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Original article

# Potential biomonitoring of the environmental contamination using snails as sentinel organism: A case study from the Manzala Lagoon, Egypt

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## ABSTRACT

For the first time, historical and recent records on fascioliasis in Egypt reveal the results of sympatric and allopatric fluke/snail interactions in the Manzala Lagoon area by exploring the susceptibility of the species *Pseudosuccinea columella* after exposure to the *Fasciola hepatica* isolates from Egypt. Many factors influence trace metal accumulation in *Pseudosuccinea columella*, including temperature and salinity. Meanwhile, other factors are important in decreasing the bioaccumulation of heavy metals as low pH, low temperature, and high organic content of the substrate. It is indicated, from this study, that the Fe and Cu contamination can be detected in *Pseudosuccinea columella* species. The abundance of these snails varied all over the seasons where the minimum value was recorded in summer while reaching the maximum in autumn. Moreover, the abundance increased with increasing rainfall and humidity during the winter season.

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## 1. Introduction

Several factors such as demographic, socioeconomic, environmental, geographic, climatic, and meteorological factors affect the health of individuals and populations. In fact, all these factors crucially affect public health (Guerreiro et al., 2014; Ebi et al., 2017). Snails are excellent biomonitors of environmental contamination because of their easy sampling, areal distribution, high tolerance to stress, and ability to accumulate varied contaminants (Soltani, 2013). Furthermore, gastropods are essential taxonomic corporations capable of biomonitoring heavy metal pollutants. This case employed *Pseudosuccinea columella* as biomonitors of heavy metal pollution. Biomonitors which are organisms (groups of

organisms) are essential precursors for qualitative and quantitative environmental pollutants (Fränzle, 2006).

In the Manzala Lagoon, increasing industrialization and agricultural activities contribute to the increase in the discharge of chemical pollution into the ecosystem. This leads to increased metal levels in natural waters that damage marine habitats (El-Nemr, 2012). Trace metals naturally exist in the Earth's crust, where they are essential for biological systems considering that they participate in numerous enzymatic processes (Pellerin and Amiand, 2009). However, cadmium, lead, and mercury typically bioaccumulate in tissues and cause intoxication, decreased fertility, and tissue damage (Nordberg, 2010; Stankovic and Jovic, 2012).

Observing the compatibility between different parasite and snail populations may thus help in better understanding and predicting the transmission of fascioliasis. The present resistant and sensitive *Pseudosuccinea columella* is due to the intermediate host snails of *Fasciola hepatica* in the Manzala Lagoon of Egypt (Fig. 1a). This study aims to evaluate the feasibility of applying certain disburbed snails to minimize the concentrations of contaminants in the ecosystem. The concept is that snails feeding on species directly exposed to dry and wet pollutant deposition and inhaling major water pollutants can be critical biomonitoring sentinels for determining the spatial distribution and bioavailability of

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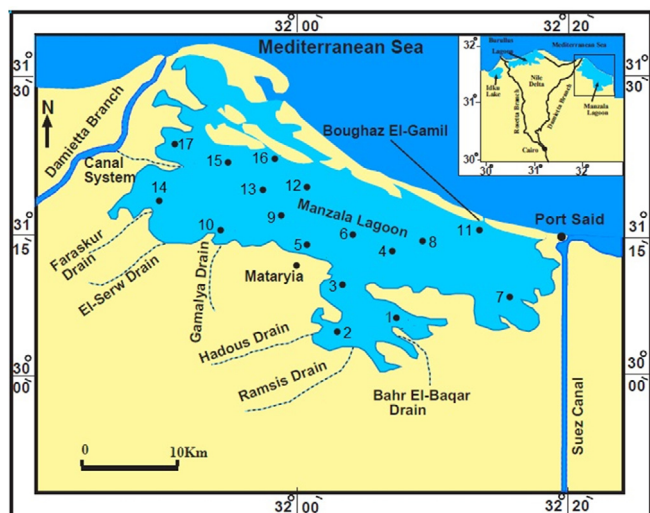


Fig. 1a. Location map.

pollutants and heavy metals. In this regard, *Pseudosuccinea columella* species amassed from the Manzala Lagoon can be used to determine the extreme toxicity and observe the bioconcentration of trace metals. These results could provide insights into the actual players in the host-parasite interactions in the Manzala Lagoon by exploring snail-parasite combinations where they are most significantly present. The spatial distribution of Lymnaeidae species in the Manzala Lagoon is beneficial to better understanding the fascioliasis distribution and delineating priority areas for control interventions. Therefore, this paper aims to provide such information on *Pseudosuccinea columella* as baseline data and verify the potential of these snails as excellent biomonitors.

