



Original Article

Comparative evaluation of phytochemical screening, in vitro antioxidant & α-Glucosidase inhibitory properties of *Ceiba pentandra* & *Basella rubra* leaf extracts: Identification of active principles by Q-TOFLCMS, ADMET prediction & molecular docking approach

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ABSTRACT

Ceiba pentandra and *Basella rubra* leaves are traditionally used in Indonesia to treat ailments like diabetes. This study aimed to validate their use by assessing their antioxidant and α-glucosidase inhibitory properties. Initially, maceration of the leaves of both plants yielded dichloromethane, methanol (MeOH), and aqueous extracts. These were phytochemically profiled, and the most active extracts underwent quadrupole time of flight- liquid chromatography mass spectrometry (Q-TOF LCMS) analysis to identify potentially active principles, followed by molecular docking to ascertain their mechanism of action. The results revealed that the aqueous and methanolic extracts of *C. pentandra* exhibited potent antioxidant activities, with IC₅₀ values of 17.66 ± 0.7 and 53.58 ± 0.25 µg/mg ascorbic acid equivalent for ferric reducing antioxidant power, respectively. In comparison, the aqueous and methanolic extracts of *B. rubra* showed lower antioxidant potential, with values of 8.73 ± 0.1 and 10.17 ± 0.08 µg/mg AAE, respectively. Additionally, the DPPH assay displayed that the aqueous and methanolic extracts of *C. pentandra* had IC₅₀ values of 157.32 ± 3.44 and 27.71 ± 1.54 µg/mL, respectively. In contrast, the aqueous and methanolic extracts of *B. rubra* had higher IC₅₀ values of 661.78 ± 1.8 and 253.76 ± 2.4 µg/mL, respectively. Owing to their significantly higher antioxidant activity, the aqueous and methanolic extracts of *C. pentandra* leaves also displayed better α-glucosidase inhibitory effects compared to *B. rubra*, with IC₅₀ values of 109.54 ± 1.72 and 10.78 ± 0.48 µg/mL, respectively. Q-TOF LCMS analysis of *C. pentandra*'s methanolic extract identified significant bioactive compounds, including *m*-coumaric acid (1), *cis*-β-D-glucosyl-2-hydroxycinnamate (2), luteolin 7-rhamnosyl (1→6)galactoside (3), avenanthramide 2s (4), robinetin 3-rutinoside (5), melanoxetin (6), scutellarein 7-glucoside (7), torosaflavone B 3'-O-β-D-glucopyranoside (8), and 2''-O-α-L-rhamnosyl-6-C-fucosyl-3'-methoxyluteolin (9). Molecular docking analysis showed that compounds 7, 8, and 3 were the most active with protein 3A4A, having affinity energies of -9.7, -10.0, and -10.3 kcal/mol, respectively. These phenolic compounds could be safe α-glucosidase inhibitors for diabetes treatment.

1. Introduction

Cellular damage is a severe condition that increases the presence of unstable molecules known as free radicals in the human body, potentially leading to health problems such as neurodegenerative, type 2 diabetes mellitus (T2DM), cancer, and cardiovascular disorders. This cellular damage triggered by free radicals is commonly referred to as

oxidative stress. The long-term effects of oxidative stress can result in metabolic disorders, including diabetes and its complications. The International Diabetes Federation (IDF) statistics reported that 415 million adults worldwide were affected by diabetes mellitus (DM) in 2015, and this number increased to 537 million adults in 2021 and is projected to reach 784 million by 2045. More than 60% of these patients are from Asian countries. While there is no definitive cure

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for DM, medications can help manage its complications (<https://idf.org/>). Studies have shown that natural antioxidants are commonly used in diabetes treatment. Antioxidants are crucial in global health for managing chronic diseases like diabetes and cardiovascular conditions. They help neutralize free radicals, reduce cellular damage, and improve overall health outcomes. Alpha-glucosidase inhibitors are a class of compounds that help manage diabetes by slowing down the digestion of carbohydrates, thereby reducing postprandial blood glucose levels. Antioxidants can play a role in this process due to their potential inhibitory effects on alpha-glucosidase. Therefore, investigating the antioxidant properties of herbs with alpha-glucosidase inhibitory effects is essential for preventing and treating diabetes in natural medicine (Sharifi-Rad et al., 2020).

A variety of herbs are widely utilized in traditional medicine systems to heal numerous ailments, including T2DM (Saleh et al., 2018). Among these, the two notable traditional medicinal plants from Indonesia are *Ceiba pentandra* (L.) Gaertn. (Family: Malvaceae), commonly known as the silk-cotton tree or kapok, and *Basella alba* Linn Var *rubra* (Family: Basellaceae), commonly known as Malabar spinach or gondola. Both plants are recognized for their therapeutic properties, including their potential to treat diabetes. *C. pentandra* can grow up to 15 meters in height (Chan, 2023). In contrast, *Basella alba* Linn var. *rubra* is an herbaceous annual or biennial climbing plant found in tropical and subtropical areas. It is a succulent, branched, and smooth vine. Both plants have numerous traditional medicinal uses in Southeast Asian countries. The fresh leaves of *C. pentandra* are administered compressed to alleviate dizziness, while the boiled roots are used to treat edema. The gum from the tree helps to relieve stomach upset, and tender shoots are used as a contraceptive. Leaf infusions are also used to treat fever, cough, hoarseness, and sore throats (Satyaprakash et al., 2013). However, *B. rubra* is traditionally used as a detoxification agent and to treat a variety of conditions, including fever, constipation, stomach aches, kidney damage, high blood pressure, diabetes, heart-swelling, vomiting blood (hematemesis), hemorrhoids, rheumatism, intestinal inflammation, wound healing, stomach ulcers, gout, vaginal discharge, liver swelling, and to increase vitality and endurance (Ajiboye et al., 2021; Sharma and Behera, 2022).

C. pentandra bark's extracts have been shown to improved glucose tolerance in normal and streptozotocin-induced diabetic rats (Satyaprakash et al., 2013). They also have α -glucosidase inhibitory properties (Syihabudin et al., 2018). Lukiat (2014) reported that the *B. rubra* leaves ethanolic extract shows antioxidant properties. Moreover, Vita et al. (2019) demonstrated that *B. rubra* leaves ethanolic extract significantly decreased blood glucose levels in white mice, and Thavamani and Subburaj (2017) discovered that the *B. rubra* leaves extract possesses significant α -glucosidase and α -amylase inhibitory effects, confirming its role in the management of diabetes. The significant medicinal effectiveness of both traditional plants requires further pharmacological investigations, specifically concerning their antioxidant and α -glucosidase inhibitory properties to meticulously confirm their potential role in diabetes management. Hence, this research aimed to determine and validate *C. pentandra* and *B. rubra* leaves potential through phytochemical screening and *in vitro* assays exploring their antioxidant and α -glucosidase inhibitory capabilities using non-polar and polar extracts. Notably, no comparative study on the antidiabetic and antioxidant properties of *C. pentandra* and *B. rubra* leaves extracts using these solvents has been reported in the literature. Hence, this is the first scientific study to compare the antidiabetic and antioxidant potentials of the leaves of both plants using different natures of extracts. This was also done by the identification of bioactive compounds as antioxidants and α -glucosidase inhibitors through Q-TOF LCMS analysis and molecular docking approach.

2. Material and methods

2.1 Sample collection, preparation and determination of herbal samples

The fresh leaves of *C. pentandra* (1.5 Kg) and *B. rubra* (850 g) (Fig. 1) were collected from Jakarta, Indonesia, in October 2023. Prof. Agustin Krisna Wardani at the herbarium Bogorinase, located at the Research Center for Biology (BRIN), Indonesia, verified the identity of both plants'

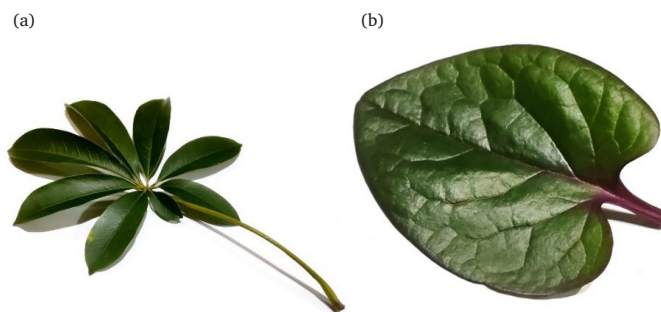


Fig. 1. (a) *Ceiba pentandra*, (b) *Basella rubra*.

leaves (#154427 for *C. pentandra* and #154430 for *B. rubra*). The leaves of both plants were pulverized using a FRITSCH PULVERISETTE 19 Universal Cutting Mill (Germany) to obtain powder material: 300 g of *C. pentandra* and 100 g of *B. rubra*.

2.2 Plant samples extraction

Fifty grams of leaf powder from each plant was separately macerated in 150 mL of DCM for 24 h. This process was carried out at room temperature, followed by filtration using Whatman filter paper 1. Subsequently, the plant material residues were subjected to another round of maceration using 150 mL of methanol. The resultant filtrates were concentrated under reduced pressure by the Buchi rotary evaporator at 45°C. Finally, the dried plant residues were macerated using 150 mL of distilled water for 3 cycles of 24 h. The resulting DCM, MeOH, and aqueous extracts were dried in the laboratory dryer at 50°C. After drying, the percentage yield of all extracts was determined using the following formula:

$$\% \text{Yield} = \frac{\text{Weight of concentrated extract}}{\text{Weight of dry plant powder}} \times 100$$

2.3 Phytochemical screening

All extracts were screened for the presence of bioactive phytoconstituents namely, flavonoids, phenolics, hydrolyzable tannins, alkaloids, anthraquinones, steroids, terpenoids, and saponins. Various chemical reagents were prepared, and specific tests were conducted to identify the phytochemicals (Ahmed et al., 2018; Roheem et al., 2020; Raduan et al., 2022).

2.4 Quantitative analysis

2.4.1 Determination of total phenolic content (TPC)

Initially, the extracts were prepared in 30% DMSO and gallic acid was taken as the standard. Then, 20 μ L of each extract was mixed with 20 μ L of Folin-Ciocalteu reagent in a 96-well plate and shaken for 60 seconds, followed by a five-minute incubation period. After incubation, 200 μ L of 7% sodium carbonate and 10 μ L of 30% DMSO were added, shaken gently for sixty seconds, and again incubated for 120 minutes in darkness at room temperature. The absorbance was taken using a TECAN Infinite 200 PRO microplate reader and measured at 750 nm. The blank contained the solvent used to prepare the extract and standard. The TPC was expressed as μ g gallic acid equivalents (GAE)/mg of dried extract. All determinations were carried out in triplicates (Raduan et al., 2022).

