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Journal of King Saud University – Science

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Original article

Assessment of different salt concentrations on the growth and phytochemical change of the ice plants

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ARTICLE INFO

Article history:

Received 20 October 2021

Revised 6 June 2022

Accepted 8 June 2022

Available online 14 June 2022

Keywords:

Mesembryanthemum crystallinum L.

Photosynthesis

Image processing technique

Salt stress

Secondary metabolites

ABSTRACT

The ice plant (*Mesembryanthemum crystallinum* L.) has become a halophyte model to study the plant photosynthetic responses C₃ photosynthesis to crassulacean acid metabolism (CAM), which is accelerated by salt stress. However, this adaptive mechanism improves water use efficiency, and water stress tolerance is still poorly known. This study examined the effect of individual and mixture of NaCl and CaCl₂ concentrations on morphological parameters and bioactive component contents of ice plants in a plant factory system. Eight salt treatments individually and a combination of sodium chloride (NaCl) and calcium chloride (CaCl₂), and Hoagland solutions were applied after the transplanting of ice plants. Morphological parameters like the number of leaves and lateral stems, leaf chlorophyll content (SPAD value), fresh weight, and dry weight of shoots and roots were measured at the adult stage. Concurrently, in the juvenile phase, the area of a canopy was evaluated using an image processing technique in HSV (hue, saturation, value) colour space. Correspondingly, ice plant secondary metabolites such as cations, anions, and radical scavenging activity were assayed in the adult phase correlated to the salt stress. The effects of salt stress on the growth of ice plants and secondary metabolite production were analysed using completely randomized block designs through the variance by one-way ANOVA with a significance level of $p < 0.05$. This study demonstrated that 400 mM CaCl₂ (T₄) enhanced the biomass and high sodium (Na⁺) and calcium (Ca²⁺) accumulations, and 200 mM CaCl₂ (T₃) accelerated the potassium (K⁺), magnesium (Mg²⁺), phosphate (PO₄³⁻), and sulfate (SO₄²⁻) accumulations. Moreover, NaCl 400 mM (T₁) and combination of 100 mM NaCl and 300 mM CaCl₂ (T₅) positively influenced the chloride (Cl⁻) deposition and combination of 200 mM NaCl and 200 mM CaCl₂ (T₆) improved nitrate (NO₃) accretion. Furthermore, 100 mM CaCl₂ (T₂) exhibited the highest antioxidant activity in ice plants grown under the plant factory system.

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Peer review under responsibility of King Saud University.



<https://doi.org/10.1016/j.jksus.2022.102168>

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1. Introduction

The common ice plant (*Mesembryanthemum crystallinum* L.) has become synonymous with the salt stress response model. It is systematically developed under controlled and salinity environmental conditions. Furthermore, it can encompass the salinity by enhancing the water usage efficiency by switching the photosynthesis mechanism C₃ to CAM (Adams et al., 1998). Moreover, salinity responses result in the compartmentalization of ions and compatible solutes, and physiologically active compounds, as reported by Kim et al., 2021. Correspondingly, these physiologically active substances, presumably phenolic compounds and flavonoids, are

profusely consumed by humans to prevent detrimental health problems (Kim et al., 2021).

Consuming ice plant leaves helps treat diabetes by inhibiting increases in blood sugar levels, as mentioned by Kang et al., 2006. Consequently, consumer interest in common ice plants, effective against diabetes mellitus, has risen recently (Kim et al., 2018). The plant is regarded as the crystalline ice plant because it is surrounded by larger, glistening epidermal bladder cells (EBCs) that resemble ice crystals (Kim et al., 2021). A large central fluid vacuole distinguishes EBCs, which uses the acclimation process to regulate salt retention and water as reservoirs for suitable solutes and phytochemicals like flavonoids, pinitol, and myo-inositol (Agarie et al., 2007; Hasegawa et al., 2000).

In addition to that, myo-inositol has been involved in the cellular process in the plant, such as signal transduction, growth regulations, and membrane response. Correspondingly, myo-inositol plays an intriguing role in plants and animals by breaking down cholesterol and lipids causing panic disorder and disease depression and regulating blood pressure, as reported by Agarie et al., 2009. Salt stress induces the accumulating high concentrations of phytochemicals (D-pinitol, myo-inositol, D-ononitol) in the cytosol. Considering the above facts, evaluating antiradical and radical scavenging properties is vital to determine the basis of the phenolic concentration of ice plant extract. According to Shalaby and Shanab, 2013, the antioxidant activity of plant tissues is typically measured using 2,2-diphenyl-1-picrylhydrazyl (DPPH) to assess radical scavenging or antiradical activity, or the potential of compounds to react with free radicals.

Notably, the ice plant is a facultative halophyte that responds to extreme salinity by switching its metabolism from C₃ to CAM photosynthesis, an ecophysiological adaptation that aids in water conservation by assimilating external CO₂ with minimal water loss by opening its stomata at night, with lower CO₂ compensation points during the diurnal cycle's dark and early light periods and appreciably lower photorespiration rates than C₃ plants (Agarie et al., 2009; Cushman et al 1990). In addition, CAM atmospheric CO₂ is converted to organic acids (e.g., malic acid, citric acid) and stored in mesophyll cells, where phosphoenolpyruvate carboxylase (PEPC) remobilizes and decarboxylates the organic acids to provide CO₂ during the day through the Calvin cycle, as revealed by Guan et al., 2020.

As mentioned above, *M. crystallinum* is a facultative halophyte for the study of C₃ and CAM due to its ability to respond to salt stress. As a result, this plant is highly tolerant of salinity and drought. It completes its life cycle on soil with a NaCl concentration equivalent to seawater (500 mM) (Agarie et al., 2009). Notably, Guan et al., 2020 found that 500 mM NaCl showed a similar leaf growth rate as the control group (0.5 × Hoagland's solution) within the initial 5 days of salt treatments and then growth delayed after 7 days. Considering the carbon fixation, salt-stressed ice plants have performed less carbon fixation compared to C₃ plants.

Correspondingly, NaCl inducible myo-inositol-1-phosphate synthase in roots and sodium absorption and transport through the xylem was coupled to a tenfold increase in myo-inositol and onitol in the xylem. Concurrently, myo-inositol serves as a substrate for producing compatible solutes and a signal for enhancing sodium absorption between leaves and roots (Nelson et al., 1999).

In addition to sodium, calcium is rich in green vegetables, facilitating the calcium deficiency in human nutrients. Considering that calcium deficiency is the most frequent mineral shortfall in modern diets, boosting the calcium content of leafy vegetables in ice plants can help consumers acquire more nutrients (Yuan et al., 2018). In addition to that Ca²⁺ ions increased the membrane integrity and storage capacity (Xu et al., 2013). Further, calcium appears to play a protective function to salt-induced mechanisms, prompting growth regulators and other substances to alleviate the adverse effects and maintain normal features of plants (Xu et al., 2013).

