein kinases and in general it tends to bind to polar heteroaromatic compounds (Sims et al., 2003). Various inhibitors normally occupy three different sub-sites (the adenine, ribose and triphosphate subsites) or three specific binding modes for ATP competitive inhibitors (Li et al., 2015). The first one is observed for compounds with a heterocyclic purine moiety. Such compounds inhibit CDK-2 by occupation preferably adenine sub-site and in general replicate the location of ATP in the binding site. H-bonds formation with the backbones of residues Glu81 and Leu83 is crucial for this binding mode. Another type of binding observed in 4-anilinoquinazoline inhibitor class. Such compounds form multiple bridged hydrogen bonds with water, also characterized by pi-pi staking with PHE80. Inhibitors with flavonoid structure provide the third way of binding. Two of the conserved H-bonds with Leu83 and Glu81 can be observed in this binding mode, additionally residues Lys33 and Asp145 which are located in the ribose and triphosphate regions of ATP-competitive binding site

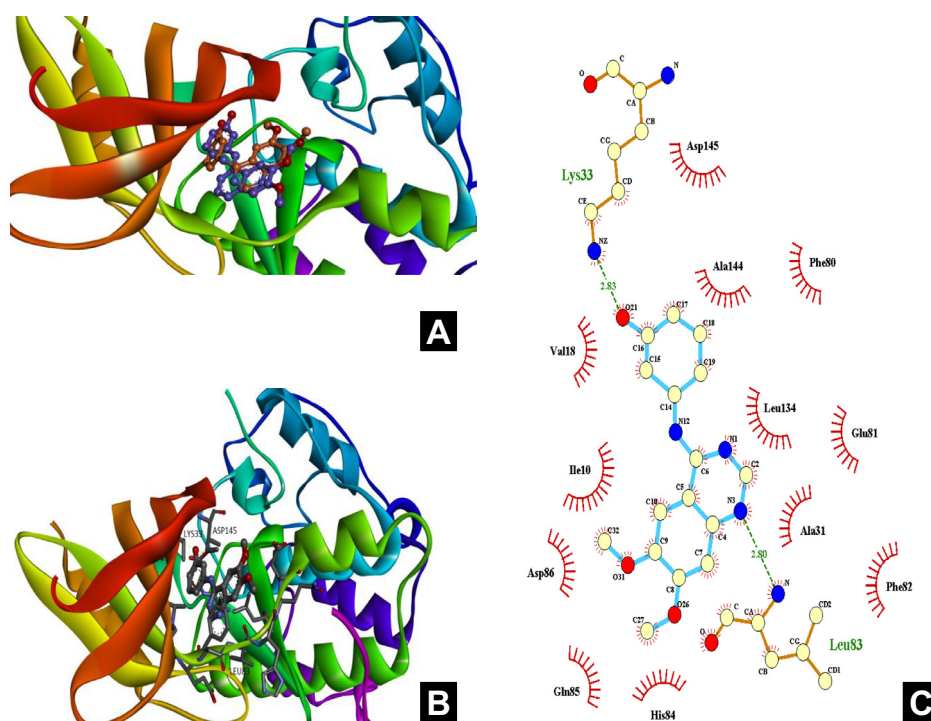


Fig. 1. A. Superposition of ligand in the protein-ligand complex, obtained by docking and using X-ray data. Orange – ligand from the pdb file, purple – position of this ligand calculated by docking. Fig. 1B-C. General view and position of active site.