2.4.2 Determination of total flavonoid content (TFC)

The extract, using 30% DMSO and the quercetin standard were prepared using the same treatment as the extract. 50 μ L of the extract was diluted in 100 μ L of 30% DMSO, then 20 μ L of 10% aluminum chloride was added to a 96-well plate, that was gently shaken and incubated for 3 minutes. After incubation, 60 μ L of 30% DMSO and 20 μ L of 1M sodium acetate were added. Then, the 96-well plate was

incubated for 40 min in darkness at room temperature. Finally, the absorbance was taken using a TECAN Infinite 200 PRO microplate reader and measured at 430 nm. The blank contained the solvent used to prepare the extract and standard. The TFC was shown as μg quercetin equivalents (QE)/mg of dried extract (Raduan et al., 2022).

2.5 Antioxidant activities

2.5.1 Ferric reducing antioxidant power (FRAP) assay

Initially, the above reagent was freshly made by adding 2.5 mL of 10 mM 2,4,6-tris(2-pyridyl)-S-pyrazine in 40 mM HCl, 2.5 mL of 20 mM ferric chloride, and 25 mL of 0.1 M acetate buffer with pH 3.6 in a Schott bottle and incubated for 10 minutes at 37°C. In a 96-well plate, 20 μL of the extract (prepared in 30% DMSO) was mixed with 40 μL of FRAP reagent, followed by the addition of 140 μL distilled water, resulting in the formation of a blue colored complex. The mixture was then incubated for 120 minutes in darkness at 37°C. Finally, the absorbance was taken using a TECAN Infinite 200 PRO microplate reader and measured at 593 nm. The blank contained the solvent used to prepare the extract and standard (ascorbic acid). The FRAP calculation was based on the ascorbic acid calibration curve and shown as μg ascorbic acid equivalents (AAE)/mg of dried extract (Nipun et al., 2021).

2.5.2 Inhibition activity of DPPH (2,2-diphenyl-1-picrylhydrazyl) radicals

The solution (0.2 mM) of DPPH was made by dissolving it (3.94 mg) in ethanol (50 mL). Later, 80 μL of DPPH solution was added to 20 μL of extract in a 96-well plate. The plate was then incubated in darkness for 20 minutes at room temperature. Subsequently, the absorbance was taken using a TECAN Infinite PRO microplate reader and measured at 540 nm. All the determinations were carried out in triplicates. The DPPH radical's percentage inhibition by the plant extracts was determined by the formula shown below:

$\% \text{ inhibition} = [(Ac - As) / Ac] \times 100\%$; where Ac is the absorbance of the control and As is the absorbance of the sample (Sarian et al., 2017).

2.6 α -glucosidase inhibitory activity

p-nitrophenyl- α -D-glucopyranoside (PNPG) as substrate was made after adding 6 mg of 4-Nitrophenyl- β -D-glucopyranoside in 20 mL of 50 mM phosphate buffer. In the 96-well plate, a total of 20 μL of plant's extract was combined with 100 μL of 30 mM buffer and 10 μL of AG enzyme (0.02) U/ μL , then was gently shaken and incubated for 5 minutes at room temperature. After incubation, 50 μL of PNPG was added and incubated for 5 minutes. After the enzyme, substrate and extract reacted, 70 μL of glycine was added to stop the reaction. The plate was incubated again for 10 minutes at room temperature before the measurement using a TECAN Infinite PRO microplate reader at 405 nm. The linear regression analysis was carried out to obtain the IC_{50} values. Quercetin, as a standard, was prepared using 50% DMSO and 50 mM phosphate buffer. The percentage inhibition activity of the enzyme was determined by the following equation:

$\% \text{ Inhibition} = [(Ac - As) / Ac] \times 100$; where Ac is the absorbance of the control and As is the absorbance of the sample (Sarian et al., 2017).

2.7 LC/MS QTOF analysis of extracts

The LC/MS-QTOF system used in this research included an Agilent 1200 liquid chromatography system with a vacuum degasser, autosampler, binary pump, and 6520 QTOF-MS with an Agilent ESI source. Chromatographic separation was achieved at 40 °C using Agilent ZORBAX Eclipse Plus C18 Rapid Resolution HT (2.1 x 100 mm) 1.8 μm with 0.1% formic acid in dH_2O (A) and 0.1% formic acid in acetonitrile (B) for positive mode ionization. The gradient elution program was 0.00 – 18.00 min, 5 - 95% (B); 18 to 23 min; 95% (B); 23.01 min, 5% (B). The total run time was 30 minutes, with 2 minutes re-equilibration before each injection. The sample injection volume was 2 μL , and the mobile phase flow rate was 0.25 mL/min. The MS was operated at 325 °C, with a gas flow of 11 L/min and nebulizer pressure at 35 psi. Agilent MassHunter Qualitative Analysis B.05.00 software

(Agilent Technologies, Santa Clara, CA, USA) was used for the data analysis (MS data (.d)). The analysis of chromatographic profiles was done using the METLIN database (Perumal et al., 2021).

2.8 Molecular docking analysis

The main bioactive compounds in *C. pentandra* extracts were investigated using *in-silico* molecular docking. The protein model employed was isomaltase (crystal structure) from *Saccharomyces cerevisiae* (PDB ID: 3A4A) (<https://www.rcsb.org/structure/3A4A>). The docking technique was optimized by re-docking the native ligand into the receptor on the 3A4A protein. The resulting root-mean-square deviation (RMSD) was less than 2, using the designed grid points X; Y; Z of 20; 26; 30, along with their coordinate grid points X; Y; Z of 21.243, -7.756, and 24.341, respectively. This specific grid box facilitated the docking of the compounds that were active and dominantly present in *C. pentandra* leaves extracts (Ahmed et al., 2018).

2.9 Physico-chemical and pharmacokinetics properties

Bioactive compounds identified through LC/MS-QTOF analysis were evaluated for their influence on drug absorption and distribution, considering factors such as solubility, lipophilicity, and molecular size (<http://www.swissadme.ch/index.php>, <https://biosig.lab.uq.edu.au/pkcsim/prediction>) (Ahmed et al., 2018).

2.10 Statistical analysis

The data were collected in three replicates and analyzed for comparative statistics e.g., mean \pm standard error, one-way analysis of variance (ANOVA), and Tukey's *post-hoc* analysis for a pair-by-pair multiple comparison. Values were considered significant when $p < 0.05$. The Sigma Plot (Systat Software, San Jose, California, USA) and IBM SPSS Statistics for Windows, Version 20.0 Armonk, NY: IBM Corp) were used to perform all statistical analyses.

3. Results and discussion

3.1 Yields of plant extracts

Dried powdered leaves of *C. pentandra* and *B. rubra* were successively macerated using dichloromethane, MeOH, and distilled water to obtain DCM, MeOH, and aqueous extracts, respectively. The various conventional and non-conventional extraction techniques employ solvents with varying degrees of polarity to selectively isolate bioactive compounds based on their differing polarities. Therefore, this strategy causes a difference in the resulting yield (Tasnuva et al., 2019). The percentage of yield for each extract is shown in Table 1.

The MeOH extract of *C. pentandra* leaves recorded the highest % yield, followed by the aqueous and DCM extracts. However, in the case of *B. rubra*, aqueous extract displayed the highest % yield followed by methanolic extract (Fig. 1S) (Supplementary file). According to Abookleesh et al. (2022), the skeletal, structural, and functional groups of secondary metabolites affect their solubility and confer on them a distinctive characteristic suitable for a wide range of medical uses. Similarly, non-polar and polar extracts from the leaves of *C. pentandra* and *B. rubra* were prepared using three different solvents with varying degrees of polarity to obtain secondary metabolites of different natures. Dichloromethane, a non-polar organic solvent, was used to extract non-polar secondary metabolites from the plant material; however, MeOH as a polar organic solvent and water as a polar inorganic solvent were employed to extract polar secondary metabolites from the plant material (Alshammari, 2021).

3.2 Phytochemical screening

The phytochemical screening of all plant extracts was done through previously described chemical tests to confirm the presence of secondary metabolites as bioactive compounds. The results displayed in Table 2 reveal that MeOH extract of *C. pentandra* leaves was the most potent in nature, exhibiting the presence of several secondary metabolites,

Table 1.
Percentage yield of *C. pentandra* and *B. rubra* leaves extracts.

Plant	Solvent	Percentage yield (%)
<i>Ceiba pentandra</i>	DCM	3.93 ± 0.04
	MeOH	7.76 ± 0.29
	H ₂ O	5.82 ± 0.16
<i>Basella rubra</i>	DCM	3.870 ± 0.04
	MeOH	15.00 ± 0.18
	H ₂ O	15.69 ± 0.10

DCM: Dichloromethane, MeOH: Methanol.

Table 2.
Phytochemical screening of *C. pentandra* and *B. rubra* leaves extracts.

Phytochemical screening	<i>Ceiba pentandra</i>			<i>Basella rubra</i>		
	DCM	MeOH	H ₂ O	DCM	MeOH	H ₂ O
a) Flavonoids						
1. Shinoda's test	+	+	+	+	+	-
2. Zinc-HCl test	+	+	+	+	+	-
b) Phenolics	+	+	+	+	+	+
c) Hydrolyzable Tannins	+	+	+	+	+	-
d) Alkaloids	-	-	-	+	-	-
e) Anthraquinones	-	+	-	-	-	-
f) Steroids	+	+	-	+	+	-
g) Terpenoids	+	+	+	+	-	-
h) Saponins	-	+	+	-	+	+

DCM: Dichloromethane, MeOH: Methanol. +: Positive, -: Negative

Table 3.
Total flavonoid content (TFC) and total phenolic content (TPC) results of *C. pentandra* and *B. rubra* leaves extracts.