Furthermore, *M. crystallinum* stomatal movement and phenotypic changes are influenced by high salt concentrations in the soil; combining the leaf succulence assay during the transition stage to facultative CAM plants improves leaf succulence and facilitates photosynthesis (Guan et al., 2020). In this regard, analysis of the ion content of ice plants is of utmost importance to confirm previous predictions that EBCs in *M. crystallinum* under salt stress conditions may act as water reservoir organs for vacuolar ion homeostasis and salt retention, as shown by Agarie et al., 2007.

Following changes in salt stress conditions, morphological characteristics may appear on above and below ground parts such as leaves, stems, and roots (Buckley et al., 1997). In general, growth analysis reveals the primary productivity and physiological phenomena of plants.

Generally, mild salt concentration influenced the canopy area development compared to the lower concentration of salt treatments. Interestingly, CaCl₂ has been involved in plant regulatory mechanisms that ensure plants adjust to unfavourable stress conditions and ameliorate the negative effects by altering physiological and biochemical mechanisms in leaves (Xu et al., 2013). Morphological parameters are based on raw data such as the number of leaves and lateral stems, SPAD value, fresh weight and dry weight of root and shoot, and canopy area, as revealed by Madhavi et al., 2021. The SPAD value is crucial for measuring the chlorophyll content of leaves, which largely determines photosynthetic capacity and plant growth (Li et al., 2018).

However, the investigation of ice plants' growth and bioactive compounds (compatible solutes, myo-inositol, pinitol) are affected by environmental conditions. To overcome the aforementioned difficulties, cultivation methods must be developed in controlled environments, such as plant factories. A plant factory is an automated system that artificially controls environmental conditions, light, relative humidity, temperature, and CO₂ concentration. It is possible to produce leafy vegetables similar to industrial products within a facility, regardless of the location (Takatsuji, 2008). According to Kim et al., 2018 various light sources have different effects on plant morphological changes. Considering this fact, in recent decades, researchers have conducted plant cultivation using light-emitting diodes (LEDs) because they have low high energy efficiency, heat emission, high power, and the capability of discrete emitting wavelengths with a narrow bandwidth. As a result, plant factories can produce high quality, higher annual yields with lower resource consumption and shorter production cycles for uniform plants without contamination (Lee et al., 2019).

Considerable research has been carried out with LEDs light treatment and sodium applications individually. Agarie et al., 2009 planted ice plants in underwater culture and added NaCl at 0, 100, 200, and 400 mM to the culture solution. This study found the maximum concentration of bioactive components, namely pinitol, ononitol, and myo-inositol, and high DPPH radical scavenging activity in NaCl 400 mM treatment, among other treatments. However, to our knowledge, a comparative evaluation of two types of salt and different concentrations, as well as the combined effect of salt on the development and growth of phytochemicals in ice plants, has not yet been carried out. Therefore, the current study attempted to investigate the effects of individual and combined concentrations of NaCl and CaCl₂ on the morphological parameters and the content of bioactive compounds in ice plants in a closed plant production system.

2. Materials and methods

2.1. Plant growth conditions

During the summer of 2021, the current investigation was conducted in the controlled plant factory at Gyeongsang National

University, Smart Farm Systems Laboratory, South Korea. The total duration of the experiment was 120 days (from early March to late June). The main environmental parameters, namely temperature, humidity, and CO₂ concentration, were monitored daily using a specific high precision sensor unit (Hanam Engineering Co. Ltd, South Korea) (Madhavi et al., 2022).

Seeds of the ice plant (*Mesembryanthemum crystallinum* L.) were sown in 50 cell plug trays [(54 cm × 28 cm × 4 cm (L × W × H)] using bio plus compost soil (cocopeat 68.86%, peat moss 11.00%, perlite 11.00%, and zeolite 9.00%) as a growing medium (Madhavi et al., 2022). Germinated seedlings were grown in a plant factory system for 28 days at 25 ± 1 °C, 60 ± 10% humidity, 1000 μmolmol⁻¹ CO₂, and under 120 μmolm⁻²s⁻¹ photosynthetic photon flux density (PPFD) using red, blue, and green LED treatment (Hanam Engineering Co. Ltd, South Korea) with 16 h/8h (light/dark) photoperiod as shown in Fig. 1.

Plants were watered three times a week with 50 ml of 0.5 × Hoagland's solution (Hoagland and Arnon, 1950). After appearing 4 leaves (28 days mature plant), 9 seedlings per treatment were transplanted in bio plus compost containing pots [(60 cm × 20 cm × 12 cm (L × W × H))] (Guan et al., 2020). For each treatment, 3 pots were used for 9 plants (20 cm apart; 3 plants per pot), with the single plant representing an independent biological replicate.

The different number of replicates were utilized for evaluating the effect of salt treatments on morphological growth, such as canopy area in the juvenile stage (n = 3), and the number of leaves, lateral stems, SPAD value, fresh weight, and dry weight of shoot and root in the adult stage (n = 9). Eventually, the cation and anion contents and DPPH radical scavenging activity(%) in the adult stage, three replicates were used.

Following the protocol by Cushman et al., 1990, plants in the control group were watered daily with 50 ml of 0.5 × Hoagland's solution (application into one pot; 3 plants), while those in the treatment group were also watered daily with 0.5 × Hoagland's solution containing the different salt treatments per pot as demonstrated in Table 1 (Guan et al., 2020). Previous studies reported that 400 mM NaCl exhibited the highest growth performance and bioactive component accumulation (Agarie et al., 2009). In this study, individual CaCl₂ concentrations were selected as 100 mM, 200 mM, and 400 mM as previous studies' NaCl individual concentrations (Agarie et al., 2007; Agarie et al., 2009). In addition to that, NaCl and a mixture of CaCl₂ and NaCl solution concentrations were maintained up to 400 mM in each salt treatment, and electrical conductivity (EC) was evaluated. EC was maintained at the same level for each solution, as denoted in Table 1. Initially, salt solutions

Table 1

Represent the eight types of salt treatments used in the experiment. mM: millimoles per litre, NaCl: sodium chloride, CaCl₂: calcium chloride, CS: T₁ to T₇: treatment groups 1, 2, 3, 4, 5, 6, and 7, respectively.

Treatments	Concentration levels of salt treatment with 0.5 × Hoagland's solution	EC value (dS/m) at 25 °C
CS	Control	4
T ₁	400 mM NaCl	39.6
T ₂	100 mM CaCl ₂	11
T ₃	200 mM CaCl ₂	22
T ₄	400 mM CaCl ₂	44
T ₅	100 mM NaCl + 300 mM CaCl ₂	42
T ₆	200 mM NaCl + 200 mM CaCl ₂	40
T ₇	300 mM NaCl + 100 mM CaCl ₂	38

were prepared and monitored EC (“Measuring soil salinity: plant stress, n.d.”). Simultaneously, the salt solution (25 ml) was mixed with 0.5 × Hoagland's solution (25 ml) and applied to the pot, as mentioned previously. EC values of final salt solutions with 0.5 × Hoagland's solution were indicated in Table 1. Eventually, the influence of salt stress on ice plant growth, and phytochemical changes were measured. The facts mentioned above affect the selection of eight salt treatments for our study.