## 2. Study area

The Manzala Lagoon (Fig. 1a) covers almost 3 % of the area of Egypt and is positioned close to the Mediterranean coast of the Delta region. It serves as the primary assembly factor of numerous polluted water resources and is considered the most important natural resource for fish manufacturing in Egypt. It presents >7 % of the country's total fish production (GAFRD 2006). Since the mid-19th century, the volume of water flowing seaward in the Damietta branch of the Nile, which forms the lagoon's western border, has decreased markedly. This stems from the artificially accelerated channelization of this Nile branch and the diversion of the Nile water into an increasing number of complicated agricultural canal systems (Stanley, 1996). The maximum extents of the rectangular lagoon are 49 km (NW–SE) by 29 km (SW–NE). The annual water temperature ranges from 15 °C to 34 °C (Table 1) while the salinity is from 0.68 g l<sup>-1</sup> in winter to > 3.0 g l<sup>-1</sup> in summer (El-Hehyawi, 1977). The lagoon is separated from the Mediterranean Sea by a chain of broad, elongated, coalescing, and low-elevation sand ridges.

## 3. Materials and methods

### 3.1. Fieldwork

The current bottom sediment samples in the Manzala Lagoon were amassed seasonally from 17 selected stations along with profiles chosen including the entire area (Fig. 1a). The sampling began in the winter of 2018 to the spring, summer, and autumn of 2019. The bottom sediments were obtained from the lagoon using a grab

sample (Ekman type) immersed to a depth ranging from 85 to 180 cm. The hydrogen ion concentrations (pH) of the selected samples were measured immediately (Table 3) using a movable pH sensor (HANNA model, HI 9125). While the mercury thermometer was used for measuring the bottom water temperature which should be lower than 5 °C. Table 2 displays the rainfall, organic matter, and sediment types in the Manzala Lagoon for the four seasons.

### 3.2. Determination of sediment organic carbon content

The organic carbon content was assessed using the acid/dichromate titration approach designated by Gaudette et al. (1974). The results of the analysis were obtained using the equation:

$$\text{Organic Carbon \%} = 10(1 - T/S)[1.0N(0.003)(100/W)]$$

In the equation, *T* is the sample titration (ml ferrous solution), and *S* is the standardization blank titration (ml ferrous solution). The *T/S* factor cancels out the effect of the ferrous solution normality; the value 0.003 refers to the milliequivalent weight of carbon, 1.0 N is the normality of K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>, and 10 is the volume of K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> in ml. *W* is the weight of the sediment sample in grams. The organic carbon is converted to organic matter by multiplying the organic carbon values by the factor of 1.8.

### 3.3. Snail study

The snails were diagnosed using morphological markers (i.e., shells with short spires, periostracum ornamented with spiral sculptures, and ureter with two distant flexures; Pointier 2008). The snail species of *Pseudosuccinea columella* were mostly gathered only at four stations (14, 15, 16, and 17) in front of the most important drains of Faraskur El-Serw and Gamalya, which have poured into the Manzala Lagoon for the duration of the four seasons (winter of 2018 and spring, summer, and autumn of 2019) (Fig. 1a). The snail samples have been gathered through handpicking. The freshwater gastropods of *Pseudosuccinea columella* are recorded as mid-way hosts for *Fasciola hepatica*. The acquired soft sediments were then analyzed to count the living and dead faunas. Then, approximately 100 g of individual sample was dehydrated in an oven at 80–100 °C for 12 h and finally, soaked in a 10 %–15 % hydrogen peroxide solution for 24 h. The analyzed samples were sieved using a 2- to 125-mm screen mash and then dehydrated again. Mollusks identification was carried out through binocular microscopic (Fig. 1b) and counting them for each station (Table 4).

#### 3.3.1. Classification

Biota.

Kingdom: Animalia.

Phylum: Mollusca; Linnaeus 1758.

Class: Gastropod; Cuvier 1795.

Subclass: Heterobranchia; Gray 1840.

Infraclass: Euthyneura; Spengel 1881.

Superorder: Hygrophila; Férussac 1822.

Superfamily: Lymnaeioidea; Rafinesque 1815.

Family: Lymnaeidae; Rafinesque 1815.

Subfamily: Lymnaeinae; Rafinesque 1815.

Genus: *Pseudosuccinea*; Baker 1908.

Species: *Pseudosuccinea columella*; Say 1817.

#### 3.3.2. Identification, distribution, and habitat

The shells were horny brown, thin, translucent, and finely striated. Their apices were pointed and had 3.5–4 weakly convex whorls with shallow sutures. The last whorls predominated with ovate apertures. Columellate margins were reflected in their upper sections, and lower margins were sharp and straight. The shell