Plants	Solvents	TFC (µg/mg QE)	TPC (µg/mg GAE)
<i>Ceiba pentandra</i>	DCM	72.59 ± 1.23 ^a	10.87 ± 0.14 ^a
	MeOH	15.36 ± 0.15 ^b	165.30 ± 5.86 ^b
	H ₂ O	2.90 ± 0.05 ^c	46.82 ± 1.63 ^c
<i>Basella rubra</i>	DCM	59.27 ± 1.02 ^d	6.27 ± 0.15 ^d
	MeOH	18.82 ± 0.10 ^e	54.86 ± 1.57 ^e
	H ₂ O	3.00 ± 0.01 ^c	27.00 ± 0.35 ^f

Note: Mean values (± standard deviation), triplicates, ^{a to f} significant difference (p<0.05). DCM: Dichloromethane, MeOH: Methanol.

namely flavonoids, phenolics, hydrolyzable tannins, anthraquinones, steroids, terpenoids, and saponins. DCM extracts of both plants showed the presence of alkaloids. However, alkaloids were found to be absent in the MeOH and aqueous leaf extracts of *C. pentandra* and *B. rubra*.

3.3 Total phenolic and total flavonoid contents (TPC & TFC)

The *C. pentandra* and *B. rubra* leaves MeOH extracts showed the highest TPC (165.30 ± 5.86 and 54.86 ± 1.57 µg/mg GAE), followed by aqueous (46.82 ± 1.63 and 27.00 ± 0.35 µg/mg GAE), and DCM extracts (10.87 ± 0.14 and 6.27 ± 0.15 µg/mg GAE). It has been confirmed by several studies that TPC has significantly correlated to the free radical scavenging effect. Polar solvents lead to the extraction of phenolic compounds, which could be one of the chief reasons for the methanol extracts exhibiting higher TPC than DCM (Alara et al. 2021).

The results of TFC and TPC are shown in Table 3 and Fig. 2S and 3S. From these results, it was observed that the DCM leaf extracts of *C. pentandra* and *B. rubra* had the highest flavonoid content (72.58 ± 1.23 and 59.27 ± 1.02 µg/mg QE) followed by methanolic extracts (15.36 ± 0.15 and 18.82 ± 0.10 µg/mg QE), and aqueous extracts (2.90 ± 0.05 and 3.00 ± 0.01 µg/mg QE). Generally, a high TPC correlates with a high TFC owing to the occurrence of several kinds of flavonoids in the plant extracts (Hikmawanti et al., 2021). However, the results of our study revealed a slightly different trend for the presence of phenolic compounds compared to previous findings. This trend was not found to be proportionate for methanolic and aqueous extracts, which could be ascribed to the existence of different types of phenolic substances, other than flavonoids, in the plant material. Furthermore, the flavonoids

detected in the DCM extract are likely to have hydrophobic groups or be methoxylated, acetylated, prenylated or have a smaller number of hydroxyl groups as substituents, thereby making them less polar in nature (Sarian et al., 2017).

Generally, phenolic compounds, including flavonoids, are a major group of secondary metabolites that serve as primary antioxidants. Variations in flavonoid or phenolic structures and group substitution influence the stability of phenoxy radicals, thus influencing the flavonoids' antioxidant properties (Shamsudin et al., 2022). Apart from the typical antioxidant properties, phenols (flavonoids in particular) can decrease their membrane fluidity, which stabilizes membranes, reduces the lipid membrane peroxidation, and consequently limits the diffusion of free radicals (Kulbat, 2016).

3.4 Biological activities: Antioxidants and α-glucosidase inhibitors

Plants' antioxidant properties are crucial in evaluating their medicinal effect and benefits. Researchers are actively investigating new plant sources to assess their efficacy, considering them possibly desirable alternatives to conventional medicines due to their minimum deleterious properties. Antioxidants isolated from plant-based extracts have been shown to be effective in counteracting a wide range of disorders, including cardiovascular disease, neurological disorders, cancer, and diabetes (Ahmed et al., 2020). These antioxidant effects are chiefly attributed to the existence of polyphenolic compounds, including flavonoids and tannins, which are the main polyphenolic compounds present in medicinal plants (Shamsudin et al., 2022). Several research studies have shown the antioxidant effects of aerial parts and roots extracts, including vegetables which denote a promising source of antioxidants with potential health benefits. However more research studies are still warranted to appropriately acknowledge and comprehend their bioavailability and medicinal functions (Alhassan et al., 2019).

This study determined the potency of the leaf extracts of *C. pentandra* and *B. rubra* in terms of antioxidant and α-glucosidase inhibitory properties. Both pharmacological properties are considered essential for using these traditional medicinal plants in treating diabetes and the complications associated with it. Therefore, the antioxidant and antidiabetic properties of leaves DCM, MeOH, and aqueous extracts of *C. pentandra* and *B. rubra* were determined through the *in vitro* DPPH, FRAP, and α-glucosidase tests, respectively (Table 4; Fig. 4S, 5S, 6S). The results of antioxidant assays displayed that the leaf methanol extracts of both plants exhibited the strongest antioxidant effects. Table 4 shows that the highest FRAP values are associated with the methanolic leaf extracts of *C. pentandra* (53.58 ± 0.25 µg/mg AAE) and *B. rubra* (10.17 ± 0.08 µg/mg AAE). Notably, the aqueous leaf extract of *C. pentandra* showed lower FRAP activity compared to its methanolic extract. The FRAP experiment evaluates the antioxidant effect of a compound by its ability to reduce ferric (Fe³⁺) to ferrous (Fe²⁺) ions. The results are often expressed in terms of ascorbic acid equivalents (AAE), typically in µg/mg. Comparison of FRAP values of the plant extract with those of a known standard helps to understand the extract's antioxidant capacity. In general, high FRAP values indicate a strong

Table 4.
Antioxidant and α-glucosidase inhibitory effects of *C. pentandra* and *B. rubra* leaves extracts.

Samples	Solvents	Antioxidant activities		α-glucosidase IC ₅₀ (µg/mL)
		FRAP (µg/mg AAE)	DPPH (IC ₅₀ µg/mL)	
<i>Ceiba pentandra</i>	DCM	3.92 ± 0.03	649.77 ± 4.56	NA
	MeOH	53.58 ± 0.25	27.71 ± 1.54	10.78 ± 0.48
	H ₂ O	17.66 ± 0.70	157.32 ± 3.44	109.54 ± 1.72
<i>Basella rubra</i>	DCM	7.76 ± 0.28	380.65 ± 2.63	NA
	MeOH	10.17 ± 0.08	253.76 ± 2.40	500.82 ± 3.38
Ascorbic acid				
	Quercetin	-	3.96 ± 0.03	-
				9.59 ± 0.29

Note: Mean values (± standard deviation) are for triplicate assay; NA: Not Active; FRAP: Ferric reducing antioxidant power; DPPH: 2,2-diphenyl-1-picrylhydrazyl; DCM: Dichloromethane, IC: Inhibitory concentration, MeOH: Methanol.

antioxidant effect, and low FRAP values suggest a weaker antioxidant effect (Sarian et al., 2017). Olugbami et al. (2014) mentioned that an ideal IC_{50} value for the DPPH assay should preferably be under $50 \mu\text{g/mL}$ to demonstrate higher antioxidant activity. A higher IC_{50} value would suggest a lower or inconsequential antioxidant effect with no therapeutic implications. Antioxidants play a significant role in inhibiting α -glucosidase, which can be useful for managing blood sugar levels and potentially preventing diabetes and its associated complications (Shamsudin et al., 2022). The correlation between antioxidants and α -glucosidase inhibition is particularly intriguing in the context of diabetes management. Studies have verified a positive correlation between the antioxidant activity of certain phenolic compounds and their capability to inhibit α -glucosidase. For instance, flavonoids, known for their potent antioxidant effects, have been shown to effectively hinder the action of α -glucosidase on the carbohydrates obtained from diet (Alhassan et al., 2019). This inhibition mechanism typically involves the binding of these antioxidative compounds to the active or allosteric sites of the α -glucosidase, thereby decreasing its action of the hydrolysis of carbohydrates (Sarian et al., 2017).

The one-way ANOVA showed that the plant-solvent interaction had a significant effect on the FRAP antioxidant assay ($\mu\text{g/mg AAE}$) (Plant * Solvent; $df = 2$; $F = 2731.707$; $p < 0.001$). To understand the differences among various solvents in greater detail, Tukey's post-hoc analysis was employed. It revealed that the MeOH extract had the highest FRAP values in comparison to the other two solvents. The MeOH extract had a significantly higher FRAP value than the DCM extract (mean difference = 26.0333 ; $p < 0.001$) followed by the aqueous extract (mean difference = 18.7167 ; $p < 0.001$) (Table 1S).

The one-way ANOVA showed that the plant-solvent interaction had a significant effect on the DPPH assay ($IC_{50} \mu\text{g/mL}$) (Plant * Solvent; $df = 2$; $F = 10502.459$; $p < 0.001$). To understand the differences among various treatments in greater details, Tukey's post-hoc analysis was employed (Table 2S). The pair-by-pair comparison through the post-hoc analysis confirmed the significantly stronger antioxidant activity of *C. pentandra* leaves. It had the strongest DPPH values in comparison to the other *B. rubra* extracts. The *C. pentandra* extracts had significantly lower DPPH values than the *B. rubra* extracts (mean difference = -153.7967 ; $p < 0.001$). In addition to this, the MeOH extract had the highest activity, followed by the aqueous extract (mean difference = -268.8183 ; $p <$

0.001) and the DCM extract (mean difference = -374.4783 ; $p < 0.001$). It was also found that the control (ascorbic acid) had a significantly higher antioxidant activity than the plant extracts (Table 2S).

The one-way ANOVA showed that the plant-solvent interaction had a significant effect on the α -glucosidase IC_{50} ($\mu\text{g/mL}$) assay (Plant * Solvent; $df = 1$; $F = 152.043$; $p < 0.001$). To understand the differences among various treatments in greater detail, Tukey's post-hoc analysis was employed (Table 3S). The post-hoc analysis further confirmed the significantly stronger *C. pentandra* leaves activity on α -glucosidase. It had the strongest IC_{50} ($\mu\text{g/mL}$) values in comparison to *B. rubra* extract. The *C. pentandra* extract had a significantly lower IC_{50} value than *B. rubra* extract (mean difference = -456.00 ; $p < 0.001$). The analysis further revealed that in addition to this, the MeOH extract had the higher activity than the aqueous extract (mean difference = -59.7133 ; $p < 0.001$). The results also confirmed that the control (Quercetin) had significantly higher α -glucosidase activity than the plant extracts (Table 3S).