2.1.1. Measurements of plant growth and morphology

Plant morphological parameters, namely number of leaves, number of lateral stems, and, SPAD value, were measured during 120 days of mature (adult phase) plants. Typically, the canopy area was high in the early stage, decreasing the later growth stage of all treatments. Therefore, canopy area measurement in the juvenile stage is crucial. Otherwise, all leaves drop and become small in the reproductive stage, as revealed by Bohnert and Cushman, 2000. The canopy area was measured during 60 days (juvenile phase) of mature plants by creating a python program with the green pixels against the reference reported by Guan et al., 2020.

The number of leaves was counted for all plant leaves except the cotyledons in each of the nine plants per treatment. The number of lateral stems was counted in every-nine plants. A portable chlorophyll meter (SPAD-502, Konica Minolta Inc., Osaka, Japan) was used to measure the value SPAD, indicating the chlorophyll content of the third leaf pair from the bottom leaf pair, in each of the nine plants per treatment (Kim et al., 2018). Subsequently, the canopy area was determined using the image processing algorithm developed by python program from Google Colaboratory using the ratio of green pixels against 4 cm² purple square reference paper with images taken by RGB camera (SONY DSC-RX100 vii, Korea) (Chaudhary et al., 2012). Three plant images were photographed in each control and salt treatment group, as demonstrated in Fig. 2. Each image of three plants was taken at 2 pm on Friday to track canopy growth and use the python program to measure canopy area (Fig. 2). Each image was captured using the RGB camera, cropped, and downscaled to a ratio of 0.25; the green, yellow and cyan colour threshold was extracted using the HSV colour model and converted to a black and white image shown by Madhavi et al., 2021. Precisely, plant and reference black and white image pixel values were calculated separately and evaluated the canopy area relevant to the reference area.

Eventually, the plants (120 days mature plant) were harvested three months after salt application. Fresh weights of shoots and roots were measured immediately after harvest using a digital balance (Model-FX-300iWP, A&D Company Limited, Tokyo, Japan). The dry weights were determined after drying the tissues in a drying oven (Shelves for 5E-DHG6310: 2 layers, Changsha Kaiyuan Instruments Co., Ltd, Changsha 410100, PR China) at 80 °C for



Fig. 1. Seedling stage (30 days after planting) of *M. crystallinum* grown under the plant factory system.

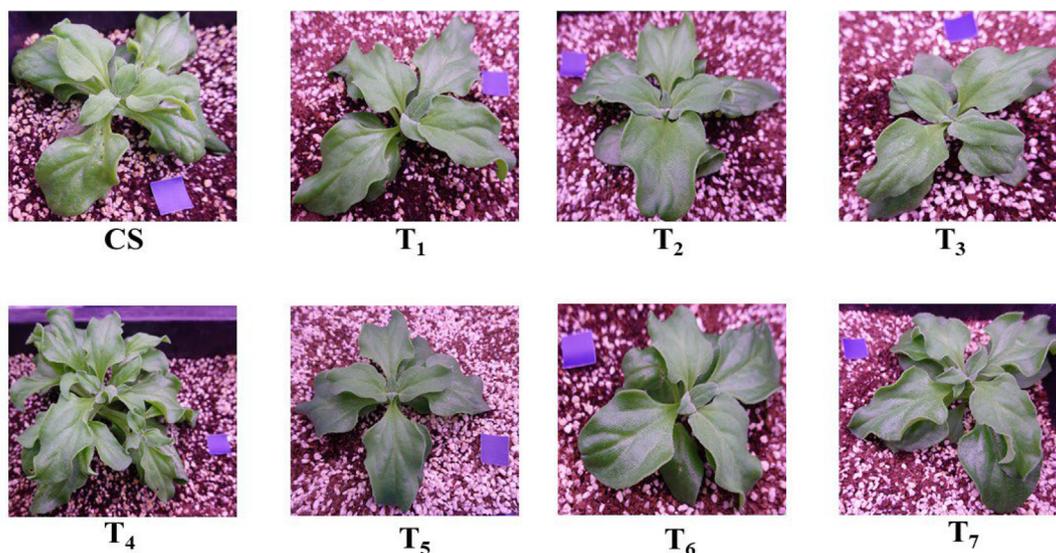


Fig. 2. Images acquisition for the canopy area measurement (60 days mature plant) of *M. crystallinum* grown under the plant factory system. The purple-coloured square paper demonstrated the reference for calculating the canopy area.

48 h (Agarie et al., 2009). Generally, all weight measurements were taken in every-nine plants per treatment.

2.2. Ion content determination

2.2.1. Cation content determination

The dried leaves of the ice plant (0.1 g) were kept in the centrifuge tubes (15 ml), and 6 ml of HNO_3 was added to each tube. The samples were filled into digestion tubes (50 °C to 220 °C) and kept for 1 h and 20 min (heating time- 30 min, holding time-20 min and cooling time-30 min) for microwave digestion in the advanced microwave digestion system (Milestone, IT/ETHOS EASY, Italy). Four millilitres of distilled water were added to the digested samples and filtered with the 0.45 μm membrane filter. One millilitre of the filtrate was taken and again diluted with distilled water up to 10 ml. Subsequently, the solution's cation (Na^+ , K^+ , Ca^{2+} , and Mg^{2+}) concentration was determined using an inductively coupled plasma spectrophotometer (Optima 8300, ICP-OES Optical System, and SCD Detector, Singapore) (Wilschefski and Baxter, 2019).

2.2.2. Anion content determination

One gram of dry leaves was ground with 20 ml of distilled water using a mortar and pestle and centrifuged at $3000\times g$ for 15 min (Eppendorf Microcentrifuge 5430 R, North America). The supernatant was filtered through Whatman 120 MM no 1 filter paper (Agarie et al., 2007). The filtrate was used for anion (Cl^- , NO_3^- , PO_4^{3-} and SO_4^{2-}) concentration determination by ion combustion chromatography (930 Compact IC Flex, Philippines) (Lopez-Ruiz, 2000).

2.3. DPPH radical scavenging activity(%)

Dried ice plant leaves (1 g) were ground with a mortar and pestle and placed in a centrifuge tube (15 ml) with 5 ml of 80% methanol. The mixture was shaken at room temperature for 12 h and centrifuged at $10,509\times g$ for 10 min. After centrifugation, the supernatant was filtered through Whatman 120 MM no 1 filter paper (Lee et al., 2015). The 2,2-diphenyl-1-picrylhydrazyl (DPPH, Sigma-Aldrich Co., MO, USA) determined the free radical scavenging activity of the ice plants modified method described by Kim et al 2021. The filtrates were used for analyzing the free radical

scavenging activity. The reaction mixture containing 1 ml of extract and 3 ml of 0.3 mM DPPH was shaken vigorously for 5 s and incubated in the dark at room temperature for 20 min. After incubation, the absorbance of the extract was measured at 517 nm using an Ultraviolet-VIS-NIR Spectrophotometer (U-4100, Hitachi, America), with methanol as the blank and methanol and DPPH solution as the control (Weeplian et al., 2018). The scavenging effect for DPPH radical was calculated using the following equation:

$$\text{Scavenging effect (\%)} = (1 - A_s/A_c) \times 100 \text{ (Weeplian et al., 2018).}$$

A_s and A_c absorbance of sample and absorbance of the control respectively, at 30 min.