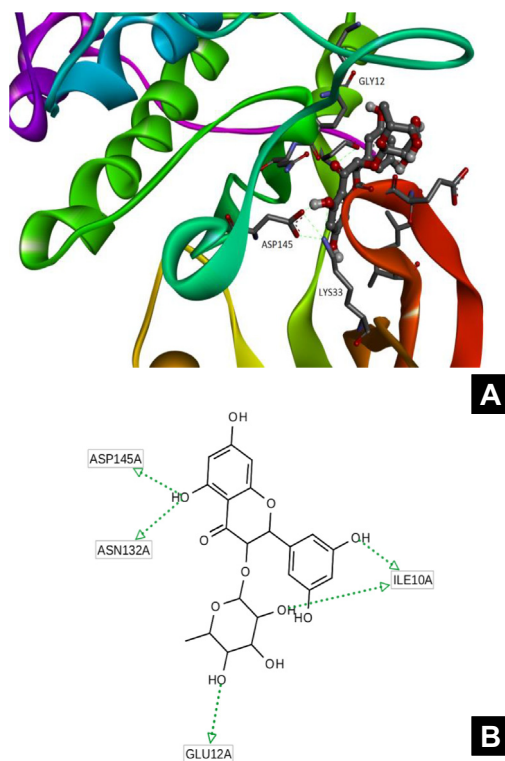


Fig. 2. A-B. Protein-ligand complex with target compound in the binding site. A (left) – general view of docking pose. B (right) – h-bond map. Different colors of the protein show different secondary structure elements. Residues of the protein are highlighted by blue color. In the ligand the red atoms represent oxygen, white atoms – hydrogen, yellow – sulfur, dark blue – nitrogen and grey – carbon.

respectively also take part in H-bond formation with the ligand. Ligand, which position is shown in the complex with CDK-2, binds in the way that is typical for the flavonoids. The Fig. 1B-C shows key interactions of this ligand with amino acids of the binding site.

3.2. Binding interaction

The 3D model of interaction of the most favored docking pose in CDK-2 (Fig. 2A) show typical flavonoid type of binding interaction of phenolic hydroxyl groups with the side chains of Asp145 and Lys33. In addition, there was a formation of hydrogen bonds between the hydroxyl group of the carbohydrate moiety and Glu12. There might be possibilities for hydrogen bonds between ligand hydroxyl groups and water molecules which occupy the binding site too. The pharmacophore model was constructed on the basis of docking data (coordinates of the ligand). This model taken into account for the key interaction established in the result of analysis of the ligand position into the binding site. The H-bonds model is shown in the Fig. 2B.

3.3. Possible modifications of inhibitor's structure

The ligand showed the ability to form a large number of hydrogen bonds with the amino acids of binding site and bridge hydrogen bonds with water molecules. This is reflected in the high values of docking scoring functions, which prompts to assume that the affinity of the ligand to the ATP-competitive binding site of CDK-2 will be significantly higher. However, when it comes to protein kinases with high degree of conservation of the binding sites, and compound of interest belongs to the class of flavonoids which are well known by their ability to inhibit different kinases, the issue of selectivity comes into picture.

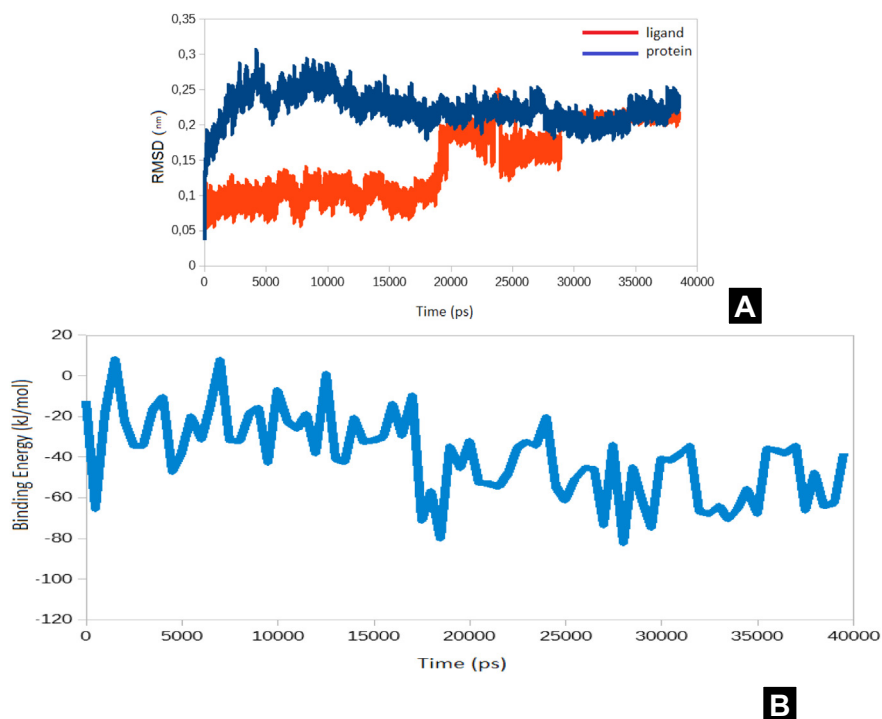


Fig. 3. A. The RMSD values of the protein-ligand complexes over the simulation time (X-axis represents time scale in nanosecond and Y-axis represent the RMSD in nanometers). B. Average binding energy. One can see the significant change of the average values of binding energy (from -28.7 kJ/mol to -50.9 kJ/mol) in about the same time when the shift of RMSD values has been occurred. Increased binding energy shows that the new type of binding can be more energetically favorable. Following is the more detailed analysis of a new type of binding.