3.5 Identification of the bioactive compounds using LC-MS QTOF

A cutting-edge QTOF LC-MS was used to identify the potent bioactive substances possessing antioxidative and α -glucosidase inhibitory effects present in the *C. pentandra* leaves. This advanced, sophisticated analytical method links the separation capabilities of liquid chromatography (LC) to mass spectrometry (MS) using a quadrupole time-of-flight (TOF) analyzer. This technique is generally employed to quantify, characterize and identify bioactive agents present in plant extracts with higher accuracy, precision and confidence (Irakli et al., 2021). Moreover, this method also allows for the exact quantification of the biologically active agents, which is necessary to understand the potency and effectiveness of the therapeutically active plant-based extracts (Mustikasari et al., 2024). Additionally, it provides powerful quantitative and qualitative abilities and is widely considered one of the most modern and promisingly accurate mass instrumentation methods available (Zhang et al., 2015). Evaluating the levels of the major compounds present in the *C. pentandra* leaves aqueous and methanolic extracts contributes to a better comprehension of their potential health benefits associated with antioxidant activity and α -glucosidase inhibitory results. The chromatograms are displayed in Fig. 2, and the predicted compounds are summarized in Table 5.

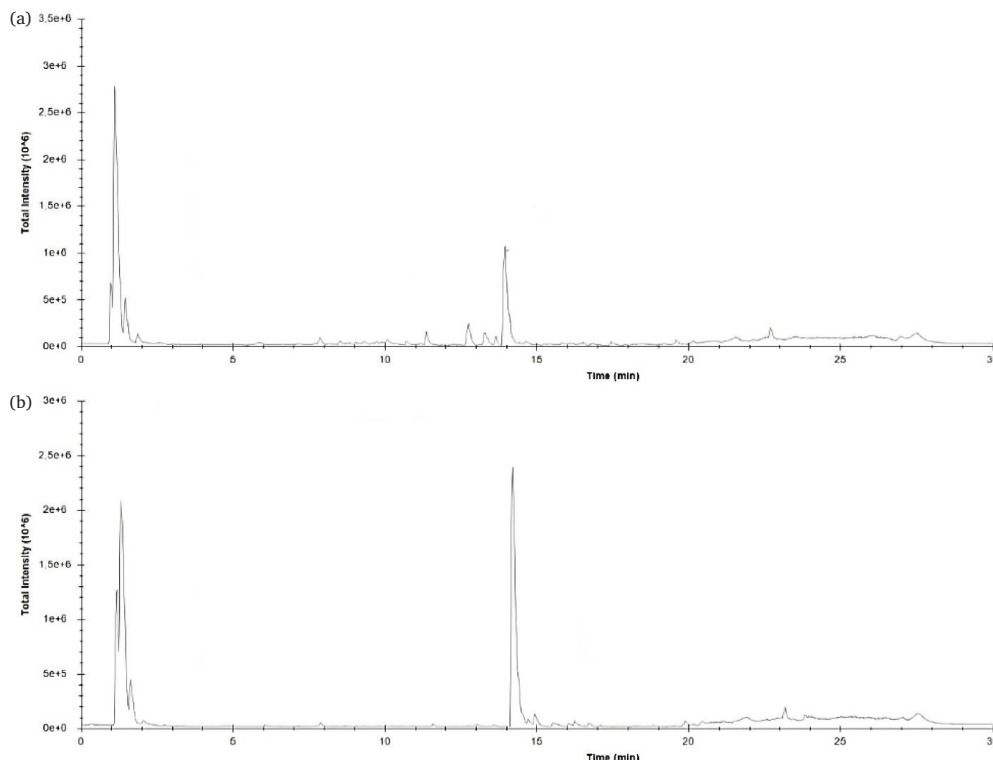


Fig. 2. LC-MS QTOF Chromatogram: (a) *C. pentandra* leaves methanol extract; (b) *C. pentandra* leaves aqueous extract. LC-MS QTOF: Liquid chromatography-mass spectrometry with quadrupole time-of-flight.

Table 5.

List of the dominant compounds detected from *C. pentandra* leaves methanol (CPM) and aqueous (CPW) extracts observed by LC-MS QTOF analysis.

No	Rt	Compounds	Score	M/Z	Levels	
					CPM	CPW
1	1.147	4-Methylaminobutyrate	99.75	118.0862	19744726	-
2	1.334	Isoamyl nitrite	99.30	118.0865	-	8363255
3	1.412	11-amino-undecanoic acid	95.75	202.1804	-	142352
4	1.439	<i>m</i> -Coumaric acid	97.47	182.0811	164399	-
5	1.482	L-Leucine	98.47	132.1015	1885364	761606
6	1.639	Citric acid	96.36	210.0604	-	360712
7	5.886	<i>cis</i> - β -D-Glucosyl-2-hydroxycinnamate	97.86	344.1337	201237	-
8	7.896	Luteolin 7-rhamnosyl(1->6) galactoside	98.02	595.1652	272497	174210
9	8.549	Avenanthramide 2s	99.13	360.1081	192313	-
10	9.168	Robinetin 3-rutinoside	93.2	611.1592	36431	-
11	10.097	Melanoxetin	96.49	303.0499	141642	-
12	10.098	Scutellarein 7-glucoside	98.98	449.1076	126568	-
13	11.378	Torosaf flavone B 3'-O- β -D-glucopyranoside	97.43	593.1872	775330	-
14	11.600	2"-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin	98.93	-	-	142678
15	12.758	11-hydroperoxy-12,13-epoxy-9-octadecenoic acid	99.31	346.2585	509040	-
16	13.305	α -9(10)-EpODE	98.66	295.2272	182743	-
17	13.311	5,8,12-trihydroxy-9-octadecenoic acid	99.04	348.2747	331585	-
18	13.671	9-keto palmitic acid	93.72	271.2255	168196	-
19	13.99	Xestoaminol C	98.2	230.248	3856468	363783
20	14.215	C16 Sphinganine	95.3	274.2752	4415674	15195404
21	14.365	Phytosphingosine	98.9	318.3003	-	502697
22	14.646	C17 Sphinganine	95.52	288.2894	141404	535367
23	26.084	Docosanedioic acid	98.62	371.3158	171753	-

Note: "CPM" referred to the methanol extract of *Ceiba pentandra* and "CPW" referred to the aqueous extract of *Ceiba pentandra*. RT: Retention time, M/Z: Mass-to-charge ratio, LC-MS QTOF: Liquid chromatography-mass spectrometry with quadrupole time-of-flight.

Based on the QTOF LC-MS results (Table 5), some compounds that are present in the *C. pentandra* leaves methanolic extract are similar to those in the water extract. For instance, L-leucine, luteolin 7-rhamnosyl(1->6) galactoside, xestoaminol C, C16 sphinganine, and C17 sphinganine were detected in both polar extracts. However, most of the flavonoids and their derivatives were present in the methanolic extract. Aqueous extract of *C. pentandra* contained isoamyl nitrite, 11-amino-undecanoic acid, citric acid, 2"-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin, and phytosphingosine, which were not present in the methanol extract (Table 5). These compounds might influence the extract's antioxidant activity and ability to inhibit the α -glucosidase. The MeOH extracts of barks and leaves of *C. pentandra* and their fractions highlighted many plant polyphenols (Nelly et al. 2023; Orabi et al. 2024). Table 4 depicts that the leaf MeOH extract of *C. pentandra* shows noteworthy potential in terms of antioxidant activity and α -glucosidase inhibition.

Based on Table 5, flavonoids and their derivatives were identified in the *C. pentandra* leaves methanolic extract. According to Chen et al. (2021), flavonoids and phenolic compounds are secondary metabolites of plants, known for their remarkable bioactive properties. Flavonoids, which are polyphenols, are considered some of the most bioactive molecules and exhibit inhibitory activities against α -glucosidase (Dirir et al., 2022). The structures of different identified flavonoids are shown in Fig. 3.

The leaf methanolic extract of *C. pentandra* contains five different flavonoids (a-e), while the aqueous extract only contains two: luteolin 7-rhamnosyl(1->6)galactoside and 2"-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin (Fig. 3). *In vitro* assessments suggest that the *C. pentandra* leaf methanolic extract demonstrates the highest effect in inhibiting DPPH stable free radical and α -glucosidase enzyme. The *C. pentandra* leaves aqueous extract demonstrates moderate effect in both DPPH (IC₅₀: 157.32 \pm 3.44 μ g/mL) and α -glucosidase inhibitions (IC₅₀: 109.54 \pm 1.72 μ g/mL). Especially, luteolin 7-rhamnosyl(1->6) galactoside has earlier been reported to be present in the active extract showing enzyme inhibitory effect against α -glucosidase enzyme (Ali et al., 2020), while 2"-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin

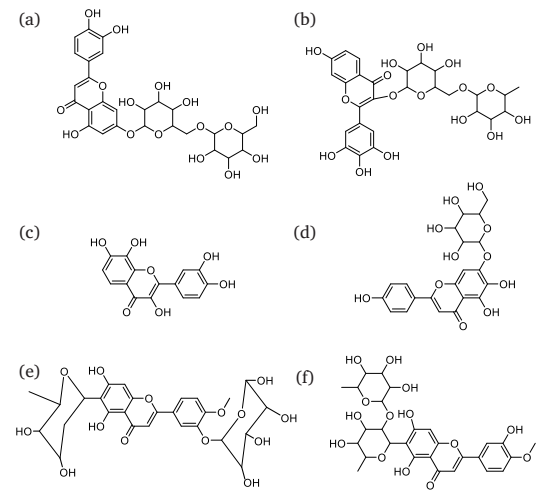


Fig. 3. The identified flavonoids of *C. pentandra* leaf's extract. (a) Luteolin 7-rhamnosyl (1->6) galactoside; (b) Robinetin 3-rutinoside; (c) Melanoxetin; (d) Scutellarein 7-glucoside; (e) Torosaf flavone B 3'-O- β -D-glucopyranoside; (f) 2"-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin.

has not yet been reported by any research to exhibit enzyme inhibitory effects against carbohydrate hydrolyzing enzymes including α -glucosidase enzyme.