2.4. Statistical analysis

Data on morphology, ion content, and DPPH radical scavenging of ice plants were collected in MS Excel (Microsoft Office 2019, Seattle, WA, USA). Moreover, for data analysis, statistical methods in the Statistical Package for the Social Sciences (SPSS) were used, including analysis of variance (one-way ANOVA) to practice with a significance level of $p < 0.05$. Significant differences between the means of the experimental data were tested using a Post-Hoc Tukey's HSD test in Statistics 10 (SPSS version: 25.0.0, IBM, New York, USA). All graphical plots were illustrated using OriginPro 9.0 (Origin Lab Corporation Northampton, MA, USA).

3. Results and discussion

3.1. Measurements of plant growth and morphology

Ice plants grown under different salt treatments showed appreciable differences in morphological parameters such as number of leaves, number of lateral stems, SPAD value, fresh and dry weight of shoots and roots, and canopy area. Subsequently, the number of leaves was significantly higher in treatment T_4 , whereas there was less leaf production with T_5 and T_6 treatments, as shown in Table 2. Consequently, higher leaf production was observed in CaCl_2 treatments. This is because calcium plays a direct role in the regulatory mechanisms that activate plants under salt stress and enhance the stimulation of growth regulators and other substances in plants (Xu et al., 2013). However, the CS, T_1 , T_2 , T_3 , and

T₇ showed no significant differences in total leaf numbers due to the different concentrations of NaCl and CaCl₂ salt treatments.

Correspondingly, the number of lateral stems was promoted by T₄ and T₇ treatments, whereas lateral stem production was suppressed by the T₆ treatment compared to other salt treatments (Table 2). Bohnert et al., 2000 found that this species is a salt accumulator, storing a high amount of NaCl and CaCl₂ in a gradient from roots to developing shoots. Furthermore, at adequate salinity levels ranging from 100 to 400 mM, this salt-loving plant grew at its best (Agarie et al., 2007). Therefore, lateral stem growth was stimulated by the high concentration of salt treatment.

Interestingly, fresh weight and dry weight of shoots and roots tended to be higher in treatments T₄ and T₇, as shown in Table 2. Furthermore, a positive correlation was observed with the number of leaves, lateral stems, and canopy area related to the ice plant's shoot and root biomass. Kim et al., 2021 also exhibited that the number of leaves, lateral stems, and leaf area of ice plants positively correlated with the root and shoot biomass. Thus, it concluded that the highest morphological growth is appreciably related to the biomass of ice plants. Agarie et al., 2007 reported that 400 mM NaCl increased the dry weight of wild-type ice plants by almost 2-fold compared to the mutant ice plants. Therefore, the higher concentration of salt impacts the biomass enhancement of ice plants.

Different salt concentrations affected the SPAD value and photosynthetic rate inside the ice plant leaves (Table 2). Moreover, the SPAD value was significantly higher in T₆ than within the other different treatments. The photosynthetic rate was not appreciably different within the salt treatments CS, T₁, T₄, T₅, and T₇. A low photosynthetic rate was observed in salt treatments T₃ and T₂. However, the SPAD value and photosynthetic rate of ice plant leaves showed a considerably high value in T₆ treatment because the similar concentration of NaCl and CaCl₂ increased chlorophylls and net photosynthetic rate (Trajkova et al., 2006). The higher concentration of NaCl and CaCl₂ decreased the contents of chlorophylls and the net photosynthetic rate. It accelerates the respiration rate and CO₂ compensation concentration in leaves, as revealed by Khavari-Nejad and Chaparzadeh (1998).

Table 2

The effect of salt treatments on the number of leaves, number of lateral stems, SPAD value, fresh weight, and dry weight of ice plants grown under different salt treatments in the plant factory system.

Salt treatments	Number of leaves	Number of lateral stems	SPAD value	Fresh weight (g/plant)		Dry weight (g/plant)	
				Shoot	Root	Shoot	Root
CS	44.67 ±2.45 ^b	9.78 ±1.64 ^{ab}	50.18 ±4.54 ^b	21.29 ±1.08 ^d	2.60 ± 0.35 ^c	0.68 ±0.01 ^d	0.56 ±0.03 ^d
T ₁	46.78 ±2.54 ^b	6.89 ±1.90 ^b	42.54 ±3.86 ^{bc}	15.67 ±0.28 ^f	1.29 ±0.15 ^e	0.54 ±0.01 ^f	0.37 ±0.03 ^f
T ₂	47.56 ±4.69 ^b	10.56 ±1.13 ^{ab}	39.92 ± 2.76 ^c	24.76 ±0.42 ^c	3.56 ±0.34 ^b	0.83 ±0.02 ^c	0.65 ±0.03 ^c
T ₃	46.33 ±4.85 ^b	7.22 ±1.99 ^b	39.90 ± 2.66 ^c	17.49 ±0.60 ^e	1.69 ±0.16 ^d	0.60 ±0.03 ^e	0.45 ±0.03 ^e
T ₄	64.89 ±5.40 ^a	13.33 ±2.24 ^a	48.02 ± 2.20 ^b	30.84 ±0.74 ^a	4.17 ±0.08 ^a	0.93 ±0.02 ^a	0.86 ±0.03 ^a
T ₅	25.44 ±2.24 ^c	8.11 ± 1.45 ^b	49.73 ±4.71 ^b	20.14 ±0.75 ^d	0.80 ±0.06 ^f	0.52 ±0.01 ^f	0.07 ±0.01 ^g
T ₆	29.44 ±2.24 ^c	5.22 ±1.99 ^b	65.08 ± 2.03 ^a	12.34 ±0.89 ^g	0.60 ±0.06 ^g	0.47 ±0.01 ^g	0.04 ±0.01 ^h
T ₇	51.11 ±4.59 ^b	11.11 ±1.54 ^a	47.89 ±1.81 ^b	28.59 ±1.08 ^b	3.76 ±0.15 ^b	0.87 ±0.01 ^b	0.74 ±0.03 ^b

Different letters indicate significant differences between groups (p < 0.05). Values represented means ± SD (n = 9).

The T₄ (258.08 ± 2.09 cm²) and T₇ (207.74 ± 1.21 cm²) salt treatments produced a larger canopy area during the early growth stage, while the T₃ treatment produced a lower canopy area (60.26 ± 2.57 cm²). Besides, salt application induces cell expansion and protein synthesis in ice plants, facilitating leaf expansion and a high growth rate (Yang and Yen, 2002). The response of canopy area for the other salt treatments in the juvenile stage was observed in T₂ (183.01 ± 2.47 cm²) followed by T₁ (174.78 ± 3.41 cm²), T₅ (133.11 ± 2.52 cm²), T₆ (103.68 ± 2.06 cm²), and CS (66.36 ± 1.42 cm²) as in Fig. 4. There were no appreciable differences between CS and T₃ during the early stage of the slow uptake of salt to plant (Figs. 3, 4).