**Table 1**  
Bottom water temperature (°C) and Chloride mg/l in the four seasons of the Manzala Lagoon.

Season St. No.	Winter		Spring		Summer		Autumn	
	Temp. °C	Chloride mg/l	Temp. °C	Chloride mg/l	Temp. °C	Chloride mg/l	Temp. °C	Chloride mg/l
1	16	1773	24	1560	32	1631	23	1564
2	15	1595	24	709	31	425	23	957
3	19	1418	24	957	30	922	24	1241
4	20	886	26	815	33	1702	24	1064
5	17	1241	25	1276	30	709	23	1347
6	17	1064	24	1489	31	1631	25	1595
7	19	3013	25	18345	32	425	26	709
8	20	1773	26	1702	33	922	23	1418
9	15	1050	27	1950	32	638	25	1064
10	18	1064	27	709	31	1134	23	1773
11	19	1078	23	5034	31	851	25	1241
12	20	1595	23	1489	32	5034	24	4431
13	18	1134	24	1702	32	1230	25	1773
14	17	638	24	1276	34	1134	23	1241
15	15	1950	24	1347	31	1985	23	1418
16	17	851	24	886	32	1134	24	1064
17	17	1985	24	922	31	1276	23	1418

widths ranged from 8 to 13 mm, and the heights ranged from 15 to 20 mm. The shells were found in Lower Egypt (the Nile and its essential canals) in the Damietta and Rosetta branches of the Nile, particularly in the El Qalubyia, El Menoufia, and Damietta regions. The snail species occur in shallow, stagnant water (with an abundance of lily pads) and on rocky beaches, which can protect them from wave action.

#### 3.4. The heavy metals analysis of water samples

Water sampling (200 ml) was processed by 5 ml of a diacid mixture ( $\text{HNO}_3\text{:HClO}_4$ , 9:4 ratio) on a hot plate and sorted through the Whatman No. 42 papers. Double distilled water was then added to increase the volume to 50 ml. The heavy metals in the water samples were analyzed following the method given by the Association of Official Analytical Chemistry (Association of Official Analytical Chemistry (AOAC), 1995) using an atomic absorption spectrophotometer (SolarM600531 v1.27).

#### 3.5. Analysis of heavy metals in sediment samples

Portions of 1 g from the dehydrated sample were weighed and moved into a hot block digester. Further, they were treated with  $\text{HNO}_3$  solution (Analar grade, BDH 69 %) in two steps; low temperature (40 °C) for 1 h and high temperature (140 °C) for 3 h. Then, diluted with purified deionized water. Finally, the values of Mn, Fe, Cu, Pb, and Zn have been counted through the AASs (Pb through the furnace AAS Model 670G and Cu and Zn by the flame AAS Model 670G). Relevant data were illustrated in parts per million (Table 5).

#### 3.6. Analysis of heavy metals in snail tissues

Live *Pseudosuccinea columella* snails have been acquired to identify the bioconcentrations of trace metals in their soft tissues in every station (Tables 6–9). The soft tissues of *Pseudosuccinea columella* snails have been carefully unglued from their shells and rinsed with distilled water and methanol to minimize contamination.

**Table 2**

Bottom sediment types, organic matter %, and statistical analysis of monthly rainfall over Manzala Lagoon.

Parameter St. No.	Sediment types				Organic Matter %							
	Winter	Spring	Summer	Autumn	Winter	Spring	Summer	Autumn				
1	Sandy silt	Silty sand	Silt	Silt	0.33	0.31	0.27	0.23				
2	Silty sand	Sandy silt	Sandy silt	Sandy silt	0.30	0.28	0.21	0.16				
3	Silty sand	Silty sand	Sandy silt	Sandy silt	0.26	0.20	0.25	0.33				
4	Silty sand	Silt	Silt	Silt	0.21	0.18	0.22	0.31				
5	Sandy silt	Silt	Silt	Sandy silt	0.30	0.30	0.28	0.30				
6	Silty sand	Silty sand	Sandy silt	Sandy silt	0.23	0.34	0.22	0.33				
7	Silty sand	Silty sand	Silt	Silty sand	0.30	0.22	0.35	0.31				
8	Silty sand	Silt	Silty sand	Sandy silt	0.25	0.20	0.22	0.27				
9	Silty sand	Silty sand	Silty sand	Silty sand	0.22	0.34	0.25	0.26				
10	Silty sand	Silty sand	Silty sand	Silty sand	0.27	0.30	0.15	0.22				
11	Silty sand	Silt	Silty sand	Sandy silt	0.41	0.36	0.11	0.35				
12	Silty sand	Sandy silt	Sandy silt	Sandy silt	0.12	0.30	0.27	0.24				
13	Sandy silt	Silt	Silty sand	Silty sand	0.27	0.32	0.16	0.22				
14	Silty sand	Silty sand	Silty sand	Silty sand	0.50	0.29	0.27	0.32				
15	Silty sand	Silty sand	Sandy silt	Silty sand	0.31	0.15	0.22	0.28				
16	Silty sand	Silty sand	Silty sand	Silt	0.23	0.17	0.27	0.17				
17	Silty sand	Silty sand	Silty sand	Silt	0.41	0.20	0.17	0.40				
Average					0.36	0.20	0.23	0.29				
Rainfall												
Rainfall Month	Jan.	Feb.	Mar.	Apr.	May	Jun.	Jul.	Aug.	Sep.	Oct.	Nov.	Dec.
Mean (mm)	17.43	12.98	9.65	4.41	2.34	0.45	0	0	0.75	4.66	9.88	15.91
Average	15.44 Winter		5.47 Spring			0.15 Summer			5.10 Autumn			