The five different flavonoids identified in the *C. pentandra* leaf methanolic extract (Fig. 3) may be one of the factors responsible for the significant inhibition of both DPPH radical and α -glucosidase. In this regard, melanoxetin has been reported to be potent in preventing diabetes via several pathways including the inhibition of α -glucosidase (Rocha et al., 2024). Moreover, another flavonoid named robinetin 3-rutinoside was identified in the extract of *M. buxifolia* through ultrahigh performance liquid chromatography – Mass spectrometry (UHPLC-MS) in negative ion mode. This further proved the plant's ability to reduce the carbohydrate hydrolyzing action of both α -amylase and α -glucosidase (Ali et al., 2020). The structural features of torosaf flavone B 3'-O- β -D-glucopyranoside illustrate potential antioxidant and α -glucosidase inhibition effects owing to their resemblance with other flavonoid glycosides, for example, quercetin-3-O- β -D-glucopyranoside. An aglycon part of quercetin may be responsible for the manifestation of antioxidative and other biological properties, while the attached glucopyranoside moiety glucopyranoside might have interacted with the α -glucosidase's active site, thereby inhibiting enzyme interaction and reducing carbohydrate breakdown/hydrolysis. Though other two flavonoids, namely luteolin 7-rhamnosyl (1->6)galactoside and scutellarein 7-glucoside have not yet been extensively evaluated for their biological effects through any studies, the stronger α -glucosidase inhibitory activity of the methanolic extract of *M. buxifolia* could likely be due to the presence of both of these (Ali et al., 2020). Therefore, further discussion post molecular docking analysis will elaborate on the identified compounds in *C. pentandra* extract as α -glucosidase inhibitors.

3.6 The prediction of the identified active compounds as α -glucosidase inhibitors using an *in-silico* approach

Active compounds identified from the LC-MS based analysis were evaluated for their interaction with the 3A4A protein. The identified compounds from the potent extracts of this herb are reported in Table 5. The molecular docking analysis results, which are provided in Table 6, clearly explain the re-docking of the native ligand, which produced an affinity binding energy (ABE) of -6.0 kcal/mol, with a RMSD (root mean square deviation) of 0.491. This finding was found to be like and even better than the previous finding published by Nipun et al. (2020), in which an RMSD value of 0.633 was reported.

Table 6 shows that quercetin can interact with 3A4A protein with an ABE of -8.4 kcal/mol. Compounds from *C. pentandra* extract were chosen based on having an ABE of less than -8.00 kcal/mol, as quercetin's activity as an α -glucosidase inhibitor, with an IC₅₀ of 9.59 \pm 0.30 μ g/

Table 6.

The profiling of dominant compounds of *C. pentandra* leaves extracts interaction with 3A4A protein.

Retention Time (min)	Dominant identified compounds of CPM		Retention Time (min)	Dominant identified Compounds of CPW	
	Methanol extract	Affinity (kcal/mol)		Aqueous extract	Affinity (kcal/mol)
-	Native ligand (ADG)	-6.0	-	Quercetin	-8.4
1.147	4-Methylaminobutyrate	-4.5	1.334	Isoamyl nitrite	-5.1
1.439	<i>m</i> -Coumaric acid	-8.9	1.412	11-amino-undecanoic acid	-5.8
1.679	L-Leucine	-5.3	1.639	Citric acid	-6.0
5.886	<i>cis</i> - β -D-Glucosyl-2-hydroxycinnamate	-8.1	11.600	2''-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin	-8.9
7.896	Luteolin 7-rhamnosyl(1->6) galactoside	-10.3	14.215	C16 Sphinganine	-6.1
8.549	Avenanthramide 2s	-8.3	14.365	Phytosphingosine	-5.8
9.168	Robinetin 3-rutinoside	-8.6	14.646	C17 Sphinganine	-6.1
10.097	Melanoxetin	-8.9			
10.098	Scutellarein 7-glucoside	-9.7			
11.378	Torosafavone B 3'-O- β -d-glucopyranoside	-10.0			
12.758	11-hydroperoxy-12,13-epoxy-9-octadecenoic acid	-6.8			
13.305	α -9(10)-EpODE	-6.7			
13.311	5,8,12-trihydroxy-9-octadecenoic acid	-6.6			
13.671	9-keto palmitic acid	-6.0			
13.990	Xestoaminol C	-5.9			
26.084	Docosanedioic acid	-6.4			

Note: "CPM" referred to the methanol extract of *Ceiba pentandra* and "CPW" referred to the aqueous extract of *Ceiba pentandra*.

mL sets a benchmark. Probably, the selected compounds with ABEs less than -8.00 kcal/mol are active and comparable to quercetin. According to Table 6, nine compounds have ABE values less than -8.00 kcal/mol. These compounds include *m*-coumaric acid (hydroxycinnamic acid), *cis*- β -D-Glucosyl-2-hydroxycinnamate, luteolin 7-rhamnosyl(1->6) galactoside, avenanthramide 2s, robinetin 3-rutinoside, melanoxetin, scutellarein 7-glucoside, torosafavone B 3'-O- β -D-glucopyranoside, and 2''-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin.

Table 7.

Binding types of the selected compound from *C. pentandra* leaves extracts towards the 3A4A protein.

Compounds	Compound types	Amino acids interaction			
		Hydrogen binding	Bond distance (Å)	Other bindings	Bond distance (Å)
Native ligand (ADG)	Control 1	ARG213; ASP352; HIS351; ARG442; ASP69; HIS112	6.46; 2.05&4.52; 2.04; 1.86; 1.74&2.89; 2.51		
Quercetin	Control 2	ASP215	2.86	ARG442; ASP352; PHE303	3.77; 4.30; 5.14&4.97
The selected compounds					
<i>m</i> -Coumaric acid	Cinnamic acid derivate	TYR158	2.88	TYR158; ASP352; ARG442; PHE178; HIS112; SER240	2.76; 3.99; 4.66&4.61; 4.36; 3.55; 3.73; 4.37; 3.53
<i>cis</i> - β -D-Glucosyl-2-hydroxycinnamate	Cinnamic acid derivate	GLU411; ARG442; GLN353; ASP307;	3.37; 2.29; 3.73; 3.22	ARG315	4.67
Luteolin 7-rhamnosyl(1->6) galactoside	Flavonoid derivate	ASN415; GLN279; GLN353	1.90&2.65; 2.42&2.61; 2.98	GLY160; LYS156; PHE314; ARG315; ASP352; ASP215; PHE178	3.77; 5.25; 4.71; 4.48&5.27; 2.70; 3.40; 2.97
Avenanthramide 2s	Phenolic alkaloids	ASP352; PRO312; ARG315; LEU313; LYS156	2.28; 2.95; 3.15; 2.62; 2.28; 1.96	PHE178; GLU277; PHE152; PHE303; GLN353; PHE314; ARG442	4.47; 2.70; 4.98; 4.37; 3.63; 5.79; 4.08
Robinetin 3-rutinoside	Flavonoid derivate	ASP215; GLN353; THR306	2.00; 1.68; 2.31	TYR158; ARG442; ASP352; PHE303	3.56&3.69; 3.77&3.46; 4.43&4.67; 3.50
Melanoxetin	Flavonoid	ASP352; GLU277; ASP307; HIS280	2.14&2.54; 2.12; 2.27; 2.43	TYR158; ARG315; ARG442	5.67; 4.78&4.31; 4.76
Scutellarein 7-glucoside	Flavonoid derivate	ASP69; ASP307	2.91; 2.18	VAL216; ASP352; PHE303; ARG315	5.32; 4.19; 4.87&5.40; 2.22&1.63
Torosafavone B 3'-O- β -D-glucopyranoside	Flavonoid derivate	ASP69; ASP215; GLU277; ARG315; SER157; ARG442	1.65; 2.55; 1.41; 2.14; 2.20; 6.25	HIS351; TYR72; TYR158; ASP215; ARG315;	4.49; 3.98; 5.18&4.95; 3.27; 3.56
2''-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin	Flavonoid derivate	LYS156; TYR158; ASP215	2.37; 2.94; 2.25	SER240; TYR158; ARG442; ASP352; PHE178; HIS112	3.49; 2.94; 4.64; 4.72&3.96; 3.48; 4.39; 3.78; 4.27

Generally, flavonoids and their derivatives are active in inhibiting α -glucosidase, as reported in Table 6. As previously explained, luteolin 7-rhamnosyl(1->6)galactoside, scutellarein 7-glucoside, torosafavone B 3'-O- β -D-glucopyranoside, and 2''-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin have not been studied extensively. This study has identified new potent flavonoid derivatives as α -glucosidase inhibitors. Therefore, it is recommended that future research should focus on isolating and synthesizing these compounds.

Furthermore, *m*-coumaric acid, at 100 mg/kg bw, reduced blood glucose levels in rats (Amalan et al., 2016). Additionally, *cis*- β -D-glucosyl-2-hydroxycinnamate shares a chemical structure with *m*-coumaric acid, a cinnamic acid derivative. However, biological testing is still required to confirm its medicinal potential, specifically to evaluate its anti-diabetic, antioxidant and anti-inflammatory effects. Avenanthramide 2s, a unique phenolic compound that functions as an antidiabetic agent and may assist in diabetes management by controlling postprandial glucose levels (Zhouyao et al., 2022). Additionally, these untested compounds should be validated by interacting with the catalytic sites of the 3A4A protein.

Table 7 also shows how the selected compounds interact with amino acid residues. The native ligand (α -D-glucose) forms hydrogen bonds with several catalytic residues, including ASP352 and ASP69; ARG442 (Fig. 4a). While luteolin 7-rhamnosyl(1->6)galactoside lacks catalytic residues that interact through hydrogen bonds, it does have two catalytic residues that bind via other interactions (Table 7). The interactions of the selected compounds of this herb are shown in Fig. 4b. Moreover, torosafavone B 3'-O- β -D-glucopyranoside is another potent α -glucosidase inhibitor that interacts with three catalytic sites. Fig. 4c illustrates how the catalytic residues interact with torosafavone B 3'-O- β -D-glucopyranoside via hydrogen bonding. The pyranose structure of torosafavone B 3'-O- β -D-glucopyranoside interacts with two catalytic residues, potentially enhancing its inhibitory effects on α -glucosidase.