3.2. Ion content determination

3.2.1. Cation content determination

The results of cation content clearly showed that the ice plant accumulated comparatively much Na⁺ in T₄ (210.72 ± 9.24 mg g⁻¹) and less in T₁ (89.58 ± 1.33 mg g⁻¹) was expressed on a dry weight basis (Fig. 5A). There were no significant differences for Na⁺ storage in ice plant leaves among T₅ and T₇ treatments and between T₁ and T₂ salt treatments. The Na⁺ storage in ice plant leaves response to other salt treatments were T₄ (210.72 ± 9.24 mg g⁻¹), T₆ (204.08 ± 6.80 mg g⁻¹), T₇ (187.34 ± 8.97 mg g⁻¹), T₅ (181.06 ± 9.43 mg g⁻¹), T₃ (156.91 ± 4.12 mg g⁻¹), CS (104.40 ± 2.89 mg g⁻¹) and T₂ (93.69 ± 3.02 mg g⁻¹) in respectively. Furthermore, T₄ and T₆ exhibited the highest Na⁺ content due to the salt transport from root cells into leaf mesophyll relying on myo-inositol, and sodium interaction between myo-inositol facilitates the sodium uptake and long-distance transport. Moreover, the synergy between Ca²⁺ ion accumulation leads to the synthesis of myo-inositol and the uptake of Na⁺ (Nelson et al., 1999). As a result, sodium is stored in EBCs and plant younger tissues, and pinitol production in the ice plant shoot top is enhanced by salt. A similar tendency was observed in the Ca²⁺ ion results, as demonstrated in Fig. 5C.

According to the K⁺ ion results, the highest value was measured in T₃ (118.23 ± 5.26 mg g⁻¹), and the lowest value CS (33.04 ± 2.46 mg g⁻¹) represented as a percentage of dry weight basis,

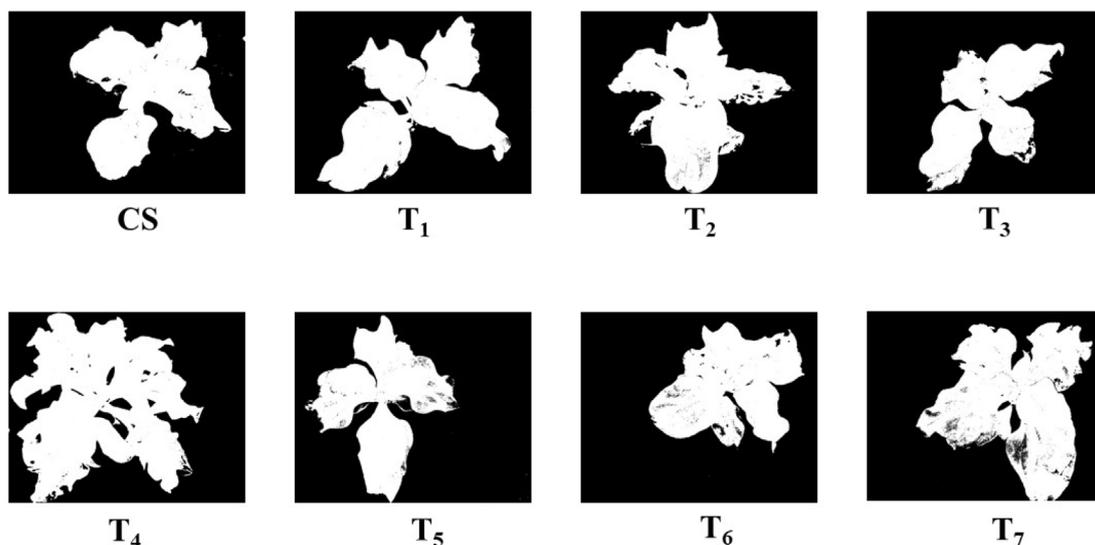


Fig. 3. HSV developed images for calculating the canopy area of ice plants in their juvenile growth stage.

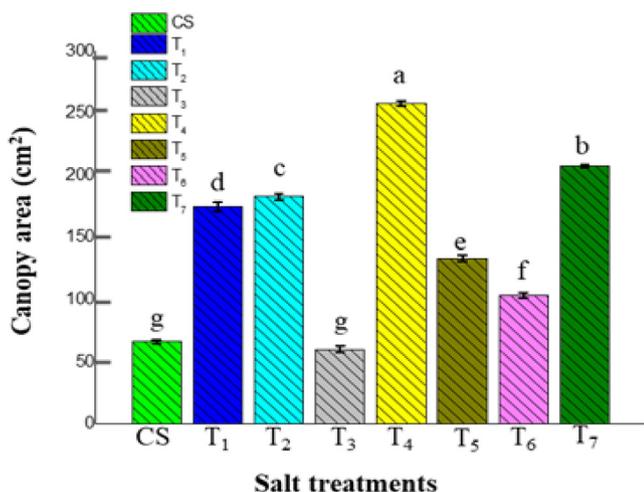


Fig. 4. The impact of various salt treatments on the canopy area of juvenile stage ice plants (60-days mature plants). The data indicate the means \pm SD (n = 3). Different letters above bars indicate significant differences at $p < 0.05$.

respectively. There was no appreciable difference noted among K⁺ ion accretion among T₄ and T₅ treatments, T₆ and T₇ treatments, and T₁ and T₂.

Based on the K⁺ ion results the remaining treatment showed the storage such as T₅ (103.37 \pm 2.75 mg g⁻¹), T₄ (98.36 \pm 5.13 mg g⁻¹), T₇ (95.62 \pm 4.35 mg g⁻¹), T₆ (93.20 \pm 5.36 mg g⁻¹), T₂ (46.50 \pm 2.52 mg g⁻¹) and T₁ (43.40 \pm 1.82 mg g⁻¹) respectively. These results have revealed that a lower concentration of CaCl₂ influences the accumulation of K⁺ ions in tissues of leaves and EBCs.

Moreover, increasing NaCl up to 400 mM reduced the K⁺ ion accumulation of photosynthetically active leaf tissues of ice plants, as illustrated in Fig. 5B. However, considering the Ca²⁺ ion, the significantly high concentration existed in T₄ in dry weight basis (20.90 \pm 0.90 mg g⁻¹) and less in T₂ (5.19 \pm 0.61 mg g⁻¹) respectively (Fig. 5C). No significant difference was observed in the Ca²⁺ ion accumulation among T₃ and T₅ treatments, and CS, T₁, and T₂ treatments. According to the results, Ca²⁺ ions in other salt treatments were observed T₆ (17.08 \pm 1.57 mg g⁻¹), T₅ (14.21 \pm 1.06 mg g⁻¹), T₃ (13.44 \pm 1.04 mg g⁻¹), T₇ (11.01 \pm 0.30 mg g⁻¹), T₁ (5.72 \pm 0.79 mg g⁻¹), CS (5.62 \pm 0.98 mg g⁻¹) dry weight basis in consecutively. Thus, results showed an increasing trend in salt

treatments and accumulation of Ca²⁺ ions in leaf tissues in ice plants.