tion in mantle fluids and remove any sediment particles. About 0.01 g of the soft tissues of each snail sample was dehydrated at 50 °C, weighed, and then dissolved through 1 ml of concentrated HNO<sub>3</sub> at 70 °C for 2 h. In the end, these samples were diluted in 5 ml of ultrapure deionized water to evaluate their heavy metal contents (Federici et al., 2007).

The concentrations of chemical substances in aquatic organisms may be calculated using two different factors: the bioconcentration factor (BCF) and the bioaccumulation factor (BAF). BCF refers to concentration levels of chemical substances in organisms only due to uptake from the surrounding water, whereas BAF additionally includes uptake from food. In this regard, the bioaccumulation sediment factor was calculated as an organism's average metal-to-average sediment concentration ratio at a given time (Zhao et al.,

2012) to evaluate the performance of metal bioaccumulation in organisms. The BCF of Gobas and Morrison (2000) (Tables 6-9) was calculated as follows:

$$BCF = \text{Concentration of biota} / \text{Concentration of water}$$

The biota sediment accumulation factor (BSAF) was estimated for designated metals in the snails through the following relationship to estimate the proportion of metals in the organism to the associated sediments (Szefer et al., 1999):

$$BSAF = C_x / C_s$$

where C<sub>x</sub> and C<sub>s</sub> are the concentrations of the designated metals in the snail tissues and the accompanying sediments, respectively. The tissues of snails were categorized as macroconcentrators (BSAF > 2),

**Table 3**

Bottom sediment samples depth and hydrogen ion (pH) in different seasons of the Manzala Lagoon.

Parameter St. No.	Depth (cm)				Hydrogen ion (pH)			
	Winter	Spring	Summer	Autumn	Winter	Spring	Summer	Autumn
1	130	150	75	60	7.03	7.03	7.55	7.55
2	100	80	100	100	7.27	7.18	7.65	7.44
3	100	*	130	120	7.8	7.66	7.89	7.8
4	150	100	80	60	8.42	7.71	8.6	7.77
5	90	140	150	90	7.09	7.45	7.55	6.84
6	*	90	180	110	7.75	7.83	7.76	7.72
7	50	50	200	75	7.05	7.2	7.72	7.6
8	70	85	100	65	8.49	8.08	8.43	7.72
9	110	130	150	120	7.61	7.85	8.18	8.81
10	110	110	150	130	7.09	7.4	8.27	8.09
11	120	110	100	100	8.86	7.52	8.35	8.75
12	80	130	150	170	8.22	8.06	8.38	8.73
13	120	140	150	120	8.43	8.8	8.38	8.24
14	110	150	130	90	8.40	8.20	7.82	7.46
15	130	150	150	150	7.76	8.05	8.27	8.33
16	150	158	180	100	7.48	7.50	8.60	7.52
17	100	170	120	120	7.92	7.56	7.80	7.66

microconcentrators ( $1 < \text{BSAF} < 2$ ), or deconcentrators ( $\text{BSAF} < 1$ ), following Dallinger (1993). The deposition of mollusk shells has been illustrated in Fig. 1b.

## 4. Results

### 4.1. Life and dead snails

Table 4 displays a comparison between the density of living mollusks and dead fauna (Table 4). This reveals the quantity of living *Pseudosuccinea columella* snails found only in the four samples from the west and northwest regions (Samples 14–17). The quantity of living *Pseudosuccinea columella* snails from the surface bottom deposits indicates the ecological explanations and inspection of relations amongst species and physicochemical environmental variables. In Egypt, *Pseudosuccinea columella* became one of the most widely distributed freshwater snail species. It is also an efficient midway host of *Fasciola hepatica* (Coelho and Lima, 2003).

This could increase the risk of the emergence of fascioliasis in another area and its occurrence in endemic areas due to the presence of greater numbers of susceptible hosts.

### 4.2. Trace metals in water samples

The Cu, Pb, Fe, Mn, and Zn (ppm) concentrations in the various samples of water in all seasons vary between 4.10–8.45, 5.21–7.67, 4.30–7.54, 3.97–5.98, and 0.28–0.34, respectively. The maximum trace metal levels were recorded in front of the El-Serw drain, where the values of trace metals in both freshwater and sediment samples enhanced significantly through the high effluent discharge level into this drain from the industrial region of Damietta.