Fig. 5 illustrates how the selected compounds interact within the catalytic sites. Luteolin 7-rhamnosyl(1->6)galactoside and torosafavone B 3'-O- β -D-glucopyranoside are the most potent compounds, with affinity binding energies of -10.3 and -10.0 kcal/mol, respectively. Furthermore, Fig. 5 shows the interactions of multiple compounds. Compounds localized in the catalytic site of the 3A4A protein inhibit α -glucosidase activity. An *in vitro* investigation demonstrated that the *C. pentandra* leaf methanolic extract inhibits α -glucosidase activity, similar to quercetin (Table 4). Hence, it may have exhibited its antioxidant and α -glucosidase inhibitory effects due to the existence of these and other unknown bioactive compounds possessing α -glucosidase inhibitory effects.

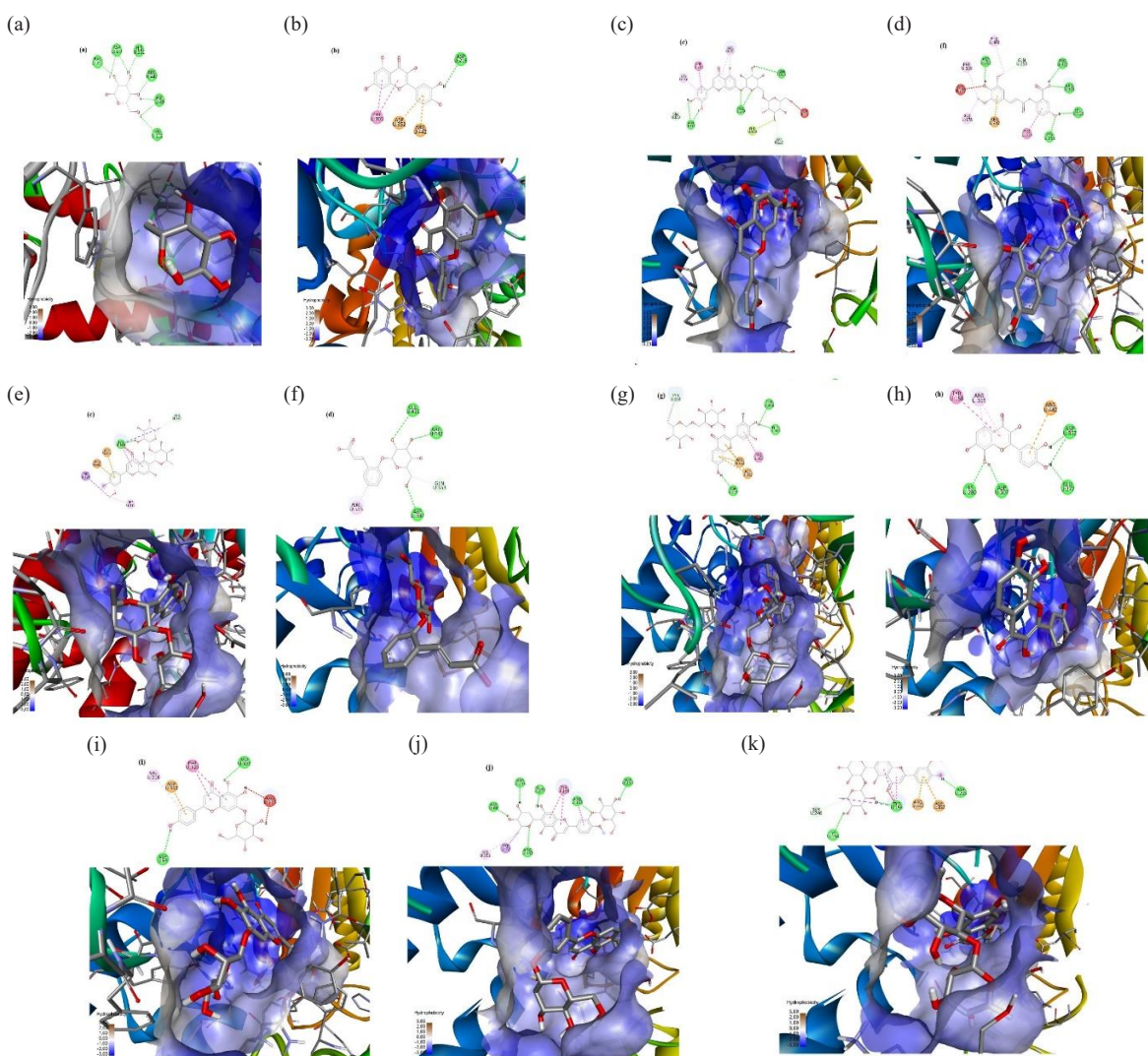


Fig. 4. The interaction of the selected compounds with 3A4A protein: (a) Native ligand; (b) Quercetin; (c) *m*-Coumaric acid; (d) *cis*- β -D-Glucosyl-2-hydroxycinnamate; (e) Luteolin 7-rhamnosyl(1->6)galactoside; (f) Avenanthramide 2s; (g) Robinetin 3-rutinoside; (h) Melanoxetin; (i) Scutellarein 7-glucoside; (j) Torosaflavone B 3'-O- β -D-glucopyranoside; (k) 2''-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin.

3.7 Physico-chemical properties and pharmacokinetics of the selected compounds of *C. pentandra* extracts

Compounds from *C. pentandra* extracts have been predicted to inhibit α -glucosidase through strong interactions with the 3A4A protein, as reported previously. These compounds were studied further for their physicochemical and pharmacokinetic properties to determine their potency for human use. Table 8 shows that the selected compounds from *C. pentandra* extracts include *m*-coumaric acid, which has a high absorption rate (>90%). Furthermore, the methanolic leaf extract of *C. pentandra* contains numerous potent compounds with drug-like properties. All selected compounds were found to be devoid of hepatotoxic effects, indicating they are safe for human use.

According to Table 8, a few compounds in the *C. pentandra* leaves extract do not comply with the Lipinski Rule of 5. This implies that these compounds do not possess drug-like properties and cannot easily cross the cell membrane, rendering them ineffective as natural drugs. Andhiarto et al. (2022) found that the ability of drugs to penetrate the cell membrane follows at least two Lipinski's rules. Besides, Tijjani et al. (2022) stated that drugs administered orally should not follow more than one of the Lipinski rules. Table 8 shows that *C. pentandra* leaf extracts contain active compounds such as *m*-coumaric acid, *cis*- β -D-Glucosyl-2-hydroxycinnamate, avenanthramide 2s, and melanoxetin, which have drug-likeness properties. Melanoxetin is a hydroxylated

flavonoid that has earlier been shown to demonstrate antioxidant, anti-inflammatory, and antidiabetic properties (Rocha et al., 2024). It also lacks the toxicity exhibited by AMES toxicity. Melanoxetin has a similar structure to quercetin. Quercetin showed no toxic properties based on AMES toxicity or mutagenic effects (Amar, 2017). Furthermore, *m*-coumaric acid and melanoxetin are predicted to be antidiabetic agents based on their physicochemical and pharmacokinetic properties.

4. Conclusions

The results revealed that both non-polar and polar extracts from *C. pentandra* and *B. rubra* leaves demonstrated antioxidant and alpha-glucosidase inhibitory effects. Notably, the methanolic and aqueous extracts of *C. pentandra* exhibited potent antioxidant and α -glucosidase inhibitory effects. The methanolic extract of *C. pentandra* was more effective in inhibiting DPPH and α -glucosidase than the aqueous extract. Based on this study's findings, luteolin 7-rhamnosyl(1->6)galactoside, torosaflavone B 3'-O- β -D-glucopyranoside and scutellarein 7-glucoside are predicted to be α -glucosidase inhibitors due to their lowest affinity binding energies of -10.3 kcal/mol, -10.0 kcal/mol and -9.7 kcal/mol., respectively. These flavonoids could serve as leads for the discovery of safe antidiabetic agents acting as α -glucosidase inhibitors. Further research is needed to isolate these compounds and evaluate their *in vivo* antidiabetic effects.

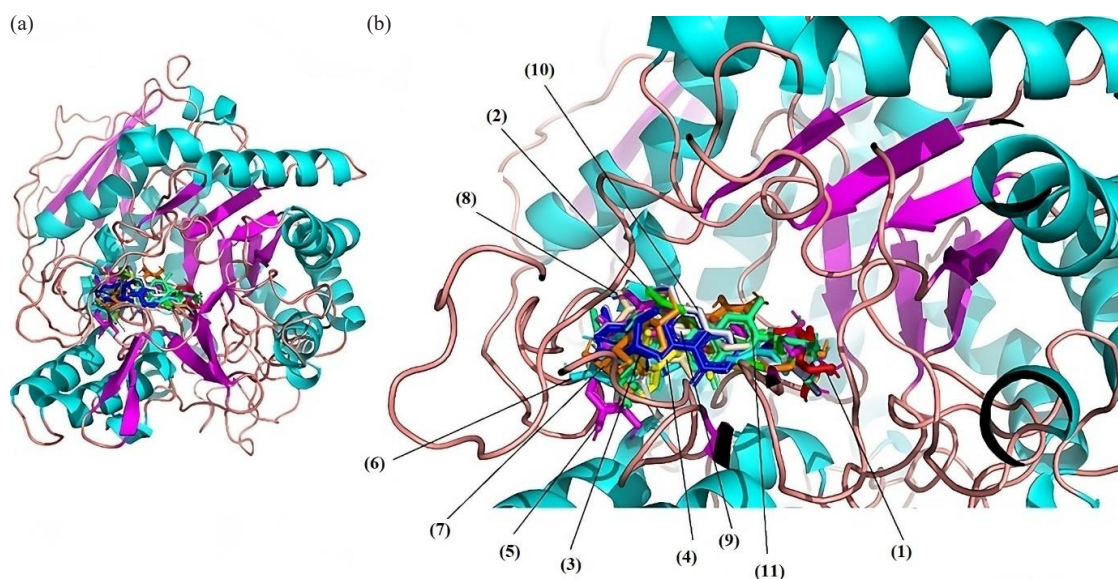


Fig. 5. Super impose of the interaction of the selected compounds with 3A4A protein: (a) normal protein; (b) zoom 20 Å. (1) Native ligand; (2) Quercetin; (3) m-Coumaric acid; (4) *cis*- β -D-Glucosyl-2-hydroxycinnamate; (5) Luteolin 7-rhamnosyl(1->6)galactoside; (6) Avenanthramide 2s; (7) Robinetin 3-rutinoside; (8) Melanoxetin; (9) Scutellarein 7-glucoside; (10) Torosaflavone B 3'-O- β -D-glucopyranoside; (11) 2''-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin.