Regarding the Mg²⁺ ion, the highest accumulation was observed in T₃ as a dry weight basis (11.62 \pm 1.06 mg g⁻¹) and the lowest in CS (3.90 \pm 0.53 mg g⁻¹) orderly (Fig. 5D). Based on the results no significant difference was observed among T₃, T₅ (11.54 \pm 0.33 mg g⁻¹) and T₆ (10.89 \pm 0.17 mg g⁻¹) treatments and between T₄ (9.37 \pm 0.93 mg g⁻¹) and T₇ (9.09 \pm 0.78 mg g⁻¹) treatments and among T₁ (4.41 \pm 0.60 mg g⁻¹), and T₂ (4.13 \pm 0.48 mg g⁻¹) treatments in respectively. According to the results mentioned above, Na⁺ and Ca²⁺ ions showed a positive tendency for accumulation. In contrast, similar trends were observed K⁺ and Mg²⁺ ions deposition in ice plant leaves. However, Na⁺, K⁺ and Mg²⁺ absorption were high in the presence of CaCl₂ salt treatment due to the NaCl suppressing the K⁺ ion concentration and CaCl₂ enhancing the leaf Cl⁻ and Ca²⁺ level, which leads to the uptake of other cations (Trajkova et al., 2006). A high concentration of NaCl worked on the lack of absorption of K⁺, Mg²⁺, Ca²⁺, also Na⁺ competes with K⁺ and replaces it in the cell, as reported by Mohamed and Basalah, 2015.

3.2.2. Anion content determination

To assess the possible content of anion sequestration on a dry weight basis, the Cl⁻ stored high in T₁ (57.35 \pm 0.63 mg g⁻¹) and less in CS (26.06 \pm 1.96 mg g⁻¹) consecutively. According to the anion storage in ice plant leaves, no significant differences were observed among T₁ and T₅ (57.28 \pm 3.58 mg g⁻¹) treatments and T₄ (35.74 \pm 2.73 mg g⁻¹), T₆ (38.46 \pm 1.63 mg g⁻¹) and T₇ (37.23 \pm 2.00 mg g⁻¹) treatments. Following the Cl⁻ sequestration appreciable difference was observed T₃ (47.45 \pm 2.18 mg g⁻¹) and T₂ (31.45 \pm 1.24 mg g⁻¹) treatments in orderly. According to the individual (NaCl) and combination effect (NaCl and CaCl₂) results, the highest concentration, 400 mM, exhibited the highest leaf Cl⁻ accumulation compared to the lower concentration of salt treatment (T₂) and control samples, as shown in Fig. 6A.

Correspondingly, NO₃⁻ and SO₄²⁻ are vital for amino acid synthesis and metabolic reactions of halophytic plants. According to the NO₃⁻ ion results from the highest value obtained in T₆ as dry weight basis (165.88 \pm 1.63 mg g⁻¹) and lowest results followed by T₁ (90.58 \pm 1.77 mg g⁻¹), T₇ (83.82 \pm 2.65 mg g⁻¹) and T₅ (75.08 \pm 4.99 mg g⁻¹) respectively (Fig. 6 B). However, no significant difference was shown among CS (120.12 \pm 4.23 mg g⁻¹), T₂ (124.58 \pm 2.90 mg g⁻¹) and T₄ (124.76 \pm 3.11 mg g⁻¹)

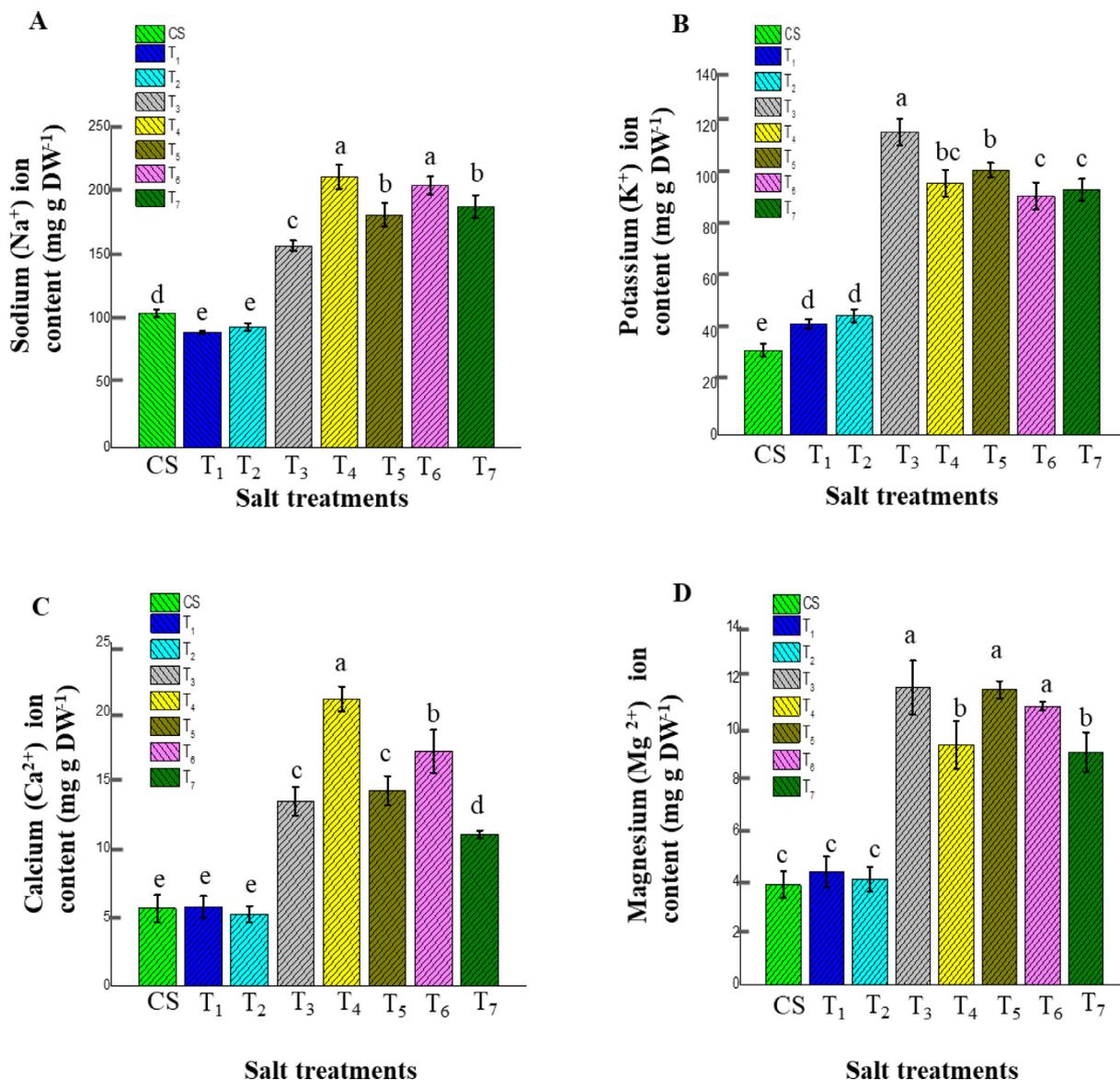


Fig. 5. The effect of different salt treatments on cation content of ice plant leaves in the adult stage (120 days mature plant). Data indicate the means \pm SD (n = 3). Different letters above bars indicate significant differences at $p < 0.05$. (A) Na⁺, (B) K⁺, (C) Ca²⁺, (D) Mg²⁺ indicated the ion content measurement based on dry weight, respectively.