### 4.3. Trace metals in bottom sediments

The percentages of all heavy metals were higher in the bottom sediment than in the water. This revealed that the metals precipi-





Fig. 1b. Snail shell samples were examined by the binocular microscope.

tated in the sediment instead of dissolving in water (Table 5), where sediments are important sinks for numerous pollutants such as pesticides and heavy metals. Also, sediments play an extensive function in remobilizing contaminants in aquatic systems under favorable situations and in water–sediment interactions (Hyun et al., 2007).

**Iron:** In summer, the minimum average concentration value of iron at the four stations was 4.80 mg/g. It increased slowly in winter (4.83 mg/g) and reached 5.03 mg/g in spring and 5.42 mg/g in autumn. The maximum annual average concentration values of iron (6.21 mg/g) were recorded at Station 16 during autumn. **Zinc:** The maximum average concentration value of zinc was recorded (36.3 µg/g) in autumn. It decreased gradually in winter and summer (33.75 and 33 µg/g, respectively) and reached its minimum average value in spring (23 µg/g). The maximum annual average concentration values of zinc (60 µg/g) were recorded at Station 16 in autumn. **Copper:** The highest concentration value for copper was monitored in summer (96 µg/g). In contrast, the lowest value was recorded in spring (76.3 µg/g). It decreased gradually in autumn and winter (86.8 and 84 µg/g, respectively). The maximum annual average values were recorded at Station 17 (163 µg/g) in summer. The most concentrated trace metal recorded were Fe and Zn, whereas the least was Cu. These findings agree with the previous observations of Abd El-Azim et al. (2018) in Lake Timsah.

**Manganese:** The highest average concentration value of manganese (35.1 µg/g) was reported in winter while the lowest was in summer (13.75 µg/g). It decreased slowly in autumn and spring (25.08 and 21.0 µg/g, respectively) till reached the minimum in summer (13.75 µg/g). The extreme annual average concentration value of manganese was recorded at Station 14 (58.0 µg/g), and the lowest value was 18.5 µg/g at Station 17 in summer. **Lead:** The highest average concentration values of lead were recorded in winter and summer and reached 15.5 and 14.8 µg/g, respectively. These gradually decreased in autumn and spring (6.3 and 4.3 µg/g, respectively). The lowest average value reached 4.3 µg/g in spring. The maximum annual average concentration value of lead was recorded at Station 17 (39 µg/g). Lead is a toxic metal originating from natural processes and anthropocentric activities (coal power plants, mining, waste gas fuel, leather whipping, paint, and battery factories). It is detrimental to the environment and human health.

#### 4.4. Bsaf

The BSAF in winter, spring, summer, and autumn in the area under investigation revealed that all the soft tissues of the *Pseudo-*

*succinea columella* snails were macroconcentrators (BSAF > 2) (Tables 6–9) except for Mn (where they were deconcentrators in summer and autumn; BSAF < 1) and Fe (where they were deconcentrators in autumn; BSAF < 1). These show that selective tissues are thus good biomonitors of pollution rates. The minimum BSAF values for Cu in the soft tissues varied from 3.1 to 4.55 in winter. These increased to 3.30–4.55 in spring. The maximum values for Cu were recorded in summer (7.60–9.62). The minimal BSAF values for Zn in the soft tissues varied from 2.20 to 3.90 in autumn. These increased to maximum values during summer (4.11–5.72). The maximum BSAF values for Pb in the soft tissues ranged from 3.11 to 4.41 in spring, whereas the minimum values (2.10–3.30) were recorded in autumn.

The minimum BSAF values for Fe in the soft tissues ranged from 1.80 to 3.00 in autumn. The maximum values ranged from 4.28 to 7.13 in summer. The minimum BSAF values for Mn in the soft tissues ranged from 1.70 to 2.40 in summer. The maximum values ranged from 3.12 to 4.28 in winter. Laszczyc et al. (2004) illustrated that enzymatic activities are responses to environmental stress. Thus, they can be used to estimate pollution at population and ecosystem levels. Meanwhile, El-Khayat et al. (2015) used physiological and biochemical parameters as indicators of histopathological changes. These parameters have also been widely used as biomarkers in the health evaluation of animals.

#### 4.5. Bcf

The BCF confirmed that the highest concentrations (ppm) of Cu ranged from 13.21 to 32.56 in autumn. Those of Fe ranged from 119.78 to 895.12 in summer, and those of Mn ranged from 10.22 to 62.16 in autumn. The highest concentrations of Zn ranged from 21.22 to 85.21 in autumn, and those of Pb ranged from 1.16 to 3.86 in summer (Tables 6–9). The BCF values obtained for Cu and Pb were >1.00, indicating that both were highly bioaccumulated and biomagnified in the tissues of *Pseudosuccinea columella* snails (Cu has higher BCF than that of Pb). Copper is essential in some enzymes for organisms (Yap et al. 2016) and serves as a micronutrient for cellular metabolism (Ismarti et al. 2017). The cuproprotein hemocyanin (Cheung and Wong, 1992) functions similarly to Fe in the metabolism of molecular oxygen in snails because the oxygen-carrying pigment in mollusks' blood is not hemoglobin. This study confirmed the high variations in BCF limits depending on the metal type (Tables 6–9). This is due to the tolerance effect of snails to the persistent toxic chemicals and can be considered effective biomonitors (Rehman et al., 2016). The concentration factors (BCF) exceeding 1 indicated that the freshwater snail *Pseudosuccinea columella* may be used to clean up the trace metal contamination caused by human activity.