Table 8.

The physicochemical and pharmacokinetic properties of the selected compounds from *C. pentandra* leaves extracts.

Compound types	Properties										
	Physicochemical						Pharmacokinetic				
	Log P	NRB	H-Ac	H-Do	DL	NVL	IA	AT	LD ₅₀	MTD	HT
<i>m</i> -Coumaric acid	1.49	2	3	2	Yes	0	92.864	No	2.232	1.232	No
<i>cis</i> - β -D-Glucosyl-2-hydroxycinnamate	-2.37	5	8	4	Yes	0	16.283	No	2.174	0.764	No
Luteolin 7-rhamnosyl(1->6)galactoside	-2.42	7	16	10	No	3	13.585	No	2.478	0.464	No
Avenanthramide 2s	2.46	6	6	4	Yes	0	42.313	No	2.648	0.606	No
Robinetin 3-rutinoside	-1.69	6	16	10	No	3	20.596	No	2.49	0.443	No
Melanoxetin	1.99	1	7	5	Yes	0	74.413	No	2.485	0.522	No
Scutellarein 7-glucoside	-0.24	4	11	7	No	2	36.191	No	2.544	0.598	No
Torosaflavone B 3'-O- β -D-glucopyranoside	-0.37	6	14	8	No	3	31.35	Yes	2.548	0.822	No
2''-O- α -L-Rhamnosyl-6-C-fucosyl-3'-methoxyluteoiiin	-0.02	5	14	8	No	3	40.663	No	2.538	0.585	No

Log P: Predicted octanol/water partition coefficient; NRB: No. of rotatable bonds; H-Ac: No. of hydrogen bonds; H-Do: No. of hydrogen donor; DL: Drug likeness; NVL: No. of Lipinski's rule violations; IA: Internal absorption; AT: AMES toxicity; LD₅₀: Oral rat acute toxicity, mol/kg; MTD: Maximum tolerated dose for human (log mg/kg/day); HT: Hepatotoxicity.

CRediT authorship contribution statement

Zahradifa Kaniabila Ananda: Literature Search, Experimental Studies, Data Acquisition, Data Analysis, Manuscript Preparation; **Taslima Begum:** Manuscript Editing and Review, Experimental Studies; **Mustofa Ahda:** Experimental Studies, Data Acquisition, Data Analysis, Manuscript Editing and Review; **Mohd Salleh Rofiee:** Data Acquisition, Manuscript Editing and Review; **Syed Adnan Ali Shah:** Data Acquisition, Manuscript Editing and Review; **Mohd Zaki Salleh:** Data Acquisition, Manuscript Editing and Review; **Bader O Almutairi:** Manuscript Editing and Review; **Syed Najmul Hejaz Azmi:** Data Analysis, Statistical Analysis, Manuscript Editing and Review; **Pankaj Sah:** Data Analysis, Statistical Analysis, Manuscript Editing and Review; **Agustin Krisna Wardani:** Manuscript Editing and Review; **Alfi Khatib:** Manuscript Editing and Review; **Syed Atif Abbas:** Manuscript Editing and Review; **Md. Abdur Rashid Mia:** Manuscript Editing and Review; **Qamar Uddin Ahmed:** Concepts, Design, Experimental Studies, Data Acquisition, Data Analysis, Statistical Analysis, Manuscript Editing and Review. All authors approved the final version of the manuscript.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Declaration of Generative AI and AI-assisted technologies in the writing process

The authors confirm that there was no use of artificial intelligence (AI)-assisted technology for assisting in the writing or editing of the manuscript and no images were manipulated using AI.

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Appendix A. Supplementary material

Supplementary material to this article can be found online at https://dx.doi.org/10.25259/JKSUS_16_2024

References

Abokleesh, F.L., Al-Anzi, B.S., Ullah, A., 2022. Potential antiviral action of alkaloids. *Molecules* 27, 903.