treatments, and T₃ (75.69 \pm 2.01 mg g⁻¹) and T₅ (75.08 \pm 4.99 mg g⁻¹) treatments in respectively.

Concurrently, a similar pattern was discovered in SO₄²⁻ ion accumulation. The maximum value for SO₄²⁻ storage in leaves tissue was observed in T₃ (24.08 \pm 2.73 mg g⁻¹) and the minimum in T₅ (9.23 \pm 0.38 mg g⁻¹) on a dry weight basis orderly (Fig. 6 D). Based on the results, there were no significant differences among CS (17.89 \pm 1.94 mg g⁻¹), T₁ (17.73 \pm 1.84 mg g⁻¹), T₂ (18.99 \pm 2.15 mg g⁻¹), T₇ (18.34 \pm 0.97 mg g⁻¹) and T₄ (15.43 \pm 2.20 mg g⁻¹) treatments in respectively.

Changes in NO₃⁻ and SO₄²⁻ concentrations in the leaves of the ice plant confirmed a comparable absorption trend (Fig. 6 B, D). Increased Cl⁻ concentrations competed with NO₃⁻ uptake, and accumulation of SO₄²⁻. Generally, NO₃⁻ and SO₄²⁻ showed the same ion exchange and sorption pattern described by Stefan et al., 2014. Furthermore and even more importantly, PO₄³⁻ is vital to plant growth and involved in several plant functions such as photosynthesis, the transformation of sugars, energy transfer, and nutrient movement, as reported by Mullins (2009). Therefore, elucidating the relationship between the energy state and stress tolerance

PO₄³⁻ ion is vital for growth hormone presumably gibberellin production and regulates the abiotic stress in halophytes (Zhu et al., 2012).

Based on the PO₄³⁻ ion deposition results, the appreciable differences depicted T₃ as dry weight basis (80.41 \pm 1.57 mg g⁻¹) followed by T₅ (52.84 \pm 1.47 mg g⁻¹), T₇ (46.92 \pm 3.06 mg g⁻¹), and T₂ (28.37 \pm 3.73 mg g⁻¹) respectively (Fig. 6 C). Nonetheless, there was no significant difference obtained for the PO₄³⁻ ion deposition among in T₄ (41.58 \pm 3.52 mg g⁻¹), CS (39.24 \pm 2.83 mg g⁻¹), T₁ (36.17 \pm 3.02 mg g⁻¹), T₆ (33.10 \pm 2.40 mg g⁻¹) in consecutively.

As aforementioned, the results have revealed that a lower concentration of CaCl₂ significantly influences the PO₄³⁻ deposit in ice plant leaves.

3.3. DPPH radical scavenging activity(%)

The DPPH assay was used to assess the scavenging activity of ice plants under various salt stresses, as presented in Fig. 7. The lower value indicates the higher antioxidant activity, and the value of DPPH radical scavenging activity denoted the negative correlation

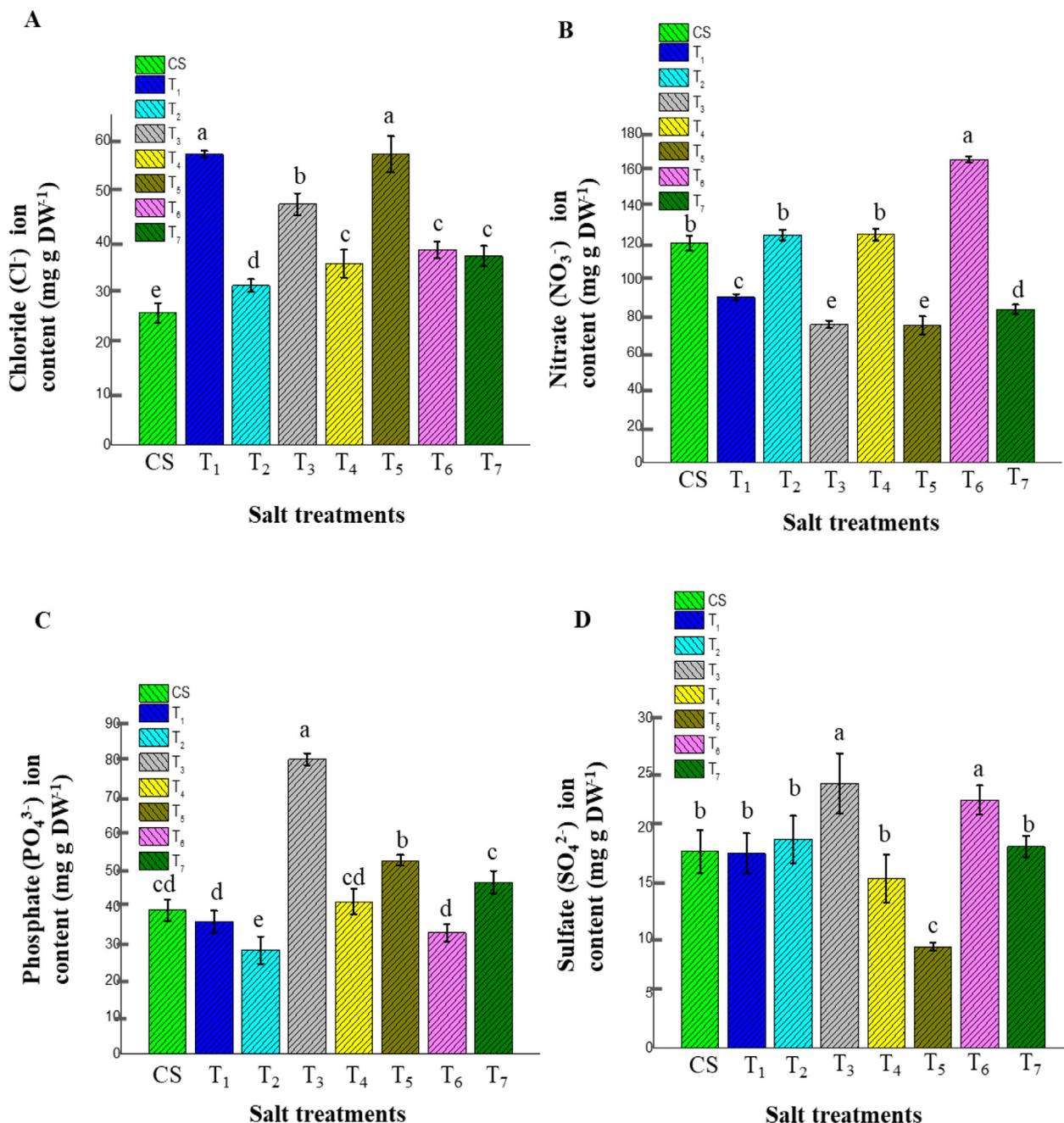


Fig. 6. The effect of different salt treatments on the anion content of adult stage ice plant leaves (120 days old plant). The data indicate the means \pm SD (n = 3). Different letters above bars indicate significant differences at $p < 0.05$. (A) Cl^- , (B) NO_3^- , (C) PO_4^{3-} , (D) SO_4^{2-} denoted the ion content measurement based on dry weight, respectively.