## 5. Discussion

### 5.1. Ecological factors affecting the abundance of *P. Columella*

#### 5.1.1. pH parameter

The canonical correspondence analysis performed using the ecological data recorded in the north and northwest sites of the Manzala Lagoon confirmed the relation of the abundance of *Pseudosuccinea columella* with pH variations. We observed a strong negative association between snail abundance and pH, implying the snails' certain tolerance for slightly acid waters. The average pH values were rounded to 8.15 (7.48–8.60). Globally, snails can tolerate environments with pH values ranging from 7 to 8. In this study, the number of snails decreased in summer (Samples 14, 16, and 17), which is characterized by relatively low pH (Tables 3 and 4).

**Table 4**Dead and alive numbers of *Pseudosuccinea columella* and the total relative densities of snail species found in 100 g in the different seasons of the Manzala Lagoon.

Season Station No.	Dead and alive numbers found in 100g of <i>Pseudosuccinea columella</i>								Total relative densities found in 100g of <i>Pseudosuccinea columella</i>			
	Winter		Spring		Summer		Autumn		Winter	Spring	Summer	Autumn
	Dead	Alive	Dead	Alive	Dead	Alive	Dead	Alive				
1	-	-	-	-	-	-	-	-	-	-	-	-
2	-	-	-	-	-	-	-	-	-	-	-	-
3	-	-	-	-	-	-	-	-	-	-	-	-
4	-	-	-	-	-	-	-	-	-	-	-	-
5	-	-	-	-	-	-	-	-	-	-	-	-
6	-	-	-	-	-	-	-	-	-	-	-	-
7	-	-	-	-	-	-	-	-	-	-	-	-
8	-	-	-	-	-	-	-	-	-	-	-	-
9	-	-	-	-	-	-	-	-	-	-	-	-
10	-	-	-	-	-	-	-	-	-	-	-	-
11	-	-	-	-	-	-	-	-	-	-	-	-
12	-	-	-	-	-	-	-	-	-	-	-	-
13	-	-	-	-	-	-	-	-	-	-	-	-
14	25	9	15	7	8	4	20	10	34	22	12	30
15	20	8	11	5	7	4	14	8	28	16	11	22
16	22	10	13	17	8	5	14	8	32	30	13	22
17	21	16	20	8	14	6	22	8	37	28	20	30
Average	88	43	59	37	37	19	70	34	131	96	56	104

This agrees with the findings of [Gutiérrez \(2004\)](#) and [Prepelitchi et al. \(2011\)](#).

#### 5.1.2. Rainfall parameter

The abundances of snails were low in summer, increased in autumn, peaked in winter, and decreased in spring ([Table 4](#)); i.e., the abundance increased with increasing rainfall. Another factor explaining the population boom of *Pseudosuccinea columella* in winter would be the almost complete disappearance of *Biomphalaria* spp. ([Prepelitchi et al., 2011](#)). The soft tissues of *Pseudosuccinea columella* could be used as biomonitors because of their

favorable aquatic behavior, tolerance to wide temperature ranges, pH, salinity, and a relatively high intrinsic rate of natural increase ([DeKock et al., 1989](#)).

#### 5.1.3. Organic matter parameter

We determined the organic matter content, including *Pseudosuccinea columella* snails, in the samples in the four seasons. The sample amassed in winter ranged from 0.23 % and 0.50 %, with an average of 0.36 %. In autumn, the organic matter content varied from 0.17 % to 0.40 %, with an average of 0.29 %. In summer, it ranged from 0.17 % to 0.27 %, with an average of 0.23 %. In spring, it

**Table 5**

Chemical parameters of bottom sediments in Manzala Lagoon (different seasons).

Parameter Stations	Mn (ug/g)	Fe (mg/g)	Pb (ug/g)	Zn (ug/g)	Cu (ug/g)
14	13.0	4.11	7	30	72
15	11.5	4.55	1	25	69
16	12.0	4.99	12	27	80
17	18.5	5.66	39	50	163
Average summer	13.75	4.8	14.8	33	96
14	24.0	5.40	6	24	80
15	23.0	5.14	5	19	89
16	24.5	4.28	4	22	60
17	12.5	5.28	2	27	76
Average spring	21.0	5.03	4.3	23	76.3
14	36.0	5.10	10	30	90
15	23.0	4.83	4	22	61
16	22.4	6.21	4	60	104
17	18.9	5.52	7	33	92
Average autumn	25.08	5.42	6.3	36.3	86.8
14	58.0	3.80	20	26	50
15	29.6	4.99	17	39	92
16	32.4	5.34	18	39	102
17	20.5	5.17	7	31	92
Average winter	35.13	4.83	15.5	33.75	84

**Table 6**

Shows the tissue/sediment (BSAF) and tissue/water (BCF) during the winter season of the Manzala Lagoon (parts per million, ppm).