- Ahmed, Q., Ali, A., Mukhtar, S., Alsharif, M., Parveen, H., Sabere, A., Nawi, M., Khatib, A., Siddiqui, M., Umar, A., Alhassan, A., 2020. Medicinal potential of isoflavonoids: Polyphenols that may cure diabetes. *Molecules* 25, 5491. <https://doi.org/10.3390/molecules25235491>
- Ahmed, Q., Sarian, M., Mat, S., ad, S., Latip, J., Arief Ichwan, S., Hussein, N., Taher, M., Alhassan, A., Hamidon, H., Fakurazi, S., 2018. Methylation and acetylation enhanced the antidiabetic activity of some selected flavonoids: In vitro, molecular modelling and structure activity relationship-based study. *Biomolecules* 8, 149. <https://doi.org/10.3390/biom8040149>
- Ajiboye, B., Diayi, A., Agunbiade, S., Akinyemi, A., Adewale, O., Ojo, O., 2021. Ameliorating activity of polyphenolic-rich extracts of *Basella rubra* L. leaves on pancreatic β -cell dysfunction in streptozotocin-induced diabetic rats. *J. Complement Integr. Med.* 19, 335-344. <https://doi.org/10.1515/jcim-2020-0304>
- Alara, O.R., Abdurahman, N.H., Ukaegbu, C.I., 2021. Extraction of phenolic compounds: A review. *Current Res. Food Sci.* 4, 200-214. <https://doi.org/10.1016/j.crf.2021.03.011>
- Alhassan, A.M., Ahmed, Q.U., Latip, J., Shah, S.A.A., 2019. A new sulphated flavone and other phytoconstituents from the leaves of *Tetracera indica* Merr. and their α -glucosidase inhibitory activity. *Natural Product Res.* 33, 1-8. <https://doi.org/10.1080/14786419.2018.1437427>
- Ali, J., Saleem, H., Mannan, A., Zengin, G., Mahomoodally, M., Locatelli, M., Abidin, S., Ahemad, N., Zia, M., 2020. Metabolic fingerprinting, antioxidant characterization, and enzyme-inhibitory response of *Monotheca buxifolia* (Falc.) A. DC. extracts. *BMC Complement Med. Ther.* 20, 313. <https://doi.org/10.1186/s12906-020-03093-1>
- Alshammari, O. (2021). Extraction of natural products using deep eutectic solvents. Doctor of Philosophy Thesis. University of Leicester, UK.
- Andhiarto, Y., Suciati, S., Praditapusa, E.N., Sukardiman, S., 2023. In silico analysis and admet prediction of flavonoid compounds from *syzygium cumini* var. album on α -glucosidase receptor for searching anti-diabetic drug candidates. *Pharmacogn J.* 14, 736-743. <https://doi.org/10.5530/pj.2022.14.161>
- Amalan, V., Vijayakumar, N., Indumathi, D., Ramakrishnan, A., 2016. Antidiabetic and antihyperlipidemic activity of p-coumaric acid in diabetic rats, role of pancreatic GLUT 2: In vivo approach. *Biomed Pharmacother* 84, 230-236. <https://doi.org/10.1016/j.biopha.2016.09.039>
- Ammar, O., 2017. In silico pharmacodynamics, toxicity profile and biological activities of the Saharan medicinal plant *Limoniastrum feei*. *Braz. J. Pharm. Sci.* 53. <https://doi.org/10.1590/s2175-97902017000300061>
- Chan, E., Yeong, S., Wong, C., Soo, O., Phua, A., Ng, Y., 2023. *Ceiba pentandra* (L.) Gaertn.: An overview of its botany, uses, reproductive biology, pharmacological properties, and industrial potential. *J. Applied Biology Biotech.* 11, 1-7. <http://dx.doi.org/10.7324/JABB.2023.110101>
- Chen, Q., Wang, X., Yuan, X., Shi, J., Zhang, C., Yan, N., Jing, C., 2021. Comparison of phenolic and flavonoid compound profiles and antioxidant and α -glucosidase inhibition properties of cultivated soybean (*Glycine max*) and Wild Soybean (*Glycine soja*). *Plants* 10, 1-14. <https://doi.org/10.3390/plants10040813>
- Rafeian-Kopaei, M., Ghasemi-Dehnoo, M., Amini-Khoei, H., Lorigooini, Z., 2020. Oxidative stress and antioxidants in diabetes mellitus. *Asian Pac. J. Trop. Med.* 13, 431. <https://doi.org/10.4103/1995-7645.291036>
- Dirir, A., Daou, M., Yousef, A., Yousef, L., 2022. A review of α -glucosidase inhibitors from plants as potential candidates for the treatment of type-2 diabetes. *Phytochem Rev.* 21, 1049-1079. <https://doi.org/10.1007/s11101-021-09773-1>
- Hikmawanti, N.P.E., Wiyati, T., Abdul Muis, M., Nurfaizah, F.A., Septiani, W., 2021. Total flavonoids content of polar extracts of *cayratia trifolia* leaves. *IOP Conf. Ser.: Earth Environ. Sci.* 819, 012056. <https://doi.org/10.1088/1755-1315/819/1/012056>
- Irakli, M., Skendi, A., Bouloumpasi, E., Chatzopoulou, P., Biliaderis, C.G., 2021. LC-MS identification and quantification of phenolic compounds in solid residues from the essential oil industry. *Antioxidants* 10, 2016.
- Kulbat, K., 2016. The role of phenolic compounds in plant resistance. *Biotechnology and Food Sci.* 80: 97-108.
- Lukiati, B., 2014. Determination of antioxidant activity and total phenol content of gendola leaf extract (*Basella rubra* Linn) and binahong leaf (*Anredera cordifolia* Stenais) as herbal drug candidates. In *Proceedings Biology Education Conference: Biology, Science, Environmental, and Learning*, 11, 195-200.
- Mustikasari, K., Santoso, M., Fadzelly Abu Bakar, M., Fatmawati, S., 2024. Antioxidant, α -glucosidase inhibitory, and cytotoxic activities of *Mangifera rufocostata* extract and identification of its compounds by LC-MS/MS analysis. *Arabian J. Chem.* 17, 105391. <https://doi.org/10.1016/j.arabj.2023.105391>
- Nelly, J.N., Achille, N.B., Claude, N.J., Junior, A., Magloire, B.S., Williferd, T.T.D., Christopher, M., K., J., Prisca, N.O.H., Bernouilli, M.T., Herve, D.O., 2023. Chemical constituents of the methanolic extracts of *Ceiba pentandra* (L.) Gaertn and the antioxidant activities of its aqueous, ethanolic and hydro-ethanolic extracts. *J. Pharmacognosy Phytochemistry*, 12, 103-118.
- Nipun, T., Khatib, A., Ibrahim, Z., Ahmed, Q., Redzwan, I., Saiman, M., Supandi, F., Primaharinastiti, R., El-Seedi, H., 2020. Characterization of α -glucosidase inhibitors from *psychotria malayana* jack leaves extract using LC-MS-Based multivariate data analysis and in-silico molecular docking. *Molecules* 25, 5885. <https://doi.org/10.3390/molecules25245885>
- Nipun, T., Khatib, A., Ahmed, Q., Nasir, M., Supandi, F., Taher, M., Saiman, M., 2021. Preliminary phytochemical screening, in vitro antidiabetic, antioxidant activities, and toxicity of leaf extracts of *psychotria malayana* jack. *Plants (Basel)* 10, 2688. <https://doi.org/10.3390/plants10122688>
- Nokhala, A., Siddiqui, M., Ahmed, Q., Ahamad Bustamam, M., Zakaria, A., 2020. Investigation of α -glucosidase inhibitory metabolites from *tetracera scandens* leaves by GC-MS metabolite profiling and docking studies. *Biomolecules* 10, 287. <https://doi.org/10.3390/biom10020287>
- Olugbami, J.O., Gbadegsin, M.A., Oduola, O.A., 2014. In vitro evaluation of the antioxidant potential, phenolic, and flavonoid contents of the stem bark ethanol extract of *Anogeissus leiocarpus*. *African J. Medicine Med. Sci.* 43, 101-109.
- Orabi, M., Abouelela, M., Darwish, F., Abdelkader, M., Elsadek, B., Al Awadh, A., Alshahrani, M., Alhasanah, A., Aldabaan, N., Abdelhamid, R., 2024. *Ceiba pentandra* ethyl acetate extract improves doxorubicin antitumor outcomes against chemically induced liver cancer in rat model: A study supported by UHPLC-Q-TOF-MS/MS identification of the bioactive phytochemicals. *Front Pharmacol* 15, 1337910. <https://doi.org/10.3389/fphar.2024.1337910>
- Perumal, V., Khatib, A., Uddin Ahmed, Q., Pathamah Uzir, B., Abas, F., Murugesu, S., Zuwairi Saiman, M., Primaharinastiti, R., El-Seedi, H., 2021. Antioxidants profile of *Momordica charantia* fruit extract analyzed using LC-MS-QTOF-based metabolomics. *Food Chem.: Molecular Sci.* 2, 100012. <https://doi.org/10.1016/j.fochms.2021.100012>
- Raduan, S.Z., Ahmed, Q.U., Kasmuri, A.R., Ahmad Rusmili, M.R., Mia, M.A.R., Wan Sulaiman, W.M.A., Mahmood, M.H., Shaikh, M.F., 2022. Antioxidant capabilities of *Litsea garciae* bark extracts and their relation to the phytochemical compositions. *MABJ* 51, 99-118.
- Rocha, S., Amaro, A., Ferreira-Junior, M.D., Proença, C., Silva, A.M.S., Costa, V.M., Oliveira, S., Fonseca, D.A., Silva, S., Corvo, M.Lisa., Freitas, M., Matafome, P., Fernandes, E., 2024. Melanoxetin: A hydroxylated flavonoid attenuates oxidative stress and modulates insulin resistance and glycation pathways in an animal model of type 2 diabetes mellitus. *Pharmaceutics* 16, 261.
- Roheem, F.O., Ahmed, Q.U., Mat So'ad, S.Z., Shah, S.A.A., Latip, J., Alhassan, A.M., Syed Mohammad, S.N.A., 2020. Assessment of free radical scavenging and digestive enzyme inhibitory activities of extract, fractions and isolated compounds from *Tetracera macrophylla* leaves. *J. Herbal Med.* 22, 100351. <https://doi.org/10.1016/j.hermed.2020.100351>
- Saleh, M.S., Siddiqui, M.J., Mediani, A., Ismail, N.H., Ahmed, Q.U., So'ad, S.Z.M., Saidi-Besbes, S., 2018. *Salacca zalacca*: A short review of the palm botany, pharmacological uses and phytochemistry. *Asian Pacific J. Tropical Med.* 11, 645-652.
- Sarian, M.N., Ahmed, Q.U., Mat So'ad, S.Z., Alhassan, A.M., Murugesu, S., Perumal, V., Syed Mohamad, S.N.A., Khatib, A., Latip, J., 2017. Antioxidant and antidiabetic effects of flavonoids: A structure-activity relationship based study. *BioMed Res. Int.* 2017, 1-14. <https://doi.org/10.1155/2017/8386065>
- Satyaprakash, R., Rajesh, M., Bhanumathy, M., Harish, M., Shivananda, T., Shivaprasad, H., Sushma, G., 2013. Hypoglycemic and antihyperglycemic effect of *Ceiba pentandra* L. Gaertn in normal and streptozotocin-induced diabetic rats. *Ghana Med. J.* 47, 121-127.
- Shamsudin, N., Ahmed, Q., Mahmood, S., Shah, S., Sarian, M., Khattak, M., Khatib, A., Sabere, A., Yusoff, Y., Latip, J., 2022. Flavonoids as antidiabetic and anti-inflammatory agents: A review on structural activity relationship-based studies and meta-analysis. *Int. J. Mol. Sci.* 23, 12605. <https://doi.org/10.3390/ijms232012605>
- Sharifi-Rad, M., Anil Kumar, N., Zucca, P., Varoni, E., Dini, L., Panzarini, E., Rajkovic, J., Tsouh Fokou, P., Azzini, E., Peluso, I., Prakash Mishra, A., Nigam, M., El Rayess, Y., Beyrouthy, M., Polito, L., Iriti, M., Martins, N., Martorell, M., Docea, A., Setzer, W., Calina, D., Cho, W., Sharifi-Rad, J., 2020. Lifestyle, oxidative stress, and antioxidants: Back and forth in the pathophysiology of chronic diseases. *Front Physiol.* 11, 694. <https://doi.org/10.3389/fphys.2020.00694>
- Sharma, A., Behera, B., 2022. A review on upodika (*Basella rubra* Linn.) - An ayurvedic nutraceutical with enormous medicinal value. *World J. Pharmaceutical Res.* 11, 237-259.
- Syihabudin, V., Sari, B.L., Utami, N.F., Apriliani, N.A., 2018. Inhibitory activity of α -glucosidase of bark of *Ceiba pentandra* Linn. *Indones. J. Pharm.*, 29, 206-213.
- Pan, H., Peto, R., Hena-Restrepo, A., Preziosi, M., Sathiyamoorthy, V., Abdool Karim, Q., Alejandria, M., Hernández García, C., Kiény, M., Malekzadeh, R., Murthy, S., Reddy, K., Roses Periago, M., Abi Hanna, P., Ader, F., Al-Bader, A., Alhasawi, A., Allum, E., Alotaibi, A., Alvarez-Moreno, C., Appadoo, S., Asiri, A., Aukrust, P., Barratt-Due, A., Bellani, S., Branca, M., Cappel-Porter, H., Cerrato, N., Chow, T., Compo, N., Eustace, J., García, P., Godbole, S., Gotzue, E., Griskevicius, L., Hamra, R., Hassan, M., Hassany, M., Hutton, D., Irmansyah, I., Jancoriene, L., Kirwan, J., Kumar, S., Lennon, P., Lopardo, G., Lydon, P., Magrini, N., Maguire, T., Manevska, S., Manuel, O., McGinty, S., Medina, M., Mesa Rubio, M., Miranda-Montoya, M., Nel, J., Nunes, E., Perola, M., Portolés, A., Rasmin, M., Raza, A., Rees, H., Reges, P., Rogers, C., Salami, K., Salvadori, M., Sinani, N., Sterne, J., Stevanovikj, M., Tacconelli, E., Tikkinen, K., Trelle, S., Zaid, H., Rottingen, J., Swaminathan, S., 2021. Repurposed antiviral drugs for COVID-19 - Interim WHO solidarity trial results. *N. Engl. J. Med.* 384, 497-511. <https://doi.org/10.1056/NEJMoa2023184>
- Tasnuva, S., Qamar, U., Ghafoor, K., Sahena, F., Jahurul, M., Rukshana, A., Juliana, M., Al-Juhaimi, F., Jalifah, L., Jalal, K., Ali, M., Zaidul, I., 2019. α -glucosidase inhibitors isolated from *Mimosa pudica* L. *Nat. Prod. Res.* 33, 1495-1499. <https://doi.org/10.1080/14786419.2017.1419224>
- Thavamani, B.S., Subburaj, V., 2017. In vitro studies on *Basella rubra* different extracts as inhibitors of key enzymes linked to diabetes mellitus. *Pharmacognosy J* 9, 107-111.
- Tijjani, H., Olatunde, A., Adegunloye, A.P., Ishola, A.A., 2022. In silico insight into the interaction of 4-aminoquinolines with selected SARS-CoV-2 structural and nonstructural proteins. In *Egbuna, C. (eds), Coronavirus drug discovery* (Elsevier), 313-333. <https://doi.org/10.1016/B978-0-323-95578-2.00001-7>
- Vita, A.D., Fatimah, N., Murtiwi, S., 2019. Antidiabetic effects of gendola leaf extract (*Basella rubra* L.) on blood sugar levels of white rats. *J. Veterinary Med.* 2, 127-132.
- Zhang, Z., Yao, S., Hu, X., Okejiri, F., He, K., Liu, P., Tian, Z., Dravid, V., Fu, J., Zhu, X., Dai, S., 2020. Sacrificial synthesis of supported ru single atoms and clusters on n-doped carbon derived from covalent triazine frameworks: A charge modulation approach. *Adv. Sci. (Weinh)* 8, 2001493. <https://doi.org/10.1002/adv.202001493>
- Zhouyao, H., Malunga, L., Chu, Y., Eck, P., Ames, N., Thandapilly, S., 2022. The inhibition of intestinal glucose absorption by oat-derived avenanthramides. *J. Food Biochem.* 46, e14324. <https://doi.org/10.1111/jfbc.14324>