with the antioxidant activity (Agarie et al., 2009). However, a scavenging activity among the salt treatments, no appreciable difference was denoted among T₅ (89.38 \pm 1.21%), T₇ (87.96 \pm 2.33%), CS (87.05 \pm 5.23%) and T₄ (86.35 \pm 4.88%) treatments and T₃ (45.98 \pm 3.96%) and T₂ (44.93 \pm 3.70%) treatments. There was significant radical scavenging activity was observed among T₆ (77.90 \pm 2.10%), T₁ (52.59 \pm 3.11%), T₂ treatments respectively.

According to the results, higher amounts of phenolic compounds are markedly promoted in the T₂ salt treatment, which has a minimum radical scavenging effect.

Salt stress was successful in promoting pinitol/ononitol accumulation in *M. crystallinum* halophytic characteristics. It was speculated that the combined salt created a turgor gradient along the axes of the mature ice plant, which resulted in the accelerated

growth of new cells. Photosynthesis and photorespiration were associated with the concentration gradient from the root to the mesophyll cell. The methylation of myo-inositol to the intermediate ononitol produces pinitol, which appears to be epimerized to pinitol. These phenolic chemicals are primarily responsible for the colour, flavour, and aroma of plants, and they are also high in antioxidant activity (Hichri et al., 2011). A similar observation was obtained from the present study from T₁ treatment as low radical scavenging activity and high antioxidant activity by decomposition of phenolic compounds.

Antioxidant activity plays a fascinating role in removing reactive oxygen species that are harmful to the human body. According to Kim et al., 2021, it helps to avoid chronic diseases and ageing. As a result, several efforts to synthesize particular phytochemicals

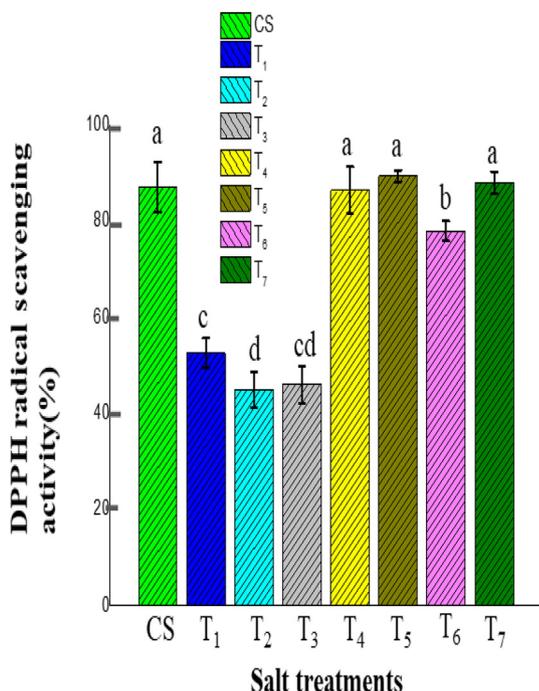


Fig. 7. The effect of different salt treatments on DPPH radical scavenging activity(%) of ice plants in the adult stage (120 days old plant). Data indicate the means \pm SD ($n = 3$). Different letters above bars indicate significant differences at $p < 0.05$.

with antioxidant activity have been conducted. The bioactive concentrations in ice plants in T₁, T₂, and T₃ treatments were high due to the mild concentration of CaCl₂, and the high NaCl concentration affects the biosynthetic pathway through the accumulation of phenolic compounds. Based on the results, the combined impact of NaCl and CaCl₂ and a higher concentration of CaCl₂ were not affected by the increased antioxidant material as fewer secondary metabolites accumulated (Agarie et al., 2009).

Specifically, Agarie et al., 2009 revealed that the radical scavenging activity determined by the DPPH assay and the antioxidant activity of ice plants increased twofold by treatment with 400 mM NaCl, which was twice that of the lettuce plants. This study obtained a similar tendency of high antioxidant activity from the high concentration of NaCl and low CaCl₂ salt treatments such as T₁, T₂, and T₃. These results indicate the high potential of ice plants as polyol-rich, highly functional foods.

4. Conclusion

The present study revealed that T₄ (400 mM CaCl₂) salt treatment considerably improved ice plant morphological parameters such as number of leaves, number of lateral stems, fresh and dry weight of shoot and root, and leaf canopy area. It has a high potential for enhancing the biomass of the ice plant's accumulation of functional materials. Consequentially, T₄ salt treatment exhibited a similar trend for the Na⁺ and Ca²⁺ deposition in ice plant leaves. Moreover, K⁺ and Mg²⁺ accumulation were high in T₃ (200 mM CaCl₂) salt treatment, with a mild concentration of CaCl₂. A negative trend was denoted by cation accumulation in ice plant leaves regarding anion deposition.

Furthermore, NaCl, including salt treatments such as T₁ (400 mM NaCl) and T₅ (combination of 100 mM NaCl and 300 mM CaCl₂), significantly influenced the Cl⁻ deposition, and T₆ (mixture of 200 mM NaCl and 200 mM CaCl₂) affected NO₃⁻ accumulation. Conversely, a mild concentration of CaCl₂ as T₃ salt treatment positively affected the PO₄³⁻ and SO₄²⁻ high accumulation in

the ice plant leaves. Simultaneously, DPPH radical scavenging activity(%) results negatively correlated with the antioxidant activity, and a high antioxidant value was observed in T₂ (100 mM CaCl₂), which has a higher potential of inducing polyol synthesis. In conclusion, T₄ implied better performance of ice plants in terms of biomass and induced cation (Na⁺, Ca²⁺) accumulation. In addition, further studies seem to be required to clarify the effect of different types of salt treatments on the growth and secondary metabolism of ice plants.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Acknowledgement

This work was supported by Korea Institute of Planning and Evaluation for Technology in Food, Agriculture, Forestry and Fisheries (IPET) through Agriculture, Food and Rural Affairs Convergence Technologies Program for Educating Creative Global Leader, funded by Ministry of Agriculture, Food and Rural Affairs (MAFRA) (717001-7).

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