Parameter Stations	Mn Tissue/sediment (BSAF)	Iron Tissue/sediment (BSAF)	Lead Tissue/sediment (BSAF)	Zinc Tissue/sediment (BSAF)	Copper Tissue/sediment (BSAF)
14	4.28	3.66	4.18	3.57	4.55
15	3.70	2.96	3.76	3.12	4.00
16	3.12	2.60	3.12	2.30	3.10
17	4.20	3.50	4.00	3.45	4.40
Average	3.83	3.18	3.77	3.11	4.01
Parameter St. No.	Mn Tissue/water (BCF)	Iron Tissue/water (BCF)	Lead Tissue/water (BCF)	Zinc Tissue/water (BCF)	Copper Tissue/water (BCF)
14	56.22	409.79	2.32	59.03	21.76
15	19.70	296.67	1.86	38.50	16.03
16	16.00	128.62	1.10	20.32	13.12
17	29.05	840.47	1.98	70.88	26.65
Average	30.24	418.96	1.82	47.18	19.39

decreased from 0.15 % to 0.29 %, with an average of 0.20 % (Table 2). The area was characterized by low organic matter contents in spring and relatively high content in winter. This explained the

relative densities of snail species found in particularly excessively organic contents in Samples 14, 15, and 17 in winter (0.50 %, 0.31 %, and 0.41 %, respectively). Similarly, Samples 14 and 17 in



**Table 7**

Shows the tissue/sediment (BSAF) and tissue/water (BCF) during the spring season of the Manzala Lagoon (parts per million, ppm).

Parameter	Mn Tissue/sediment (BSAF)	Iron Tissue/sediment (BSAF)	Lead Tissue/sediment (BSAF)	Zinc Tissue/sediment (BSAF)	Copper Tissue/sediment (BSAF)
Stations					
14	3.82	6.24	4.41	3.57	4.55
15	2.70	5.10	3.85	2.78	3.60
16	2.60	4.28	3.11	2.50	3.30
17	3.70	6.20	4.20	3.40	4.45
Average	3.21	5.46	3.89	3.06	3.98
Parameter	Mn Tissue/water (BCF)	Iron Tissue/water (BCF)	Lead Tissue/water (BCF)	Zinc Tissue/water (BCF)	Copper Tissue/water (BCF)
St. No.					
14	55.18	400.11	2.16	58.14	20.12
15	19.70	241.13	1.56	35.91	14.88
16	16.00	112.16	1.02	19.77	12.86
17	29.05	780.55	1.67	69.12	25.13
Average	29.98	383.46	1.60	50.24	18.25

**Table 8**

Shows the tissue/sediment (BSAF) and tissue/water (BCF) during the summer season of the Manzala Lagoon (parts per million, ppm).

Parameter	Mn Tissue/sediment (BSAF)	Iron Tissue/sediment (BSAF)	Lead Tissue/sediment (BSAF)	Zinc Tissue/sediment (BSAF)	Copper Tissue/sediment (BSAF)
Stations					
14	2.40	7.13	3.70	5.72	9.62
15	2.00	6.14	2.92	4.80	8.40
16	1.70	4.28	2.73	4.11	7.60
17	2.34	7.00	3.65	5.60	9.50
Average	2.11	6.14	3.25	5.06	8.78
Parameter	Mn Tissue/water (BCF)	Iron Tissue/water (BCF)	Lead Tissue/water (BCF)	Zinc Tissue/water (BCF)	Copper Tissue/water (BCF)
St. No.					
14	61.00	511.67	3.86	71.70	26.78
15	20.13	300.11	2.11	38.50	18.71
16	12.89	119.78	1.16	20.32	11.99
17	34.12	895.12	2.74	84.35	30.12
Average	32.04	456.67	2.47	53.72	21.90

autumn were characterized by relatively high numbers of snails because of their relatively high organic matter contents (0.32 % and 0.40 %, respectively). These findings agree with previous observations recorded by Sami et al. (2020) in their study on the bivalves of Lake Timsah (Egypt).

## 5.2. Factors controlling the distribution of trace elements

### 5.2.1. Metabolic parameters

The presence of excessive levels of Fe, Cu, and Pb in comparison with the low levels of Zn and Mn in the soft tissues of *Pseudosuccinea columella* is attributed to the roles of these metals as constituents of metabolically important biomolecules according to Langston et al. (1998). Notably, the soft tissues of the *Pseudosuc-*

*cinea columella* snails could store Cu and Zn because their concentrations exceeded the environmental concentration and were attributable to bioaccumulation (Gundacker, 2000).