3.4. Analysis of the trajectory and MM-PBSA analysis

The Fig. 3A shows the RMSD values of the protein-ligand complex. The substantial RMSD value can be explained by the mobility of the loops. At the same time, the visual analysis of the trajectory confirms the stability of the secondary structure of the protein. There was a significant shift in RMSD values for ligand after 20 ns of simulation. The RMSD rises from an average of 0.10 nm with SD of 0.01 nm to 0.19 nm with SD of 0.02 nm. This shift may indicate a new binding mode for ligand. The MM-PBSA calculations were performed to explore the influence of this event on binding energy. The Fig. 3B shows binding energy – time dependence during simulation. The significant change of the average values of binding energy (from -28.7 kJ/ mol to -50.9 kJ/ mol) in about the same time when the shift of RMSD values has been occurred. Increased binding energy shows that the new type of binding can be more energetically favorable.

3.5. Analysis of the conformational changes

Several representative frames from the part of trajectory after 30 ns has been studied for the analysis of a new type of binding. One can observe the change of the position of the carbohydrate moiety as well as changes in the conformation of the benzene ring. This position of the ligand enables to form an additional hydrogen bonds with the Glu12. The detailed picture of this interactions have been shown in the Fig. 4A-B.

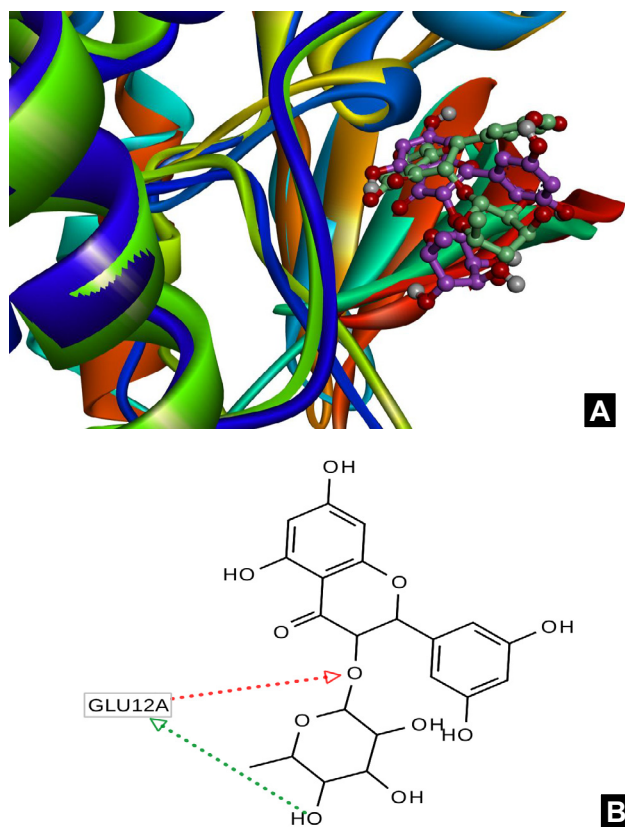


Fig. 4. A-B. A. Superposition of protein-ligand complex (green color for ligand) with target compound in the binding site obtained by docking and extracted from molecular dynamics trajectory (magenta color for ligand). B. h-bond map. Different colors of the protein show different secondary structure elements.

4. Conclusions

The binding study shows that the flavonoid type of binding is the most probable binding mode for the ligand. The molecular dynamics simulation suggests that protein-ligand complex is stable at least in the time period of 40 ns. This study also show that after 20 ns of simulation, there is a shift in ligand position, and it might be due to binding type change. According to the data obtained via MM-PBSA calculations, this shift leads to even stronger binding of ligand. Such binding flexibility can be explained by highly hydrophilic nature of both ligand and binding site. Despite that molecular dynamics simulation studies show good chances for actual binding, it is important to note that the issue of selectivity is very actual for flavonoids and as a consequence for ligand.

Acknowledgments

Research supported by the King Saud University, Deanship of Scientific Research, College of Science, Research Center.

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