Marine gastropods typically collect and store Cu and use it to synthesize hemocyanin, a blood pigment. Thus, the Cu concentrations in the soft tissues of *Pseudosuccinea columella* may be attributed to Cu in hemocyanin (Dallinger and Wieser, 1984). So, the Cu concentration was high in the soft tissues since these were nurtured with blood. The soft tissues of the *Pseudosuccinea columella* snails can thus be used as biomonitoring agents for Cu and Fe contamination in water samples (Cheung and Wong, 1992). Furthermore, *Pseudosuccinea columella* is a beneficial bioindicator of trace metal pollution in aquatic ecosystems.

**Table 9**

Shows the tissue/sediment (BSAF) and tissue/water (BCF) during the autumn season of the Manzala Lagoon (parts per million, ppm).

Parameter	Mn Tissue/sediment (BSAF)	Iron Tissue/sediment (BSAF)	Lead Tissue/sediment (BSAF)	Zinc Tissue/sediment (BSAF)	Copper Tissue/sediment (BSAF)
Stations					
14	2.64	3.00	3.30	3.90	6.44
15	2.00	2.71	2.93	2.90	5.36
16	1.78	1.80	2.10	2.20	3.49
17	2.60	2.92	3.11	3.71	6.40
Average	2.26	2.61	2.86	3.18	5.42
Parameter	Mn Tissue/water (BCF)	Iron Tissue/water (BCF)	Lead Tissue/water (BCF)	Zinc Tissue/water (BCF)	Copper Tissue/water (BCF)
St. No.					
14	62.16	515.12	3.16	70.88	29.20
15	21.69	326.17	2.58	39.11	17.44
16	10.22	120.13	1.16	21.22	13.21
17	35.28	891.59	2.74	85.21	32.56
Average	32.34	463.25	2.41	54.11	23.10

### 5.2.2. Physicochemical parameter

Factors critical in reducing the bioaccumulation of heavy metals are low pH, low temperature, and excessive organic content of the substrate. The temperature and pH effects can be explained by increased stress at decreased temperature and pH, resulting in reduced food intake (Elder and Collins, 1991). The pH parameter is vital in the immobilization of Pb by microbes in Pb-contaminated environments, affecting the microbial quantity and enzyme activity. It is suggested that biomineralization is swift and that the crystal structures of its products are proper when the pH was 8–9. Similarly, temperature is an important factor influencing the growth rate and metabolic activity level of microorganisms. The activity of enzymes, including urease and phosphatase, varies with temperature, resulting in changes in the biomineralization ability.

## 6. Conclusions

Human activities are principal factors responsible for freshwater mollusks' invasion of aquatic environments, and this phenomenon has increased in the last decades. Thus, the presence of *Pseudosuccinea columella* in Egypt in the wild is not surprising. The average concentration of Zn was higher in autumn in sediment (36.3 µg/g) than those in the water and tissues. Meanwhile, the average concentration of Pb was higher in winter in sediment (15.5 µg/g) than those in the water and tissues. The average concentration levels of Pb were higher in winter than those in the other seasons. The average concentration level of Fe was higher in autumn (5.42 µg/g) than those in the water and tissues. Nevertheless, the Cu average concentration was higher in summer in sediment (96 µg/g) than those in the water and tissues. This observation showed that the average concentration levels of Cu and Fe in the tissues were higher in winter than those in the other seasons.

This study suggests that *Pseudosuccinea columella* can be a potential bioindicator for metal pollution and toxicity testing. In all seasons, the trace metal concentrations in the samples in all stations were greater than those in the snails' tissues. This snail can be observed in various habitats, such as small streams, canals, and

lakes, but seems to prefer standing running freshwaters. In Egypt, *Pseudosuccinea columella* may be increasing in distribution. Given that it is an efficient intermediate host of *Fasciola hepatica*, this situation could increase the risk of the emergence of fascioliasis in new areas and the prevalence of the disease in endemic ones because of the presence of greater numbers of susceptible hosts.

The BSAF identified that *Pseudosuccinea columella* were macro-concentrators for Fe and Pb, whereas its soft tissues were macro-concentrators for Cu and Zn. These findings indicated that the differences in metal distribution could be attributed to the differences in tissue physiology and metal handling, storage, and detoxification strategies. The BCF results confirmed the ability of *Pseudosuccinea columella* snails to accumulate heavy metals. The snails had a higher BCF category for Cu and Pb in water but a low category for sediment. This revealed that *Pseudosuccinea columella* snails are good accumulators of Cu and Pb in seawater. The abundance of living *Pseudosuccinea columella* snails in the front of the El-Serw drain in winter due to the content of organic matter and increasing rainfall, the high concentrations of chloride and alkalinity owing to sewage, high pH value, and the concentrations of nitrites.

### Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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### Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.jksus.2022.102341>